# Electron Paramagnetic Resonance (EPR) SDSL – Site Directed Spin Labeling

**1. Discovery of EPR** – E. Zavoisky, USSR, 1945

http://kfti.knc.ru/eng/about/index.html

#### 2. Development as a Biophysical Tool. - Synthesis of a stable

free radical (the nitroxide), and labeled biomolecules

(Spin labels) (Humphries & McConnell *in* Methods of Experimental Physics, vol. 20, G. Ehrenstein and H. Lecar, eds., 1982, Academic Press, p. 53-122.)

#### **3.** Applications – dynamics of molecular motions in macromolecules

and supramolecular assemblies. Examples taken from Hubbell, Cafiso and Altenbach. 2000. Identifying conformational changes with site-directed spin labeling. *Nat. Struct. Biol.* **7**:735-739. & Columbus and Hubbell. 2002. <u>A new spin on protein dynamics</u>. *Trends in Biochemical Sciences.* 27:288-295. (Review)

# 2. Basic Properties of ESR

- **1. Electrons have 'spin'**. A circulating electric charge has a magnetic moment.  $\mu = -q\beta S$
- 2. Energy Levels are Nondegenerate in a Magnetic Field. Electrons (and some nuclei) with spin = ½, have two energy levels, which results in one transition.

 $E = -\mu \cdot \mathbf{H}_0 = g\beta \mathbf{S} \cdot \mathbf{H}_0$ 

In an external magnetic field applied along the z-axis, the energies can be on the z-component of the spin,  $+/- \frac{1}{2}$ 

**3. The Transition Energy** is given as the difference in the low and high energy levels:

$$h_{\rm V} = E(S_{\rm z} = +1/2) - E(S_{\rm z} = -1/2)$$

 $h_{\rm V} = g\beta H_0$ 



### 2. Basic Properties of ESR

The Transition Energy is also expressed as:

$$h_{\mathcal{V}} = g\beta H_0 \qquad \rightarrow \qquad 2\pi \nu = g\beta H_0 / \hbar$$
$$\omega = \gamma H_0$$

 $\gamma$  is the magnetogyric (or gyromagnetic) ratio



# 2. Nitroxide Spin Labels



Nitroxide Spin Labels are Stable between pH 3 and 10

They must be tetra-substituted for stability



They are sensitive (quenched by) reductants

![](_page_3_Figure_6.jpeg)

Labels have been incorporated into biomolecules

![](_page_4_Figure_1.jpeg)

![](_page_4_Figure_2.jpeg)

**Proteins** 

![](_page_5_Figure_0.jpeg)

The Local Field Produced by the Magnetic Moment of <sup>14</sup>N Results in Hyperfine Splitting

(Humphries & McConnell, 1982)

#### Hyperfine Splitting Depends on Orientation of Molecule with Respect to the Magnetic Field

![](_page_6_Figure_1.jpeg)

![](_page_6_Figure_2.jpeg)

#### ← 'Powder Pattern' of an Isotropic (Unoriented) Immobilized Label

#### Hyperfine Splitting in Oriented Spin Label Crystals (Spin Label doped in Cholesteryl Chloride)

![](_page_7_Figure_1.jpeg)

![](_page_7_Figure_2.jpeg)

# Effects of Mobility

![](_page_8_Figure_1.jpeg)

![](_page_8_Figure_2.jpeg)

Immobilized

# **Magnetic Field Dependence of EPR Line Shapes**

#### Increase Magnetic Field

- → Increase energy level separation
- → Shorten lifetime of excited state by a change in the spontaneous decay rate
- → Shift the observation window of molecular dynamics from longer to shorter time scales

![](_page_9_Figure_5.jpeg)

P. P. Borbat et al., Science 291, 266 - 269 (2001)

![](_page_9_Picture_7.jpeg)

# **Dipole-Dipole Exchange Interactions**

![](_page_10_Figure_1.jpeg)

### Exchange Interactions in Double-Labeled T4 Lysozyme Samples Reflect Domain Motion

![](_page_11_Figure_1.jpeg)

#### Measurement of, and changes in, Restricted Mobility

Scaled mobility:  $M_s = \frac{\left(\delta^{-1} - \delta_i^{-1}\right)}{\left(\delta_m^{-1} - \delta_i^{-1}\right)}$  where  $\delta$ ,  $\delta_m$ ,  $\delta_i$  are the measured peak width, the most mobile label peak width and the least mobile label width, respectively. In these examples  $\delta_m = 2.1$  Gauss and  $\delta_i = 8.4$  Gauss.

![](_page_12_Figure_2.jpeg)

(Columbus & Hubbell, 2002)

-1.2

1.0

0.8

0.6 🚬

T/BS

0.4

50