

Electromagnetic Radiation

Go to overhead →
(see also p. 531)

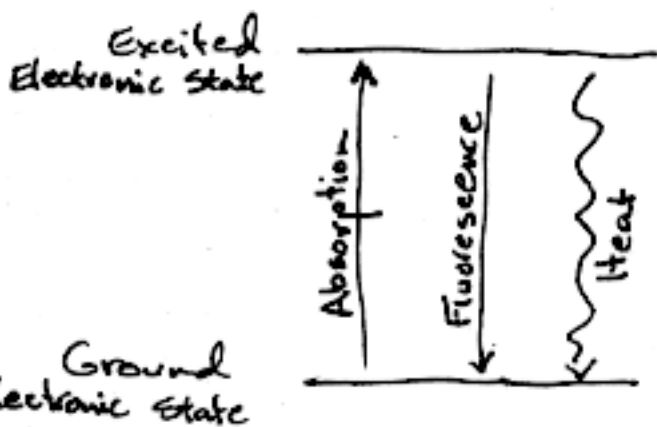
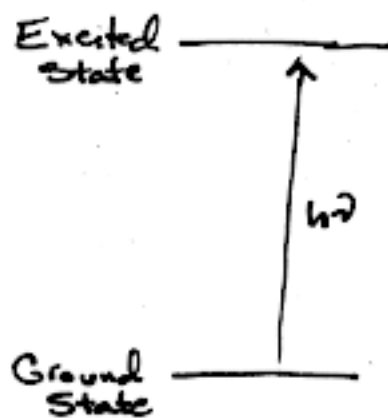
$$\text{Frequency} = \nu = \frac{c}{\lambda} \leftarrow \text{wavelength}$$

So a plot of λ is a plot of $1/\nu$

X-Rays	10^{-12} m		Very High E
UV light	10^{-8} m (100nm)		High E
Red light		(700nm)	Lower E
Microwaves	10^{-2} m		Very Low E

Absorption

Light (any E.M.)
can induce transitions
between states of
different energies



$$I(\nu) = \frac{c}{n} \rho(\nu)$$

Intensity

refractive index

speed of light

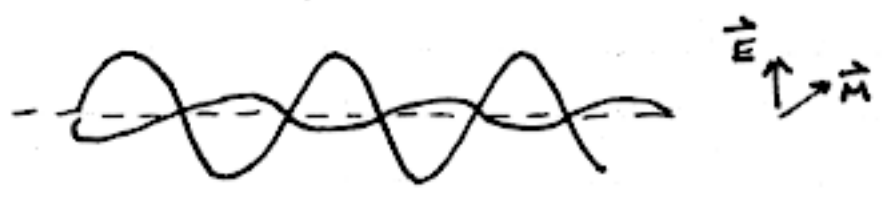
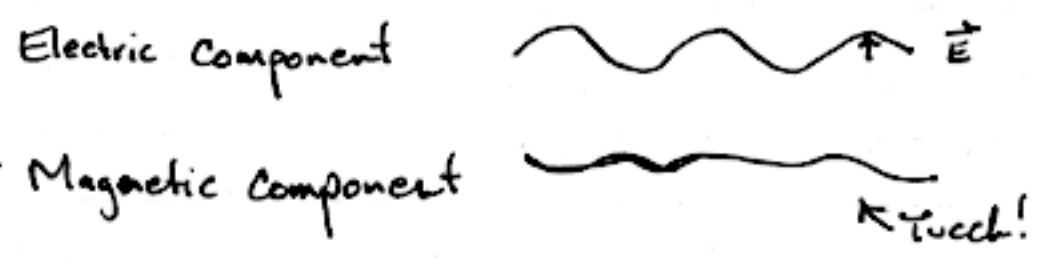
Energy/volume (Energy density)

Radiation - Induced \Rightarrow How?

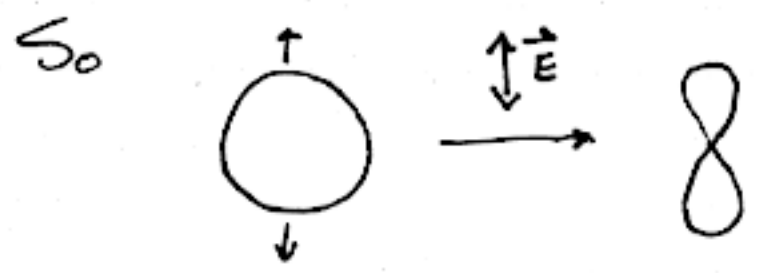
Absorption
and
Stimulated
Emission

ΔE must match $h\nu$
but that's not enough.

Light/E.M. is an electro/magnetic field



Electrons (charged) interact w/ the ~~mag~~ electric field, changing their ~~speci~~ spatial distribution.



Not easy to visualize completely, but...

We say that a radiation field
"Connects" the two states

Two different wave functions (orbitals)

Transition Dipole Moment

$$\mu_{0A} = \int \psi_0 \mu \psi_A dv$$

Electric vector of EM

Tells us the probability that the electric field (E, $\vec{\mu}$) can induce a transition from ψ_A to ψ_0

How? Qualitatively, at least.

If $\vec{\mu}(E)$ "operates on" (distorts) the electron distribution ψ_A so that it now looks like ψ_0 , then

Actually $D_{0A} = |\mu_{0A}|^2$

$$\mu_{0A} = \int \psi_0 (\psi_0) dv = 1$$

⇒ Maximal probability.

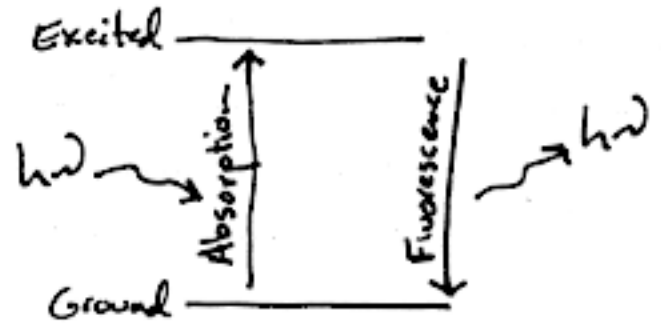
If a bit less perfect, then $\mu_{0A} < 1$ but it can still be non-zero.

Note that E can also act on ψ_0 to make it look like ψ_A

$$\text{In general, } \int \psi_0 \mu \psi_A dv = \int \psi_A \mu \psi_0 dv$$

⇒ Light stimulates UP and Down.

Light stimulates both absorption and emission.



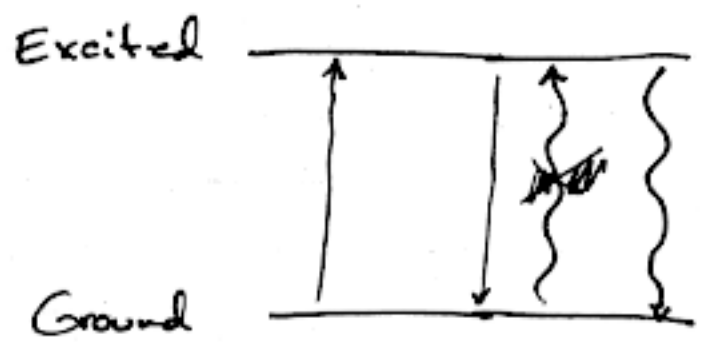
(Can also have spontaneous emission → more complex)

Result: If ground and excited states have equal populations, then

transitions up = # transitions down

→ No net absorption

Common in NMR, where ΔE is small rel. to kT
Not common in electronic spec, ΔE large " " "
Boltzmann important here.



We talk of lifetime of the excited state.
How quickly does it convert back
via ↓ and {