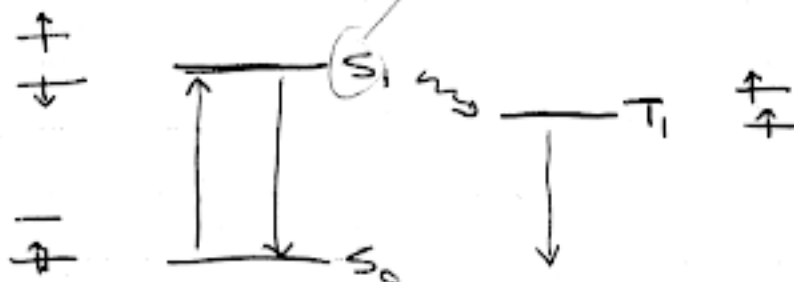


12/7/01

Q: Why the "S"?

(1/8)

Phosphorescence (glow in the dark paint)



When an e^- goes from ψ_1 to ψ_2
it does not simultaneously change its spin

Paired spins \Rightarrow singlet (s) state

But Pauli says that $\begin{array}{c} \uparrow \\ \uparrow \end{array}$ is lower E than $\begin{array}{c} \uparrow \\ \downarrow \end{array}$

So with some probability a spin can flip.

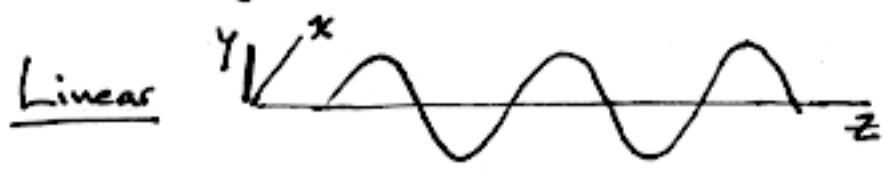
Unpaired spins \Rightarrow triplet (t) state.

But then $T_1 \xrightarrow{h\nu} S_0$ is "forbidden" (unlikely)

because the spin flips simultaneously with the orbital transition.

Since it's "forbidden," it occurs very slowly
(hours sometimes,
compared with $\mu s \rightarrow ns$ for fluorescence).

Polarized light



in the picture above
 E_y oscillates
 $E_x = E_z = 0$

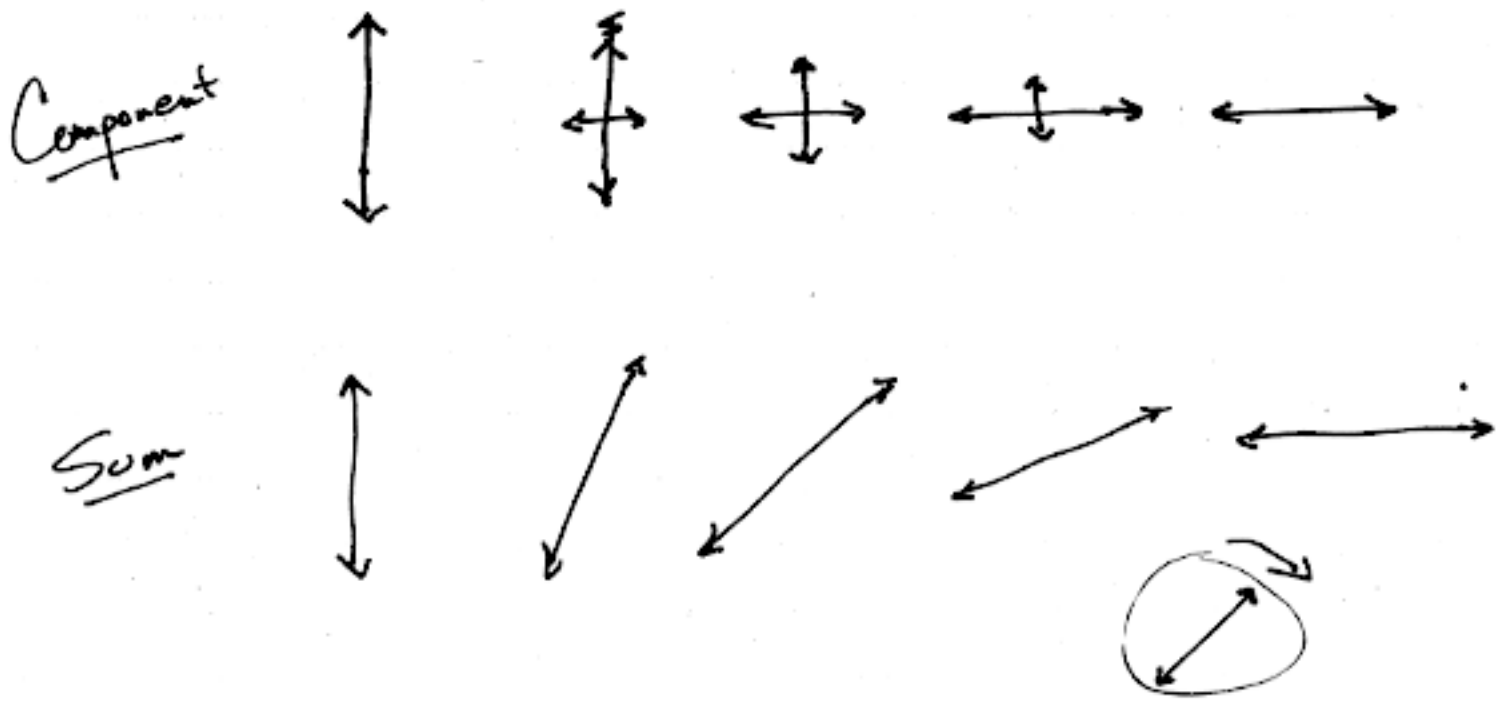
light propagating in z-direction
 polarized in y-direction

We saw before that
 this particular polarization
 can only excite ~~light~~
 some transitions.

CIRCULAR

IMAGINE ADDING ^{two} \vec{E} vectors ~~prop~~

Perpendicular and exactly out of phase



Result is polarized light rotating as it propogates.

— See pretty pictures in text

Mathematically

$$\vec{E}(z,t) = \frac{1}{2}E_0 \sin\left[2\pi\left(\frac{z}{\lambda} - \frac{ct}{\lambda}\right)\right] \hat{i} + \frac{1}{2}E_0 \sin\left[2\pi\left(\frac{z}{\lambda} + \frac{ct}{\lambda} + \frac{1}{4}\right)\right] \hat{j}$$

OR

$$\underbrace{\frac{1}{2}E_0 \sin\left[2\pi\left(\frac{z}{\lambda} - \frac{ct}{\lambda}\right)\right] \hat{i}}_{x\text{-component}} + \underbrace{\frac{1}{2}E_0 \cos\left[2\pi\left(\frac{z}{\lambda} + \frac{ct}{\lambda}\right)\right] \hat{j}}_{y\text{-component}}$$

⇒ Harder to imagine, but molecules with a particular chirality, absorb left circ polarized light a bit more than right circularly polarized light.

SCATTERING (BIREFRINGENCE)

$$\text{Rotation} = \phi = \frac{\pi}{\lambda} (n_L - n_R)$$

We have not talked about this. Don't worry.

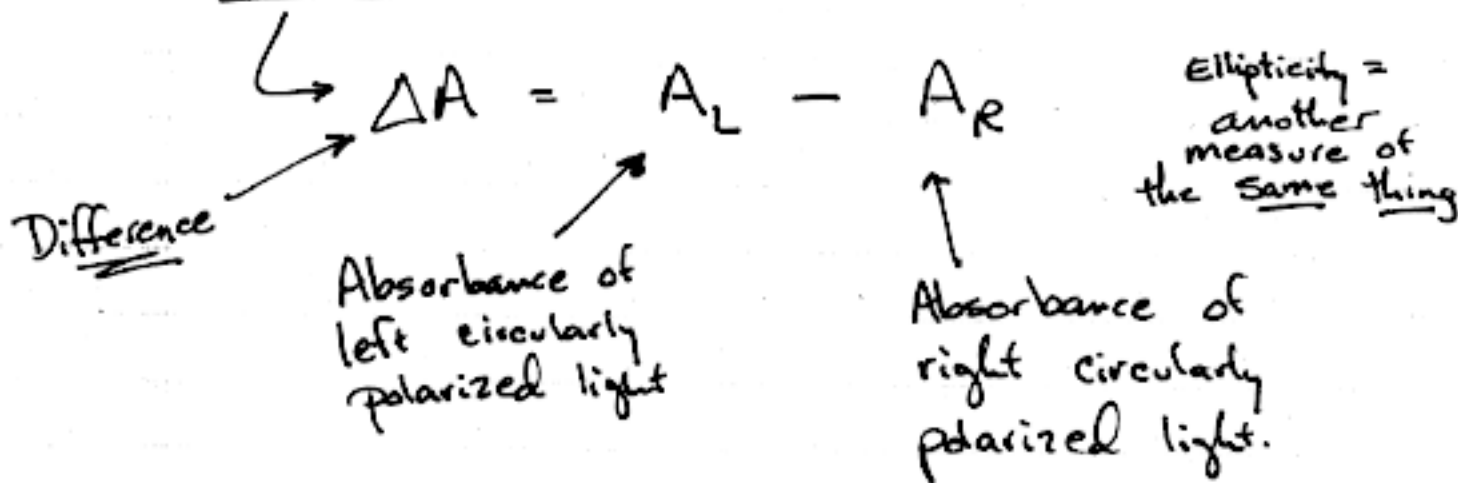
birefringence left - right

Related to refractive index.

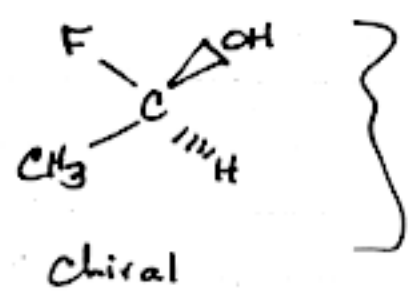
SKIP

Circular Dichroism

IGNORE: ELLIPTICITY
OPTICAL ROTATION



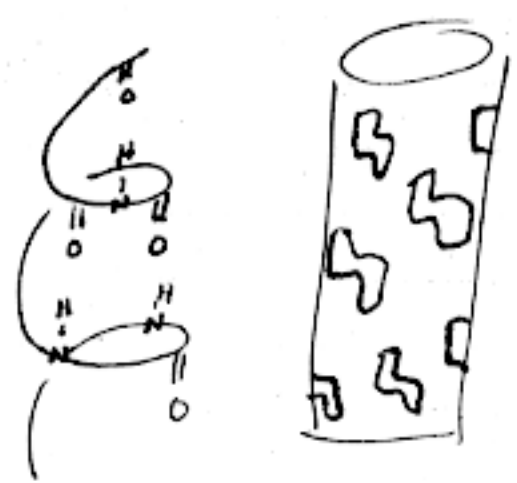
How IT'S USED



} But not interesting to us now.
Q: Why?
A: BIG BOX too little

The common uses arise from Coupling of transitions.

α -helix



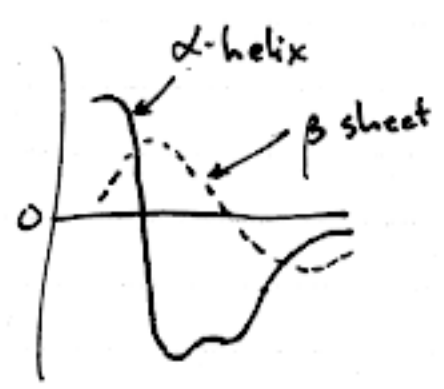
A collection of peptide π -boxes.
They actually interact in a way dependent on their geometrically defined inter-relationships

Result: Absorb more one than the other.

CD pattern is characteristic of a right-handed α -helix

Different pattern for β -sheet.

So \Rightarrow can use CD to distinguish.

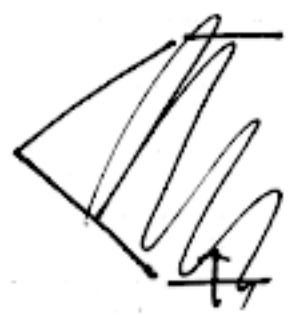


Also turns, etc.

A real protein is then the sum of

- X% α -helix
- Y% β -sheet
- Z% turn
- Reminder "Random"

INTRO TO NMR — Opposite end of energy scale (small)
 \therefore long wavelength (radio frequency)



Apply \vec{B}_0

\uparrow or \downarrow Same energy

\uparrow lower E

\uparrow higher E

