

## Which version of the Chem 250 Final Exam do I have?

### Version 1

1. Which molecule below has the **highest** boiling point?
  2. Which two molecules above are constitutional isomers?  
1) **A and D**    2) C and D    3) B and C  
4) **A and B**    5) none are constitutional isomers
  4. In the molecule at right, the ideal bond angle around the **1-carbon** is:
- 

### Version 2

1. Which molecule below has the **lowest** boiling point?
2. Which two molecules above are constitutional isomers?  
1) **A and B**    2) C and D    3) B and C  
4) **A and D**    5) none are constitutional isomers
4. In the molecule at right, the ideal bond angle around the **4-carbon** is:

### Version 3

1. Which molecule below has the **lowest** boiling point?
2. Which two molecules above are constitutional isomers?  
1) **A and D**    2) C and D    3) B and C  
4) **A and B**    5) none are constitutional isomers
4. In the molecule at right, the ideal bond angle around the **1-carbon** is: