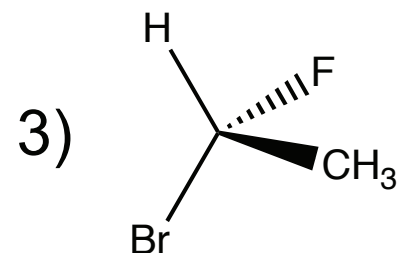
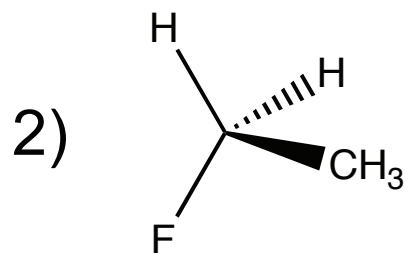
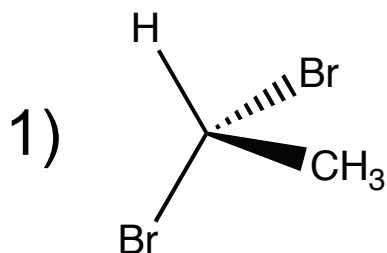


Which has a chiral center?

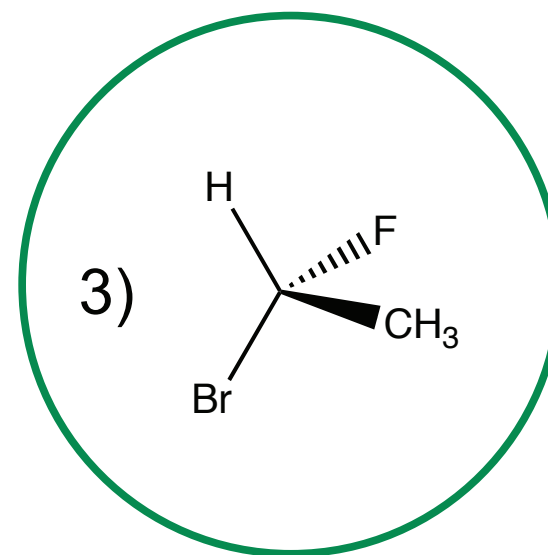
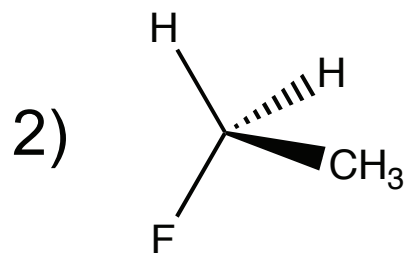
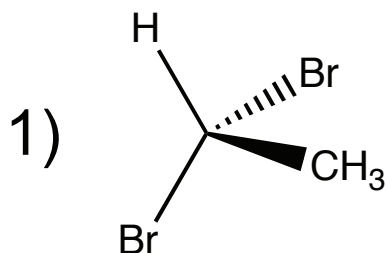


4) None of the above

5) All of the above



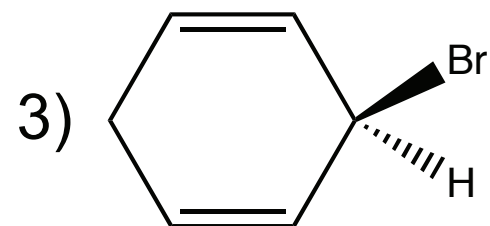
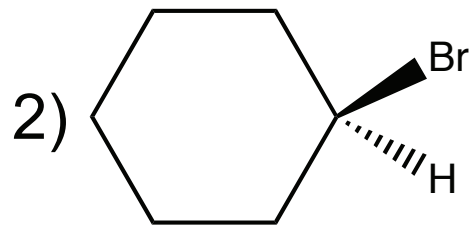
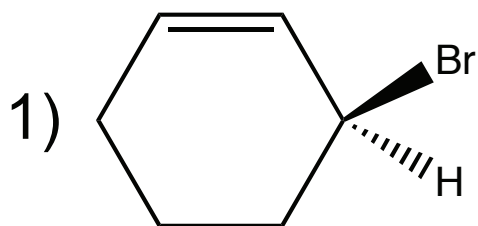
Which has a chiral center?



4) None of the above

5) All of the above

Which has a chiral center?

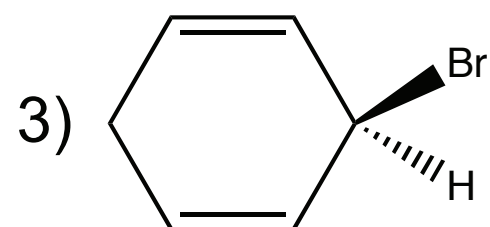
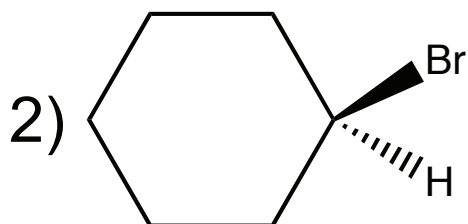
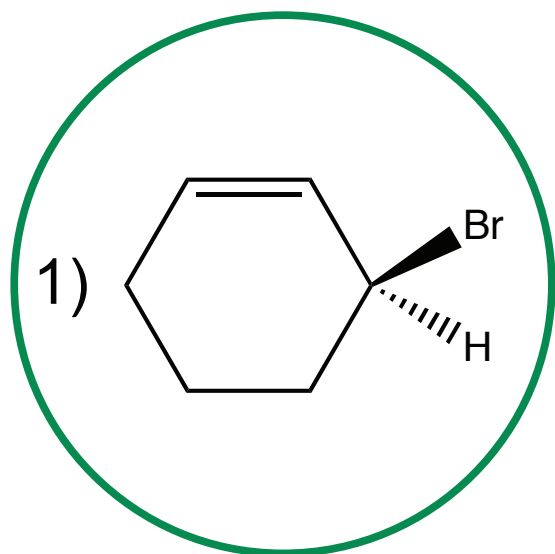


4) None of the above

5) All of the above



Which has a chiral center?



4) None of the above

5) All of the above

Rank boiling points Lowest → Highest

- 1) CH_3CH_3 CH_3NH_2 CH_3OH
- 2) CH_3CH_3 CH_3OH CH_3NH_2
- 3) CH_3NH_2 CH_3OH CH_3CH_3
- 4) They are all about the same

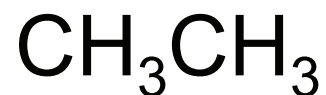
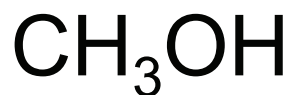
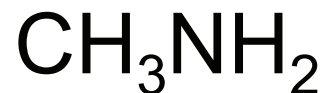
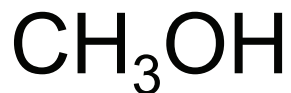
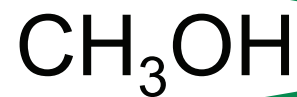
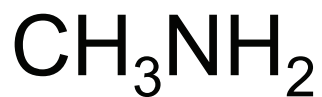


Rank boiling points Lowest → Highest

Only London
dispersion forces

Limited H-bonding
interactions

Strong H-bonding
interactions



4) They are all about the same