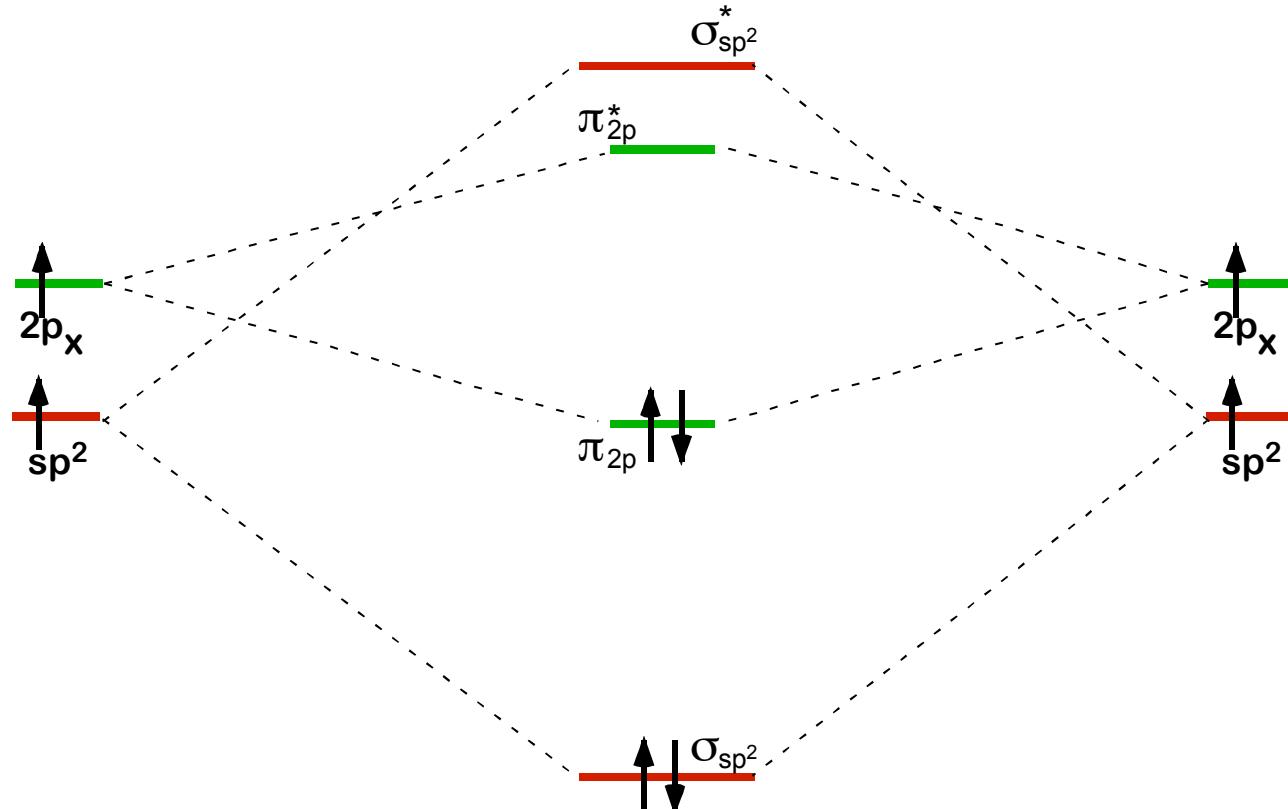
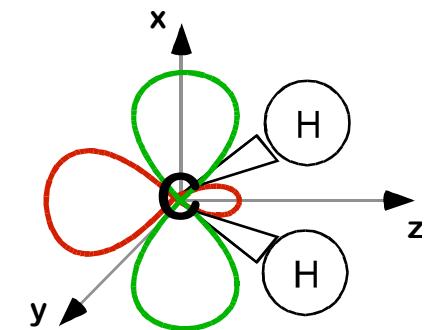
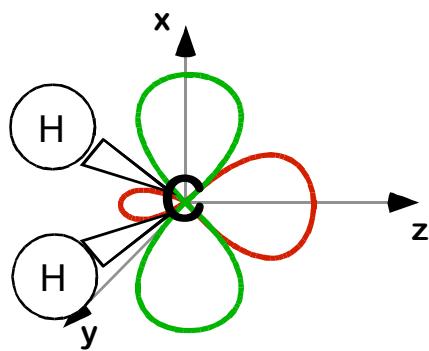




Bond Order = $(4-0)/2 = 2$

4 electrons
between the carbons



Atomic Hybrid Orbitals

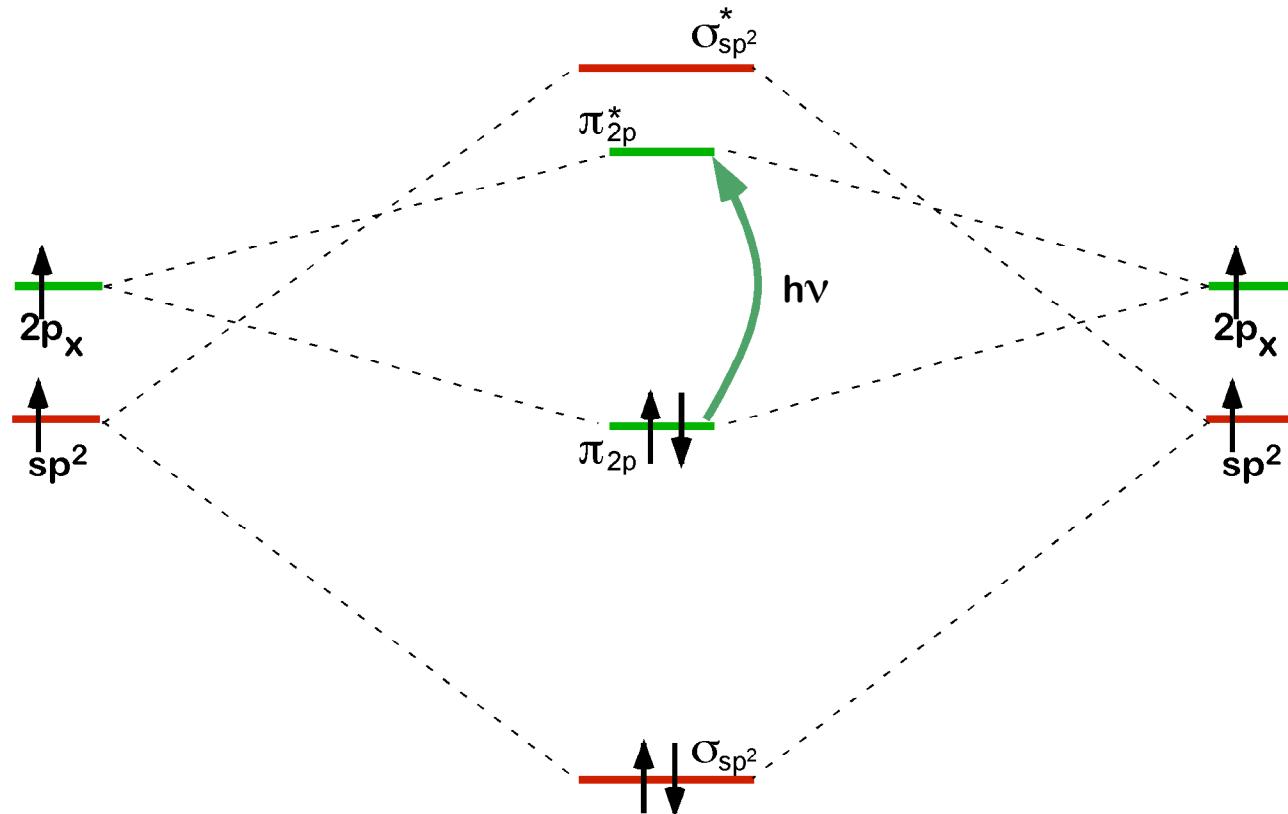
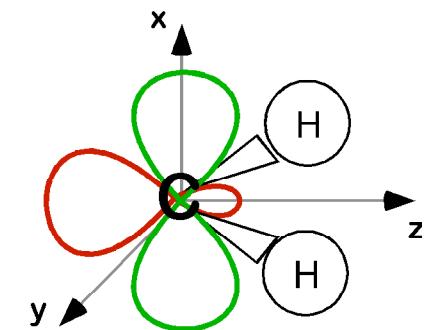
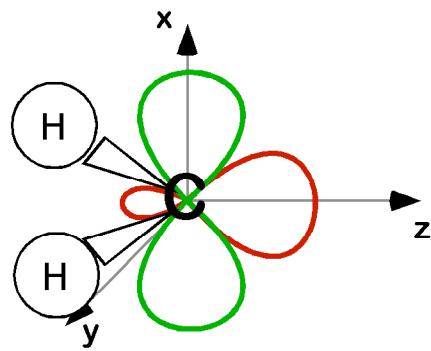
Molecular Orbitals

Atomic Hybrid Orbitals



Bond Order = $(4-0)/2 = 2$

4 electrons
between the carbons



Atomic Hybrid Orbitals

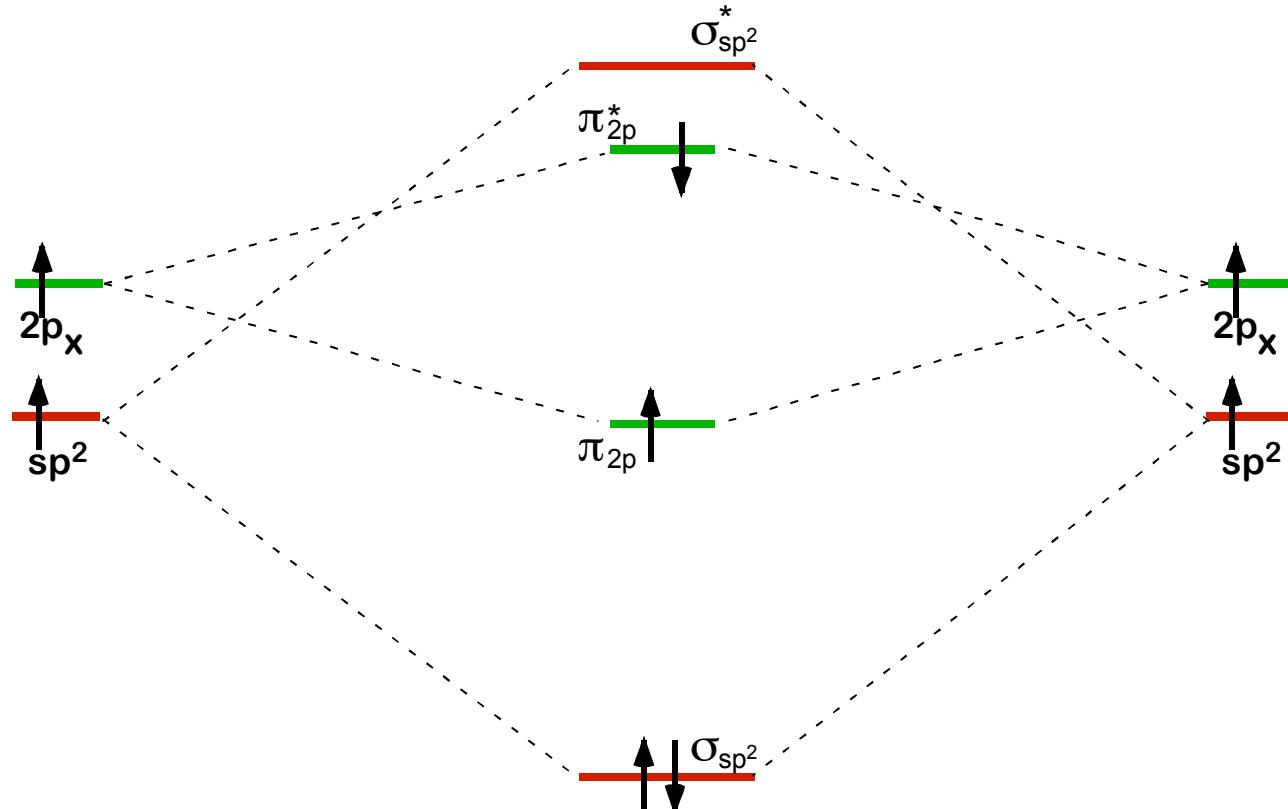
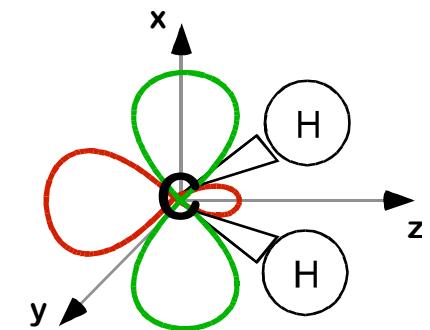
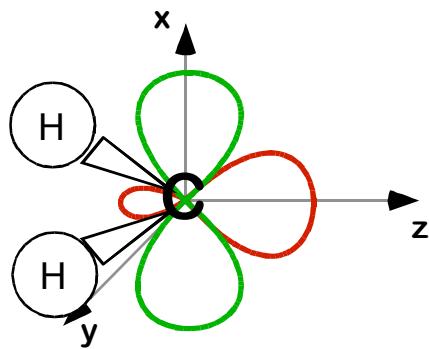
Molecular Orbitals

Atomic Hybrid Orbitals



Bond Order = $(3-1)/2 = 1$

4 electrons
between the carbons



Atomic Hybrid Orbitals

Molecular Orbitals

Atomic Hybrid Orbitals