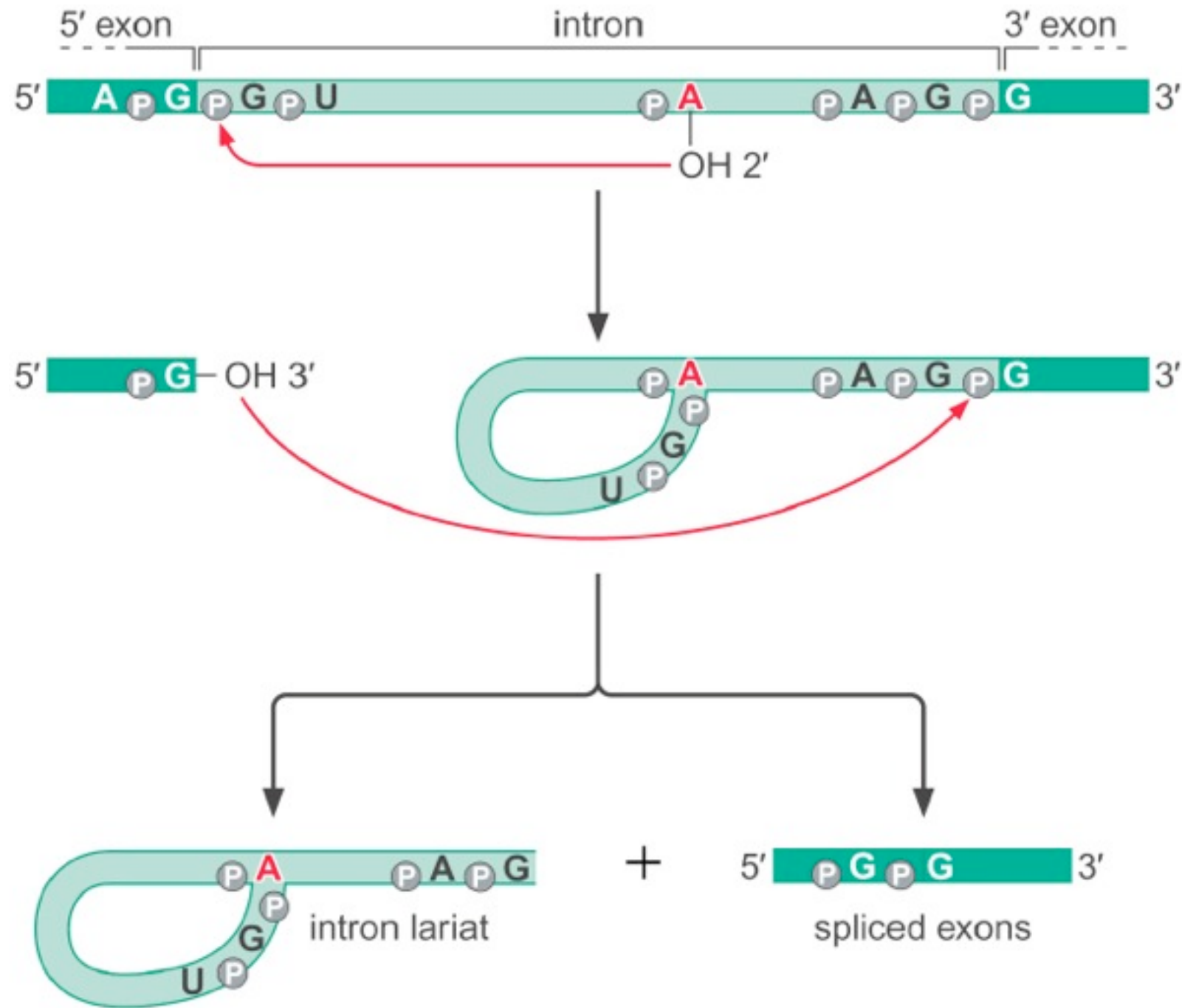


Nucleic Acid Structure

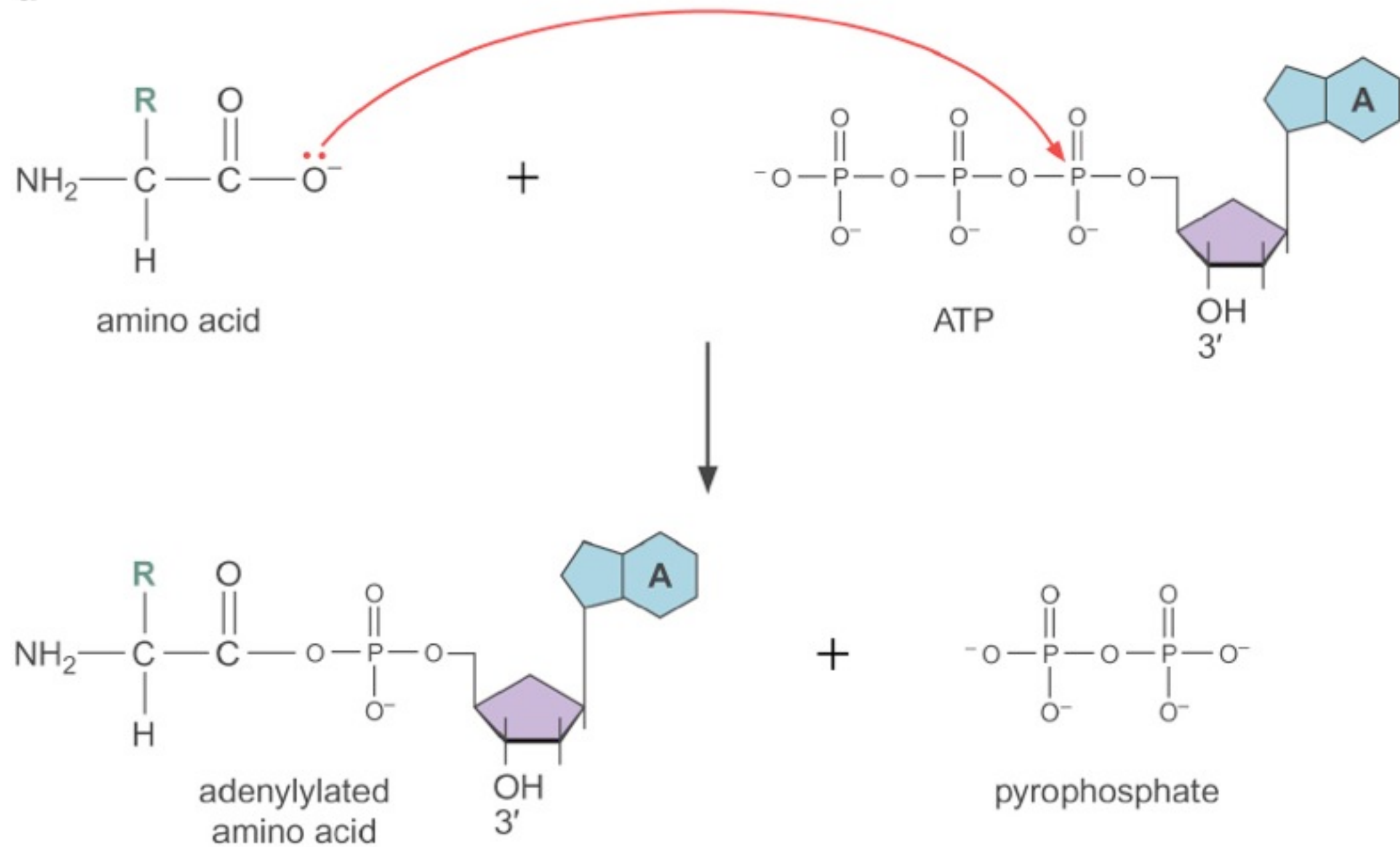
- Professor Craig Martin
- Dept of Chemistry
- <http://people.chem.umass.edu/cmartin>

RNA Splicing

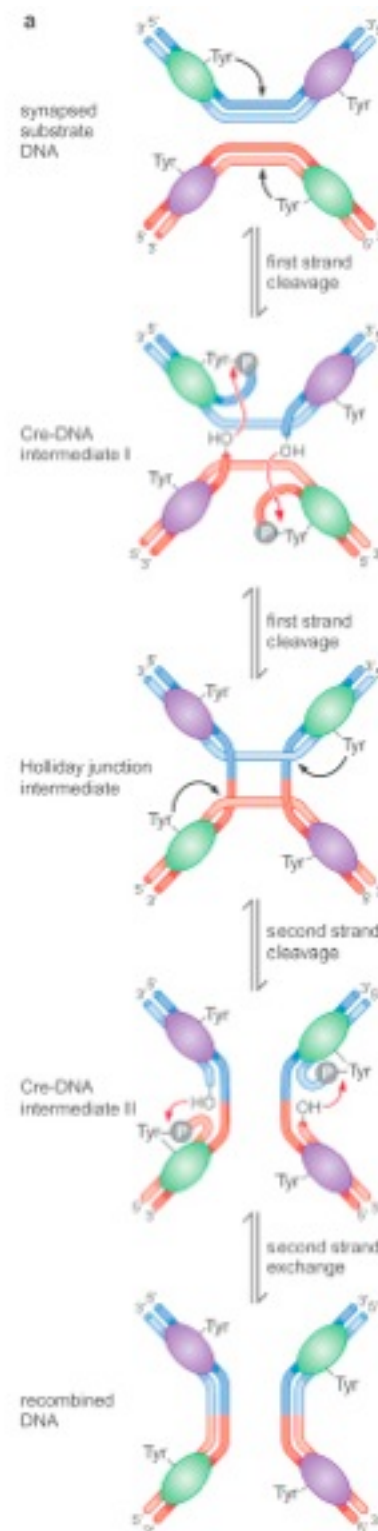


Translation

a



Recombination



DNA

DNA

A, C, G, and T - right?

Which is more stable?
(which has a higher melting temperature?)

ACCGCCACCGAAG
TGGCGGTGGCTTC

or

ACCGCCACCGAAG
TGGCGGTGGCTTA

Which is more stable?
(which has a higher melting temperature?)

ACCGCCACCGAAG
TGGCGGTGGCTTC

51.6° C

or

ACCGCCACCGAAG
TGGCGGTGGCTTA

Which is more stable?
(which has a higher melting temperature?)

ACCGCCACCGAAG
TGGCGGTGGCTTC

51.6° C

or

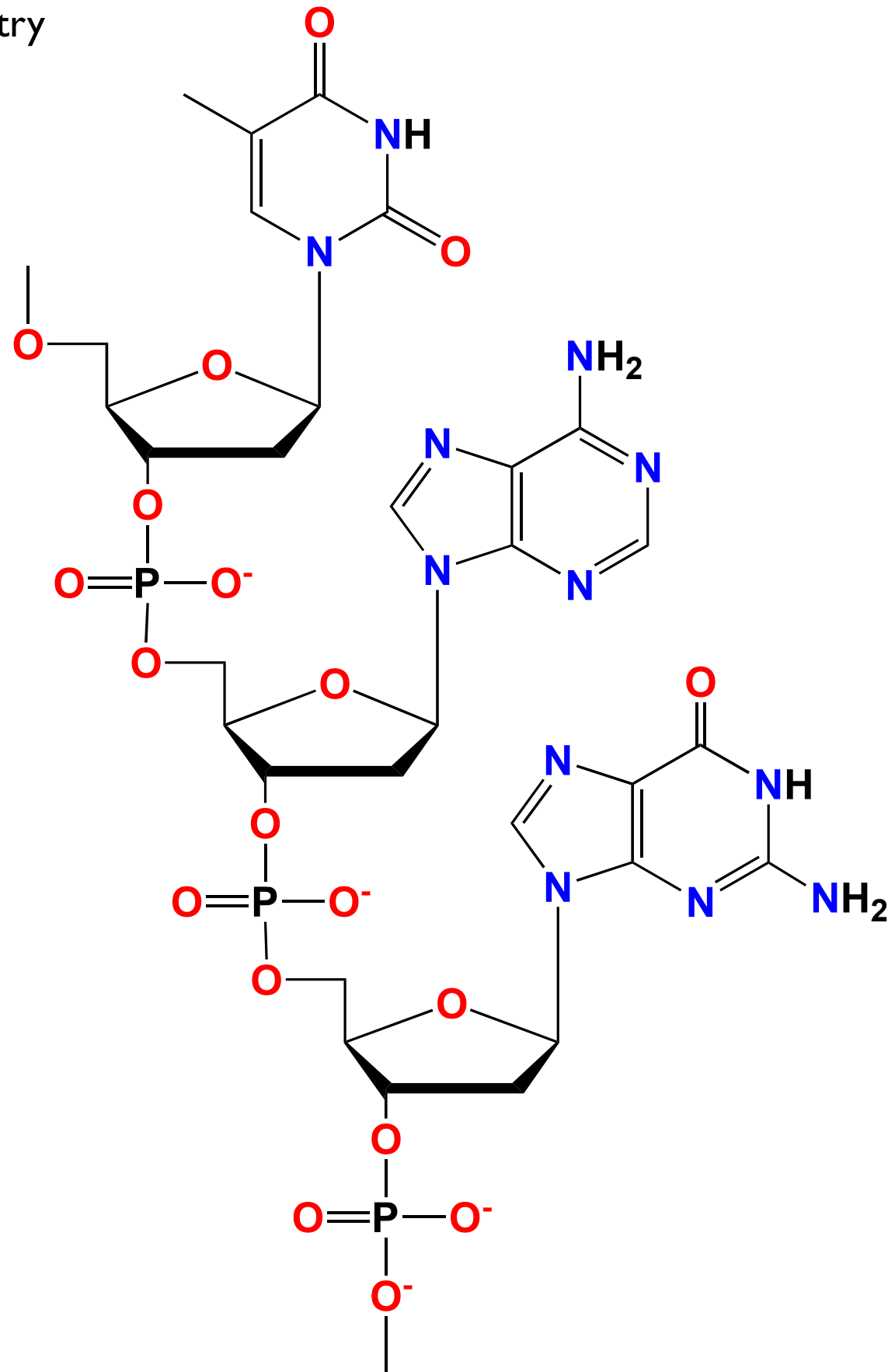
ACCGCCACCGAAG
TGGCGGTGGCTTA

52.5° C

Calculations from <http://www.idtdna.com/analyzer/Applications/OligoAnalyzer/>

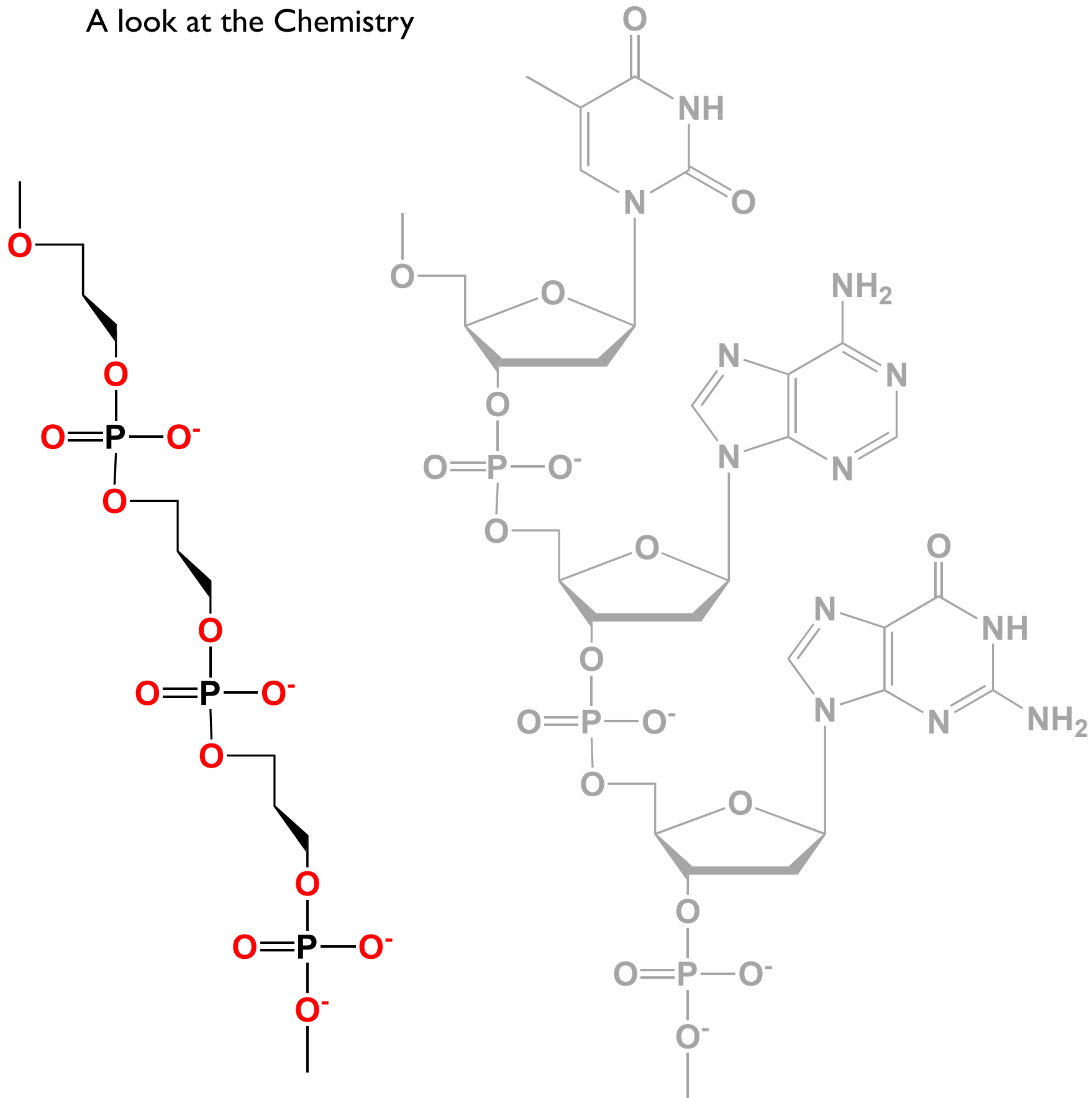
DNA

A look at the Chemistry



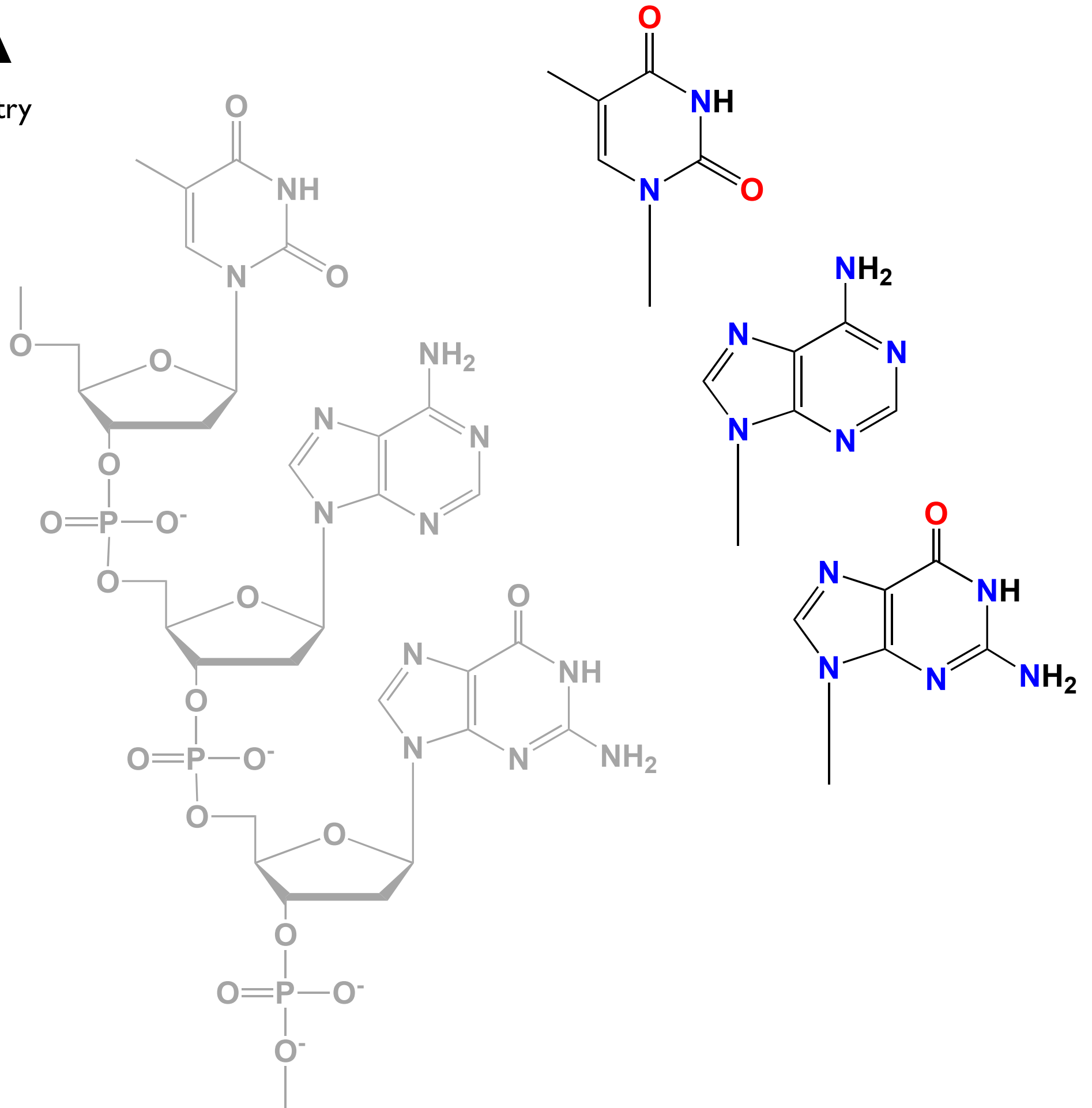
DNA

A look at the Chemistry



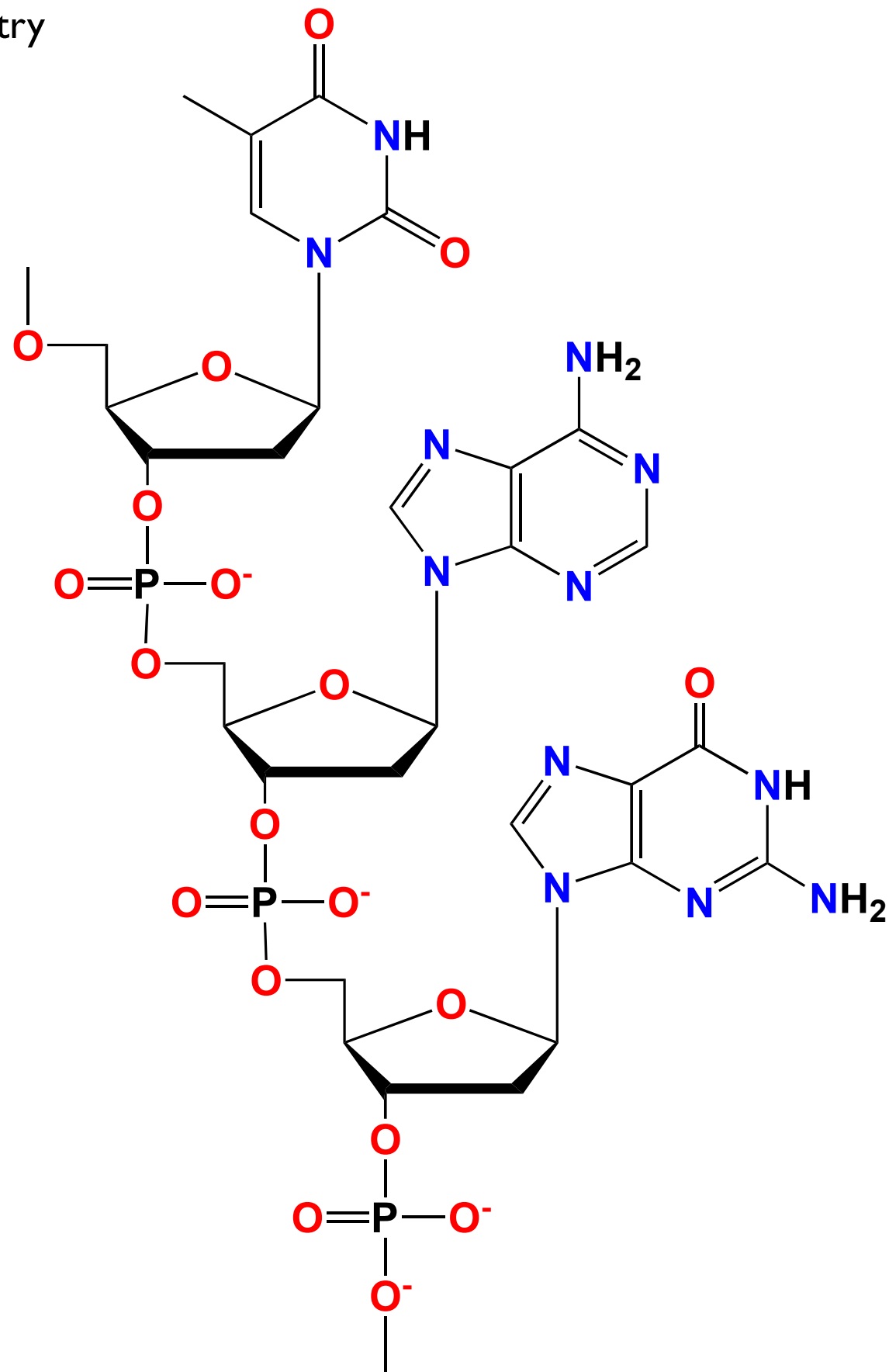
DNA

A look at the Chemistry



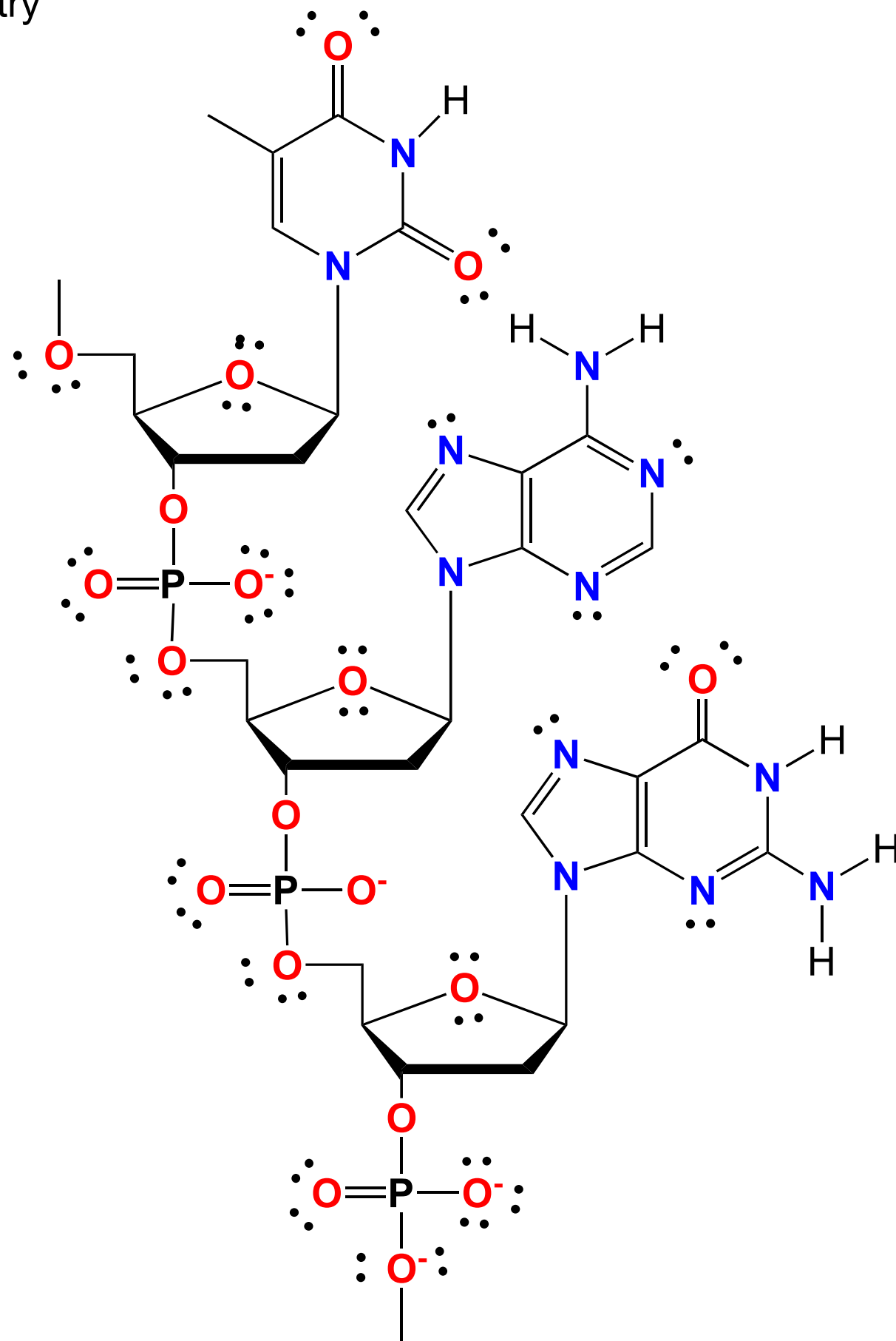
DNA

A look at the Chemistry



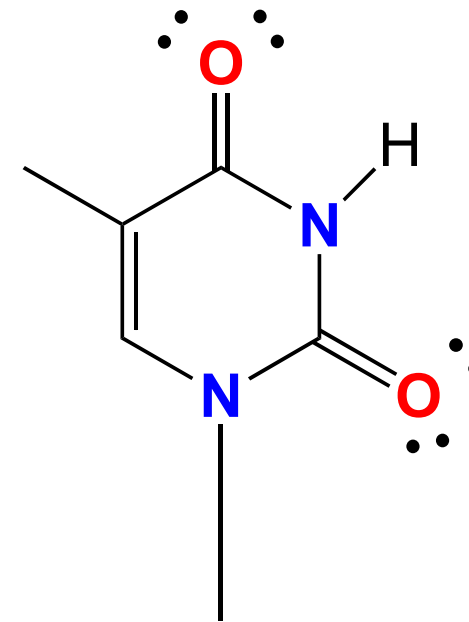
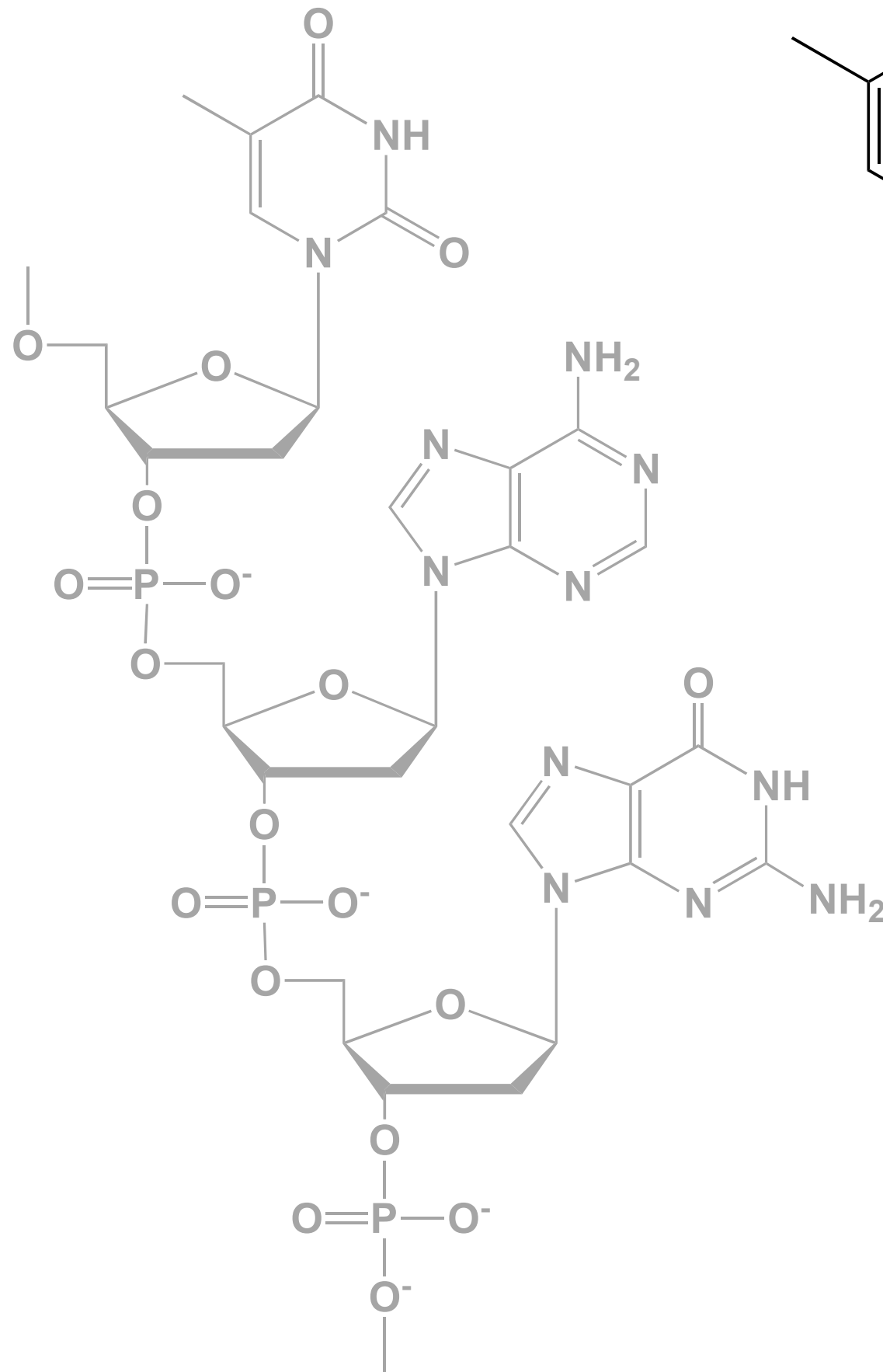
DNA

A look at the Chemistry



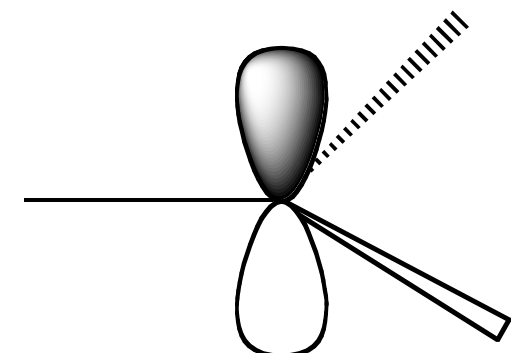
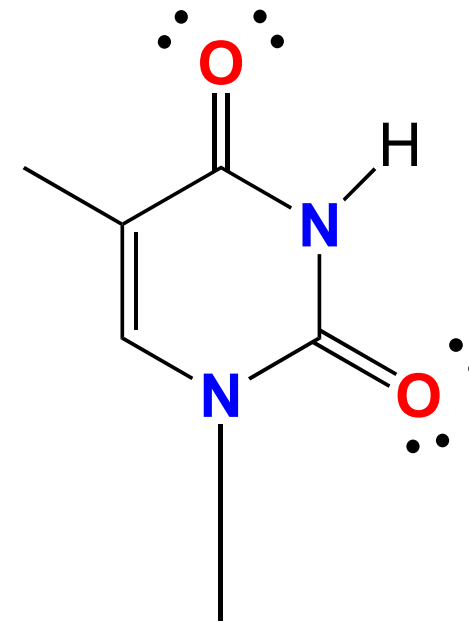
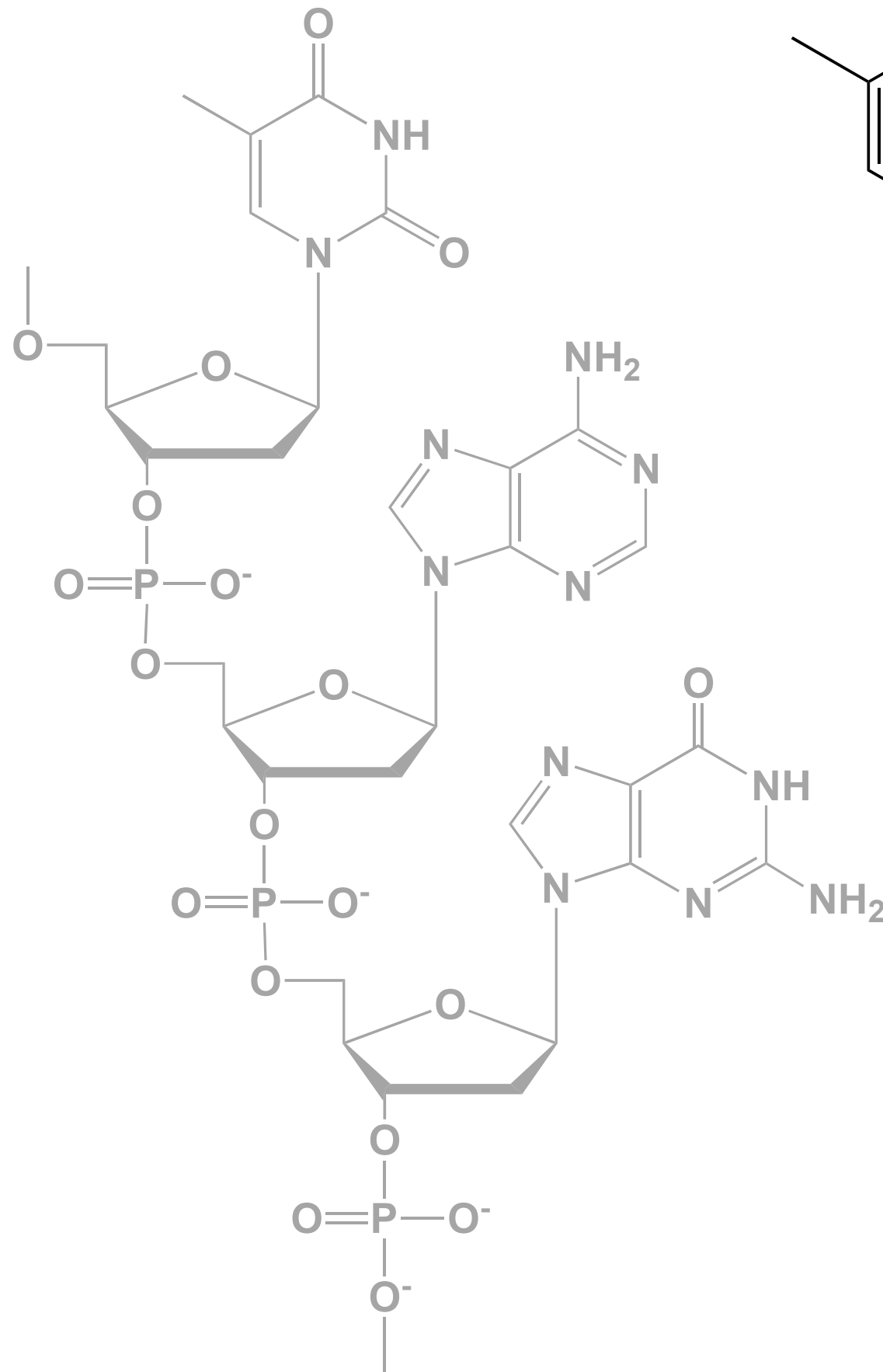
DNA

A look at the Chemistry



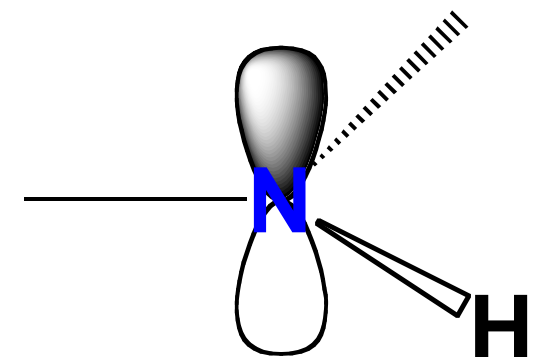
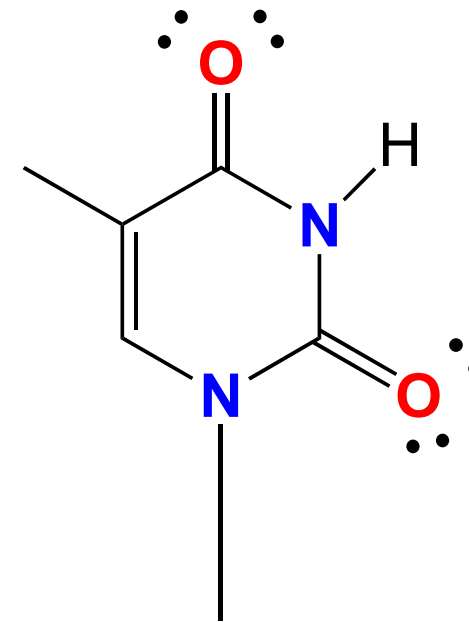
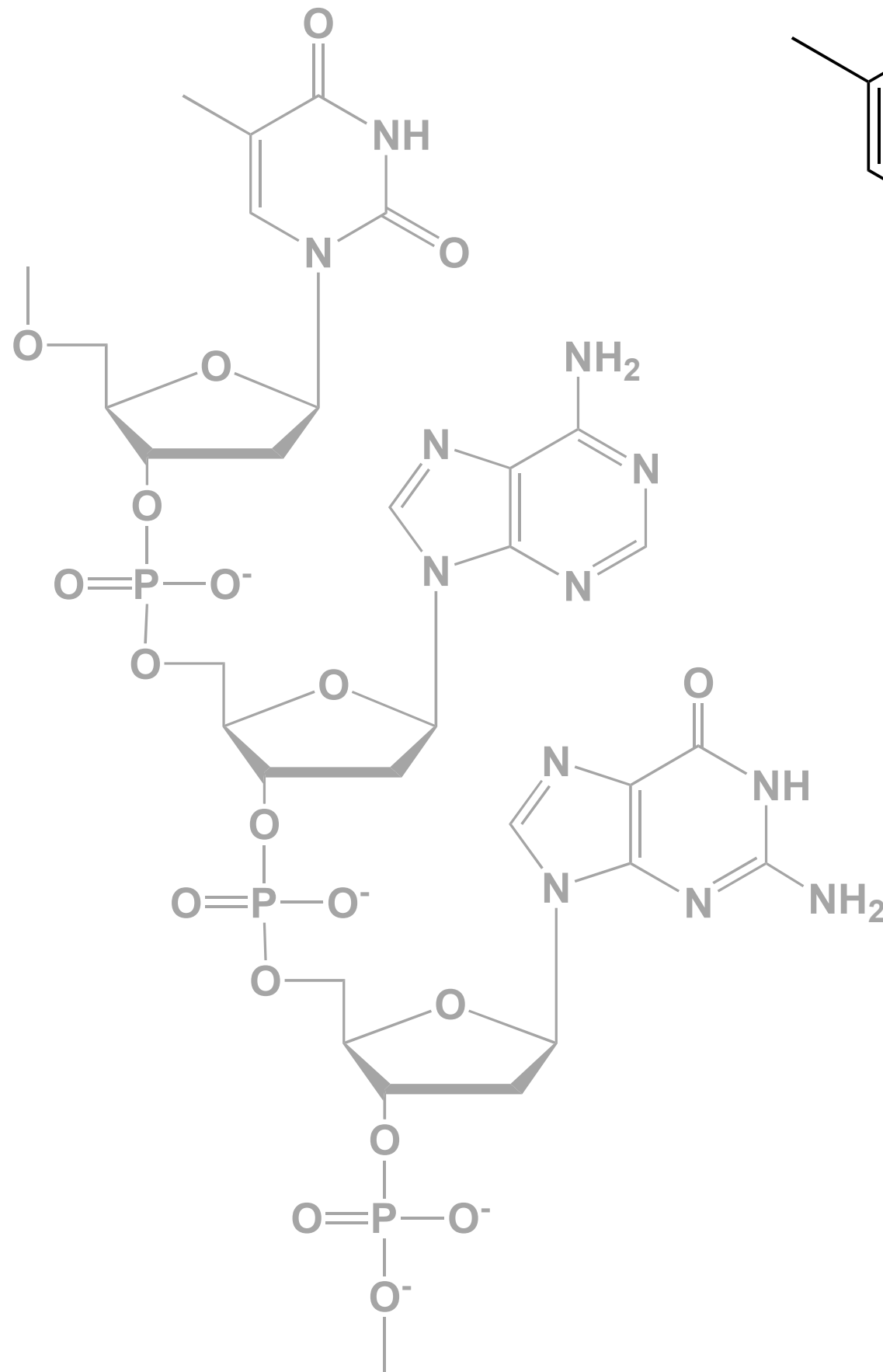
DNA

A look at the Chemistry



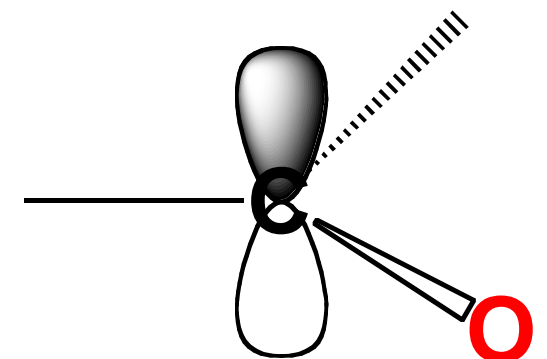
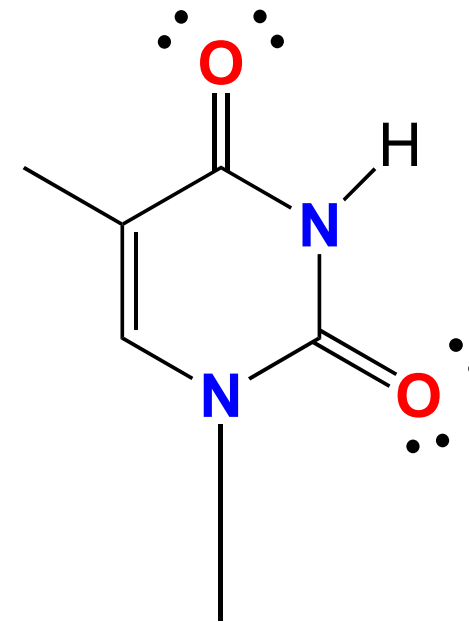
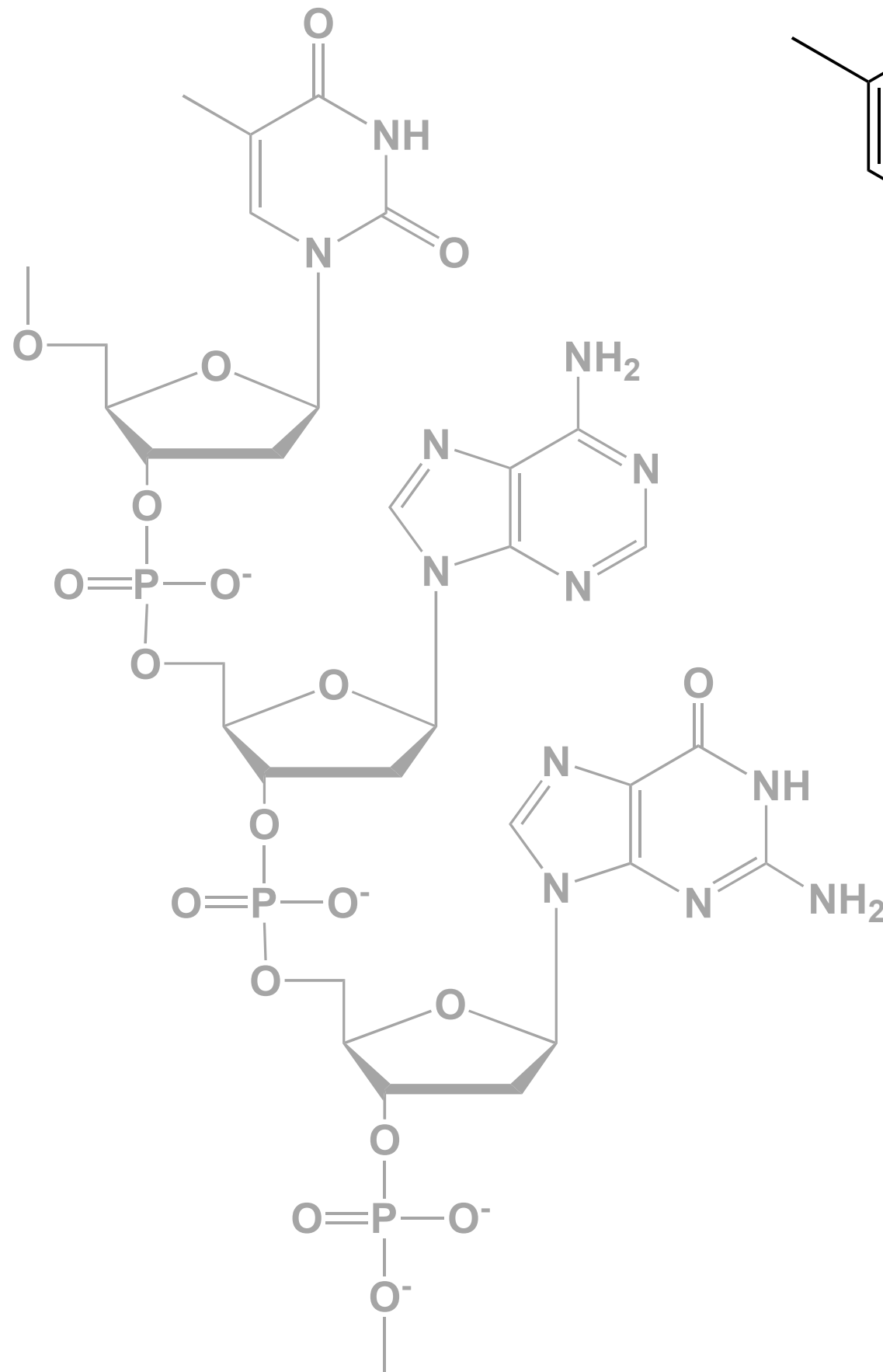
DNA

A look at the Chemistry



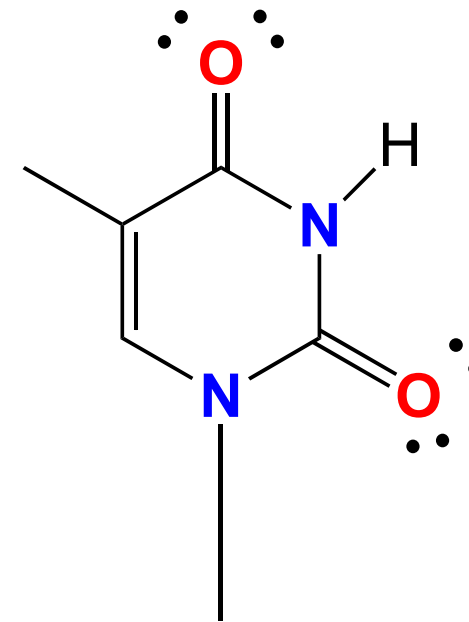
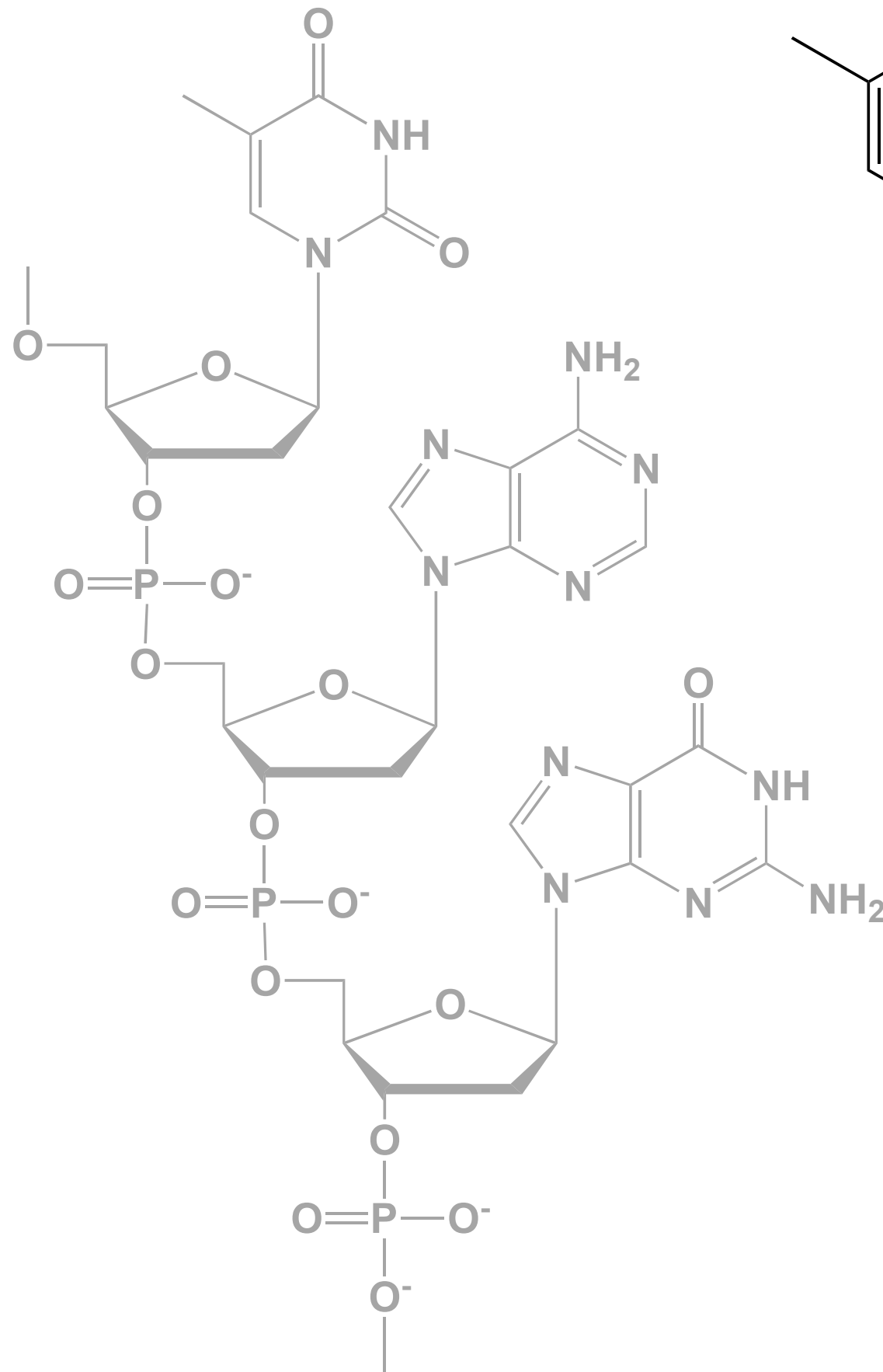
DNA

A look at the Chemistry



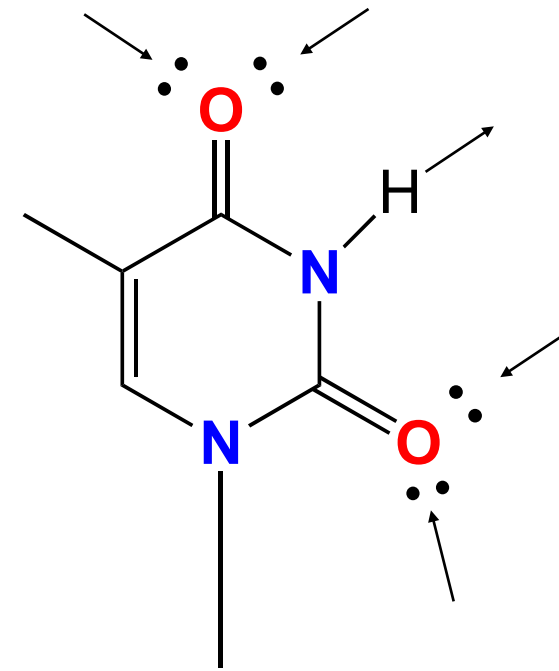
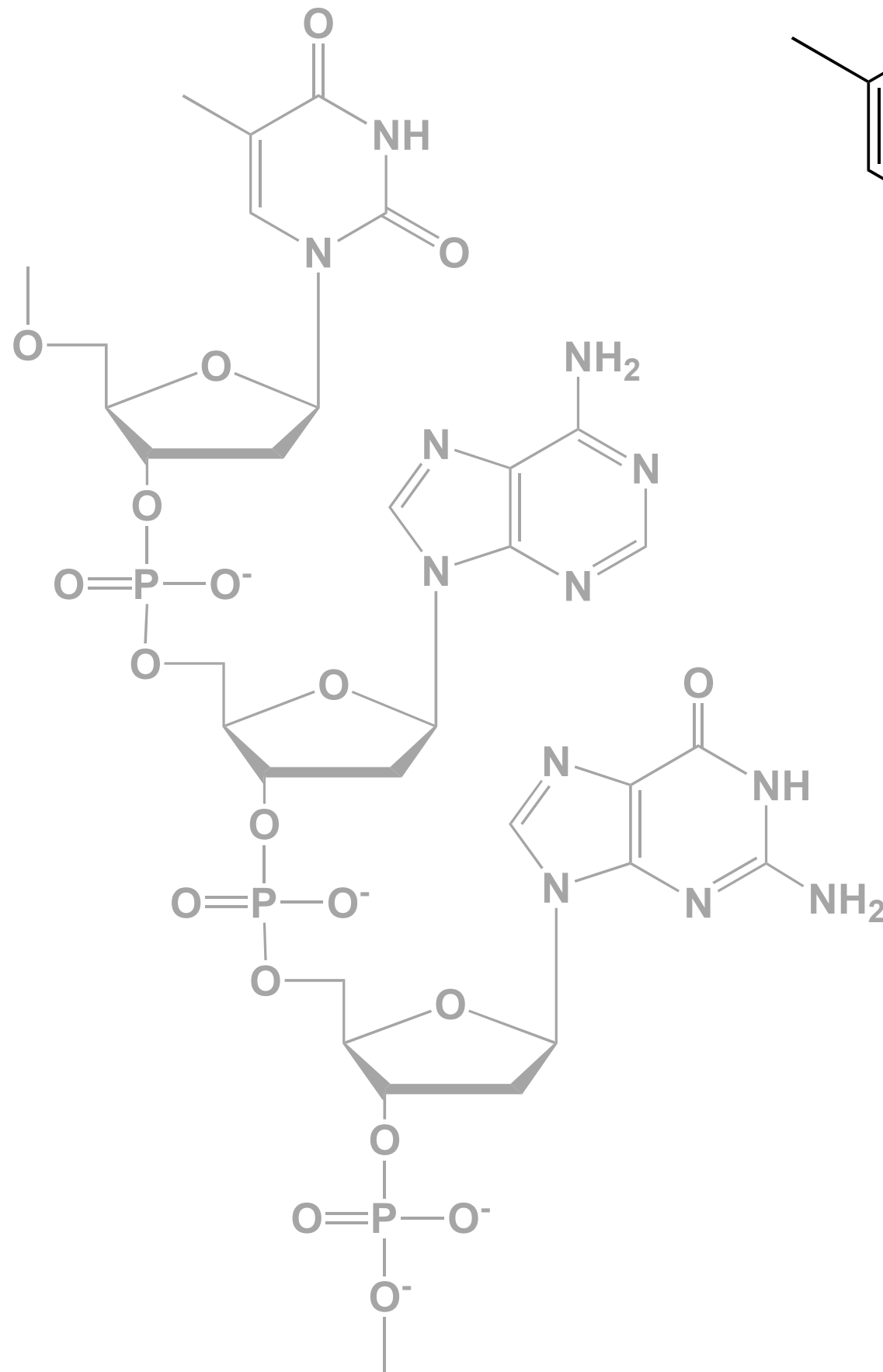
DNA

A look at the Chemistry



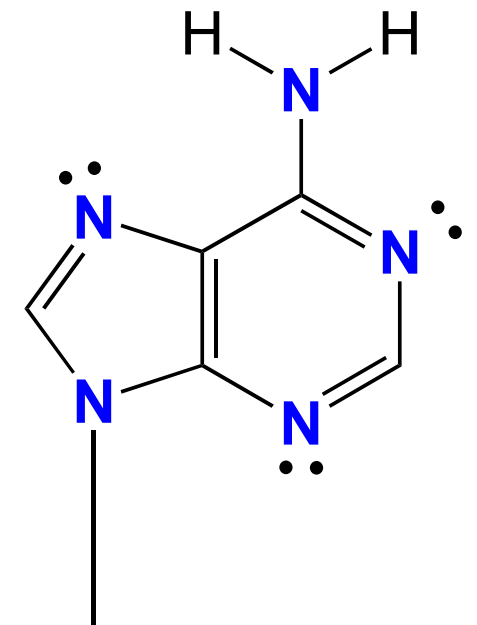
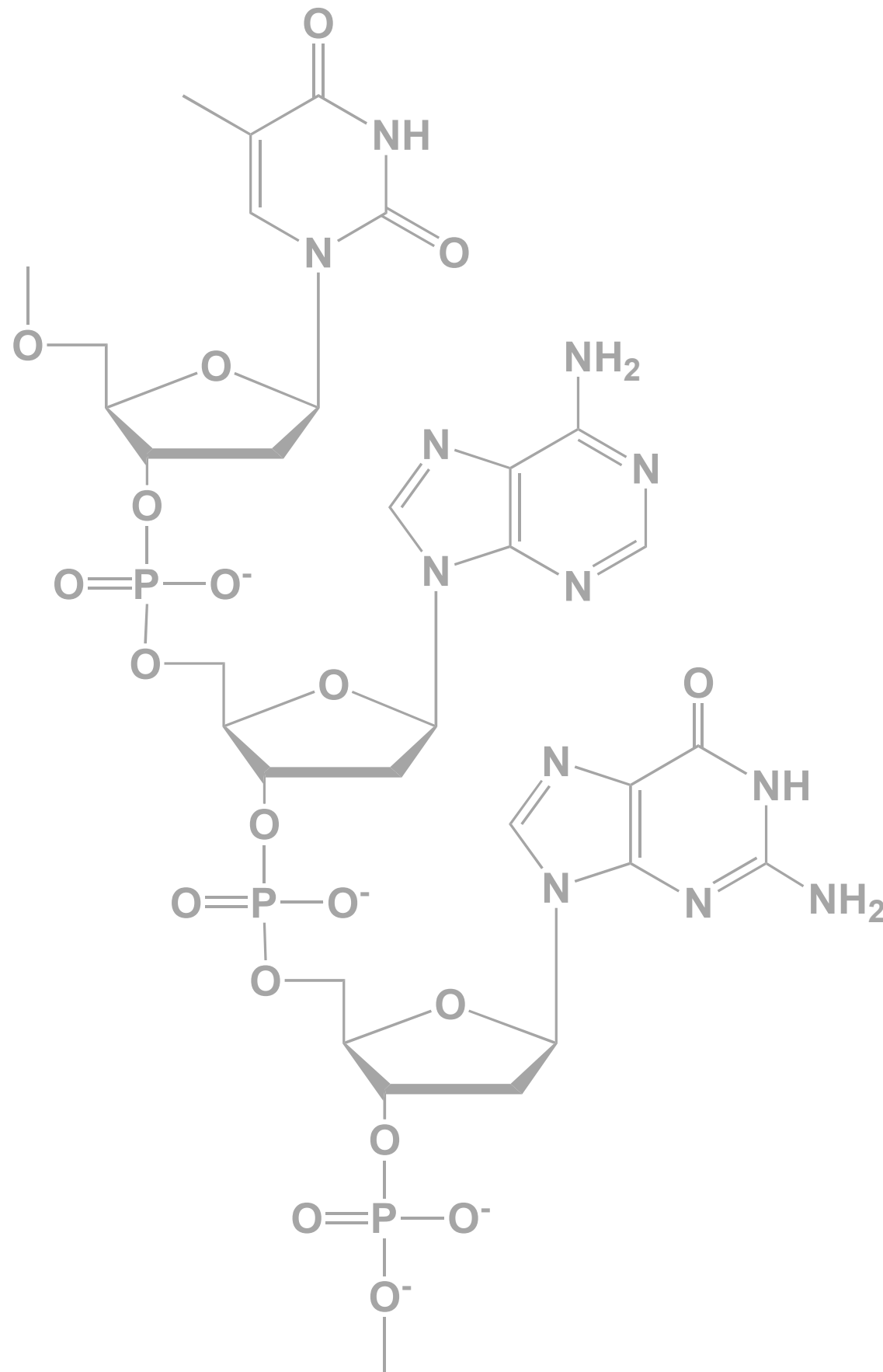
DNA

A look at the Chemistry



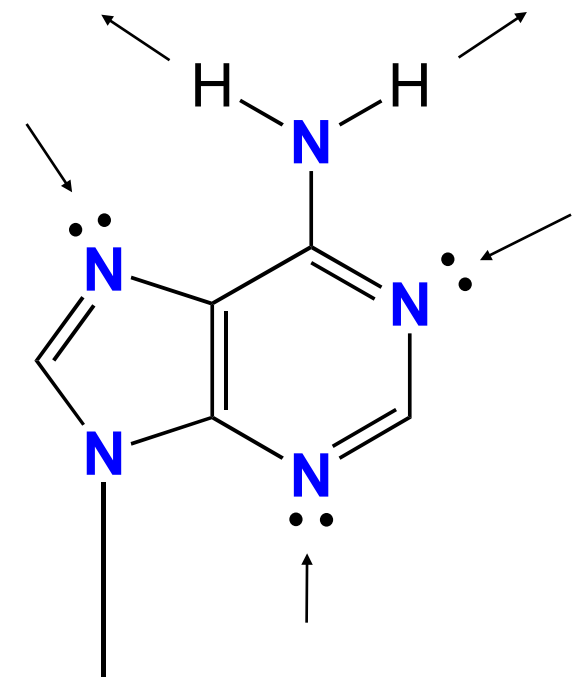
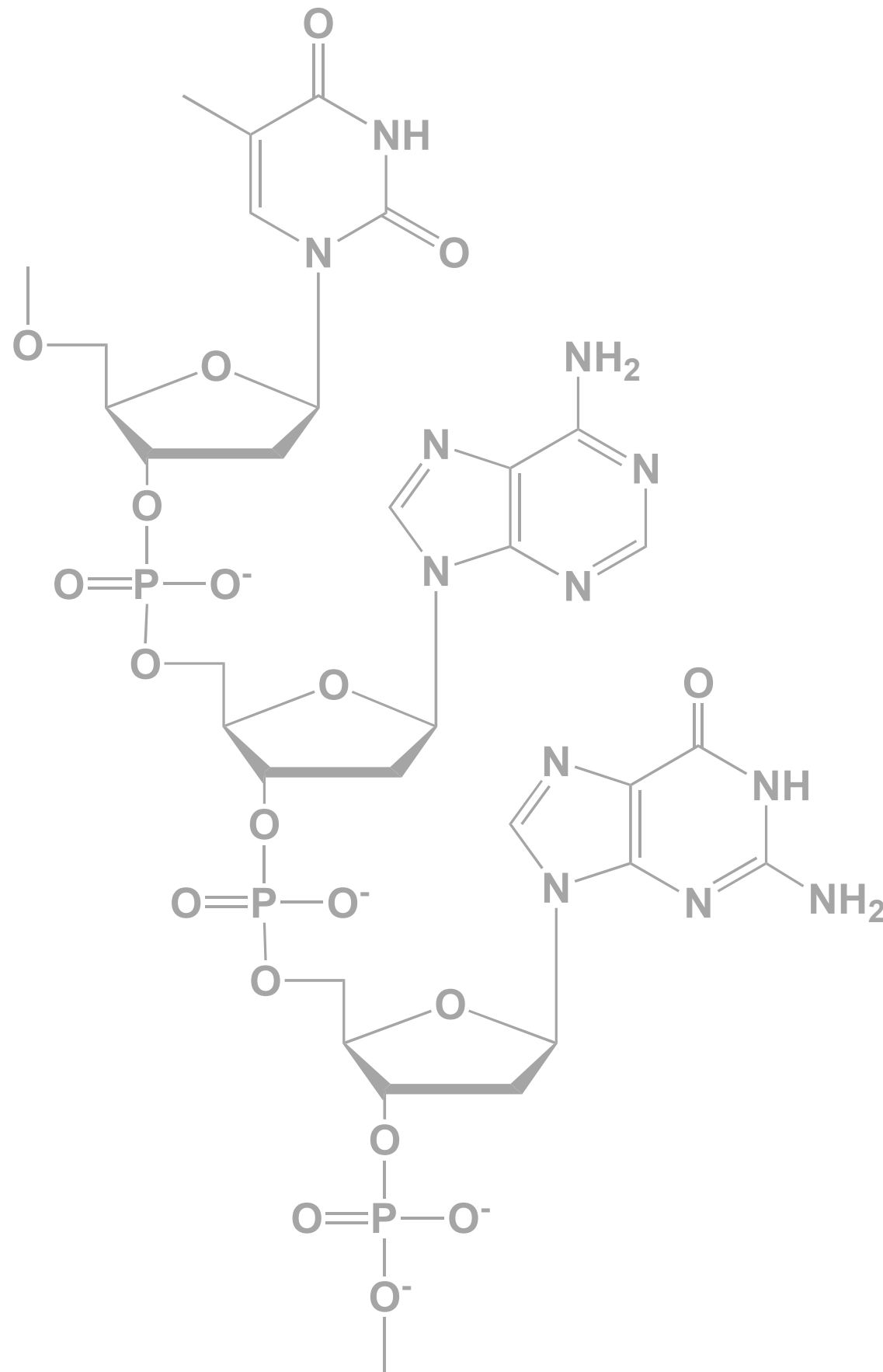
DNA

A look at the Chemistry



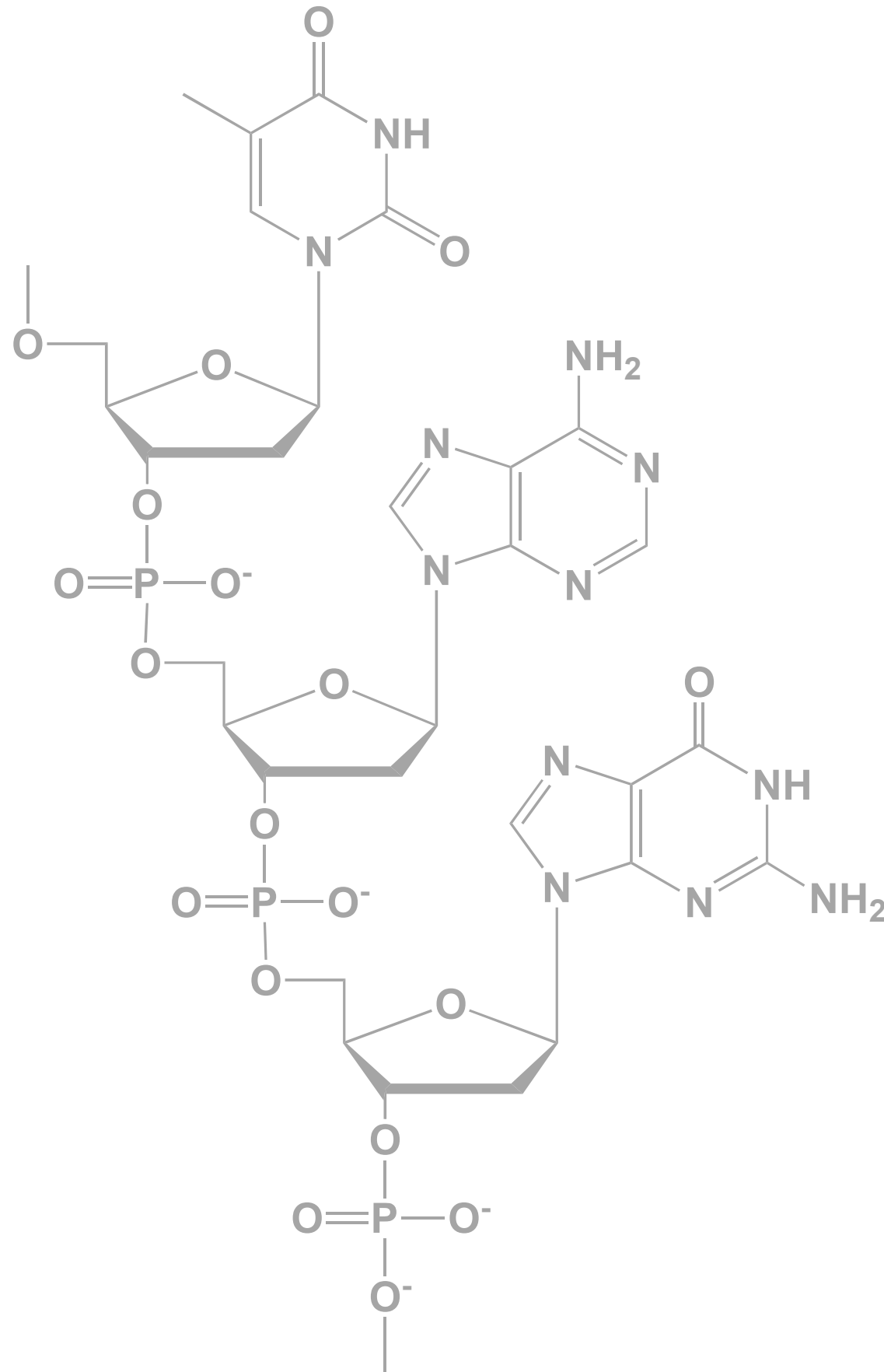
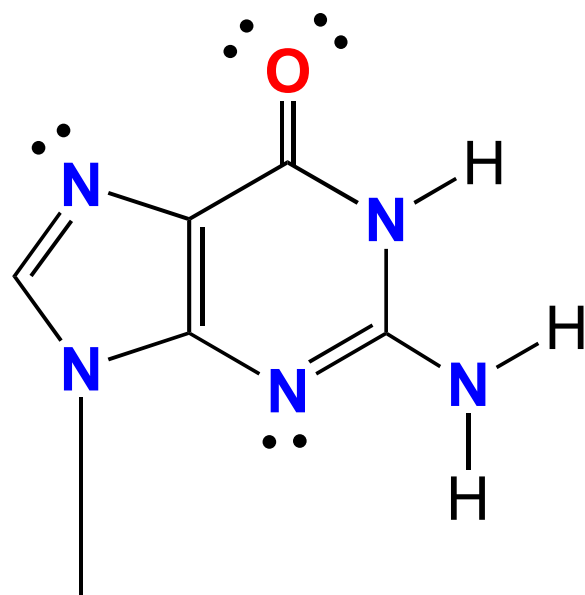
DNA

A look at the Chemistry



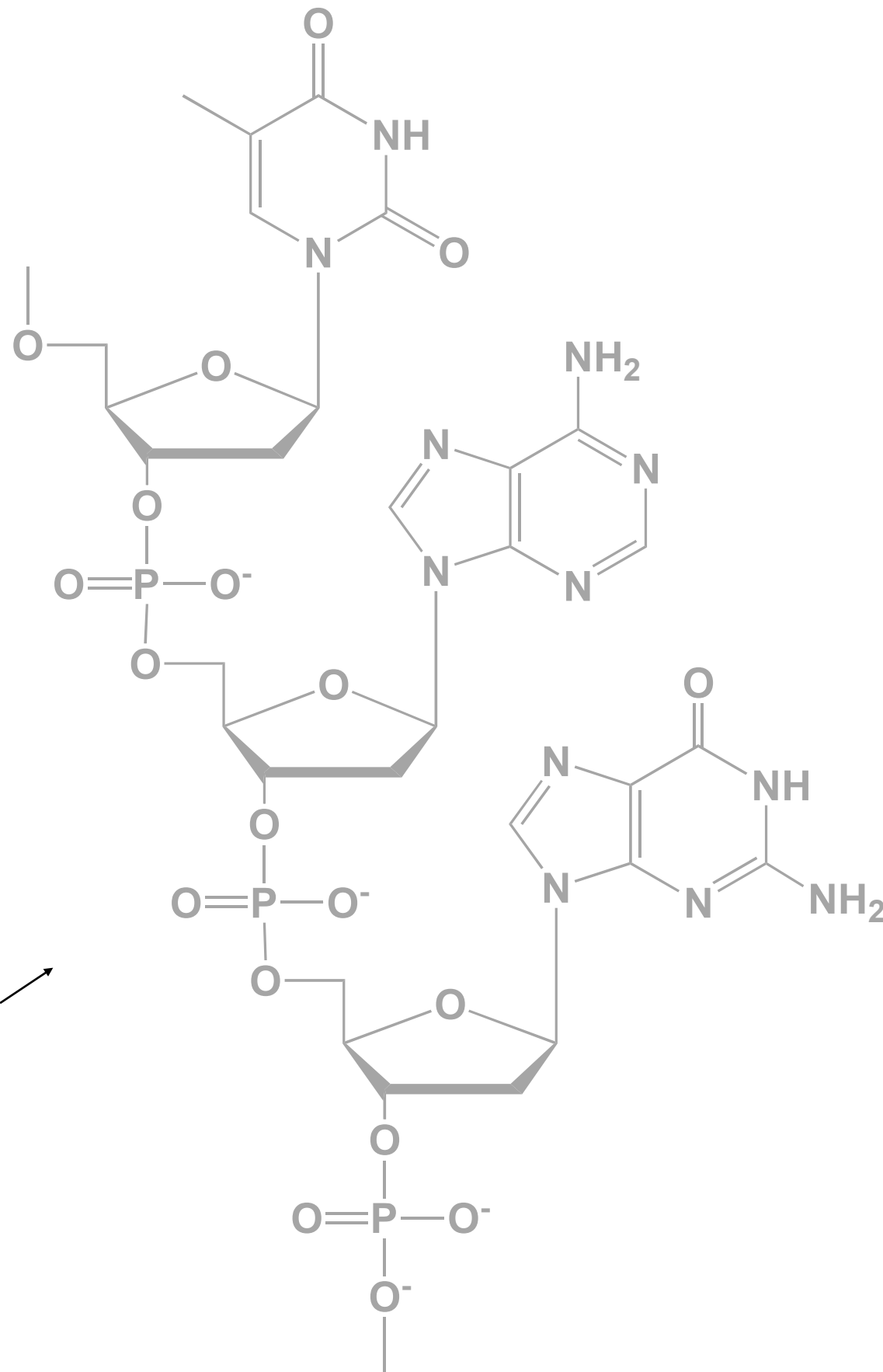
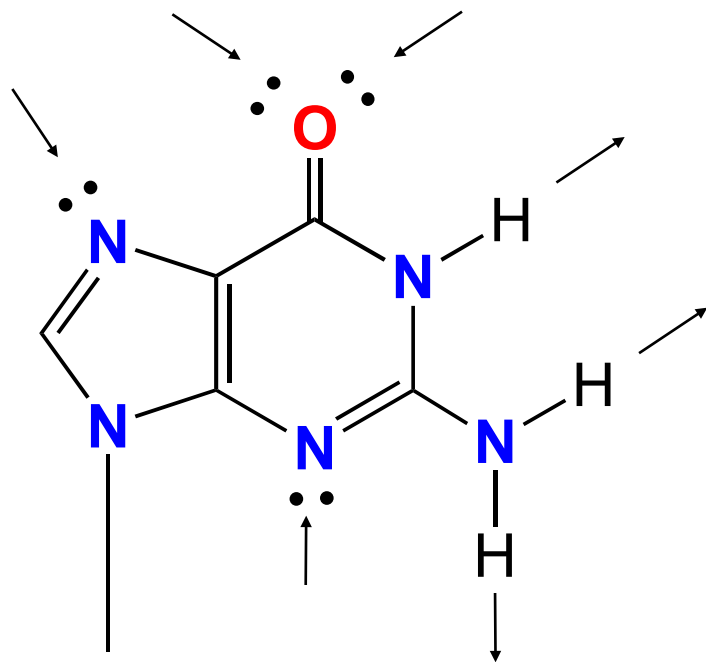
DNA

A look at the Chemistry

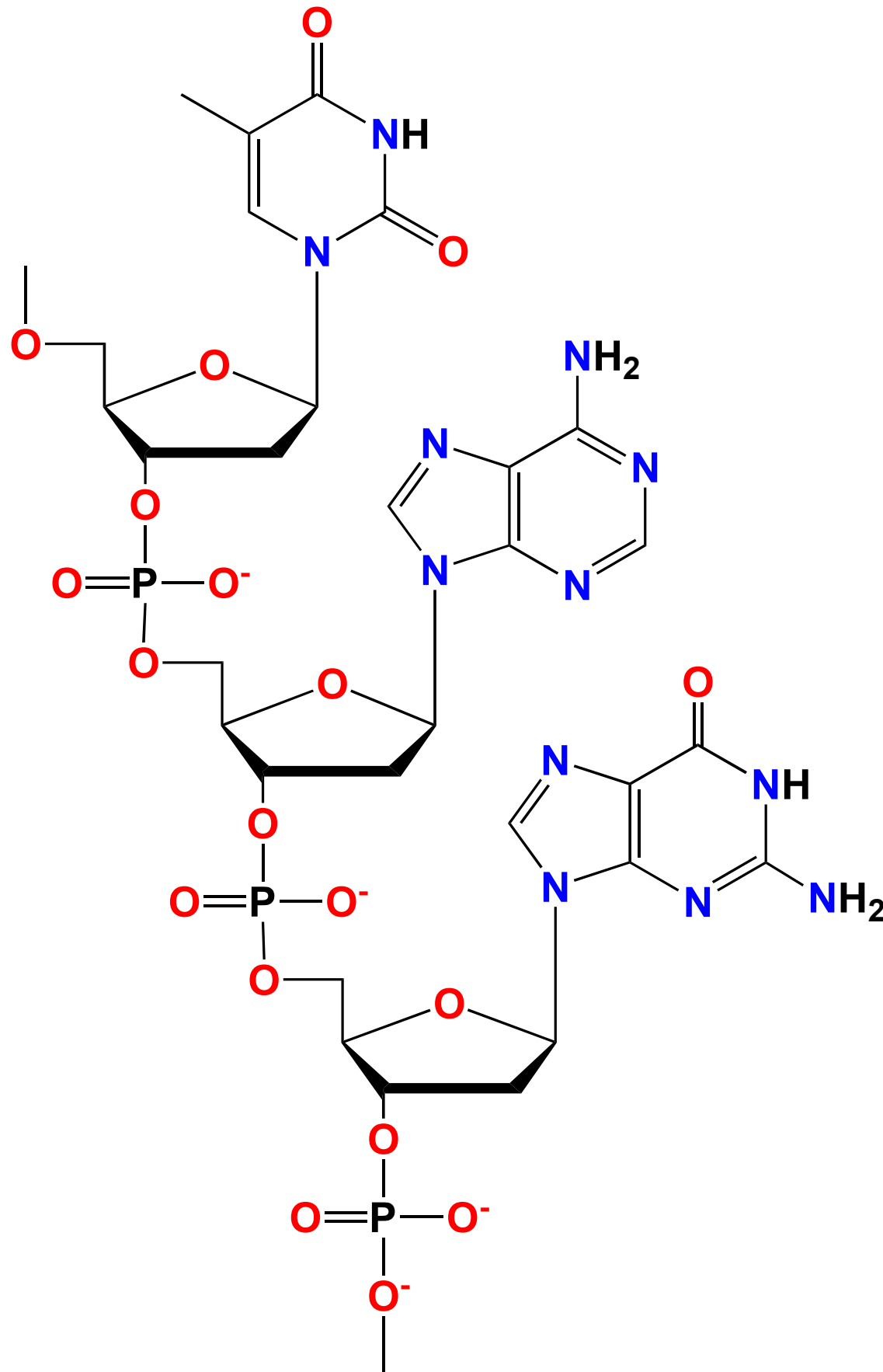


DNA

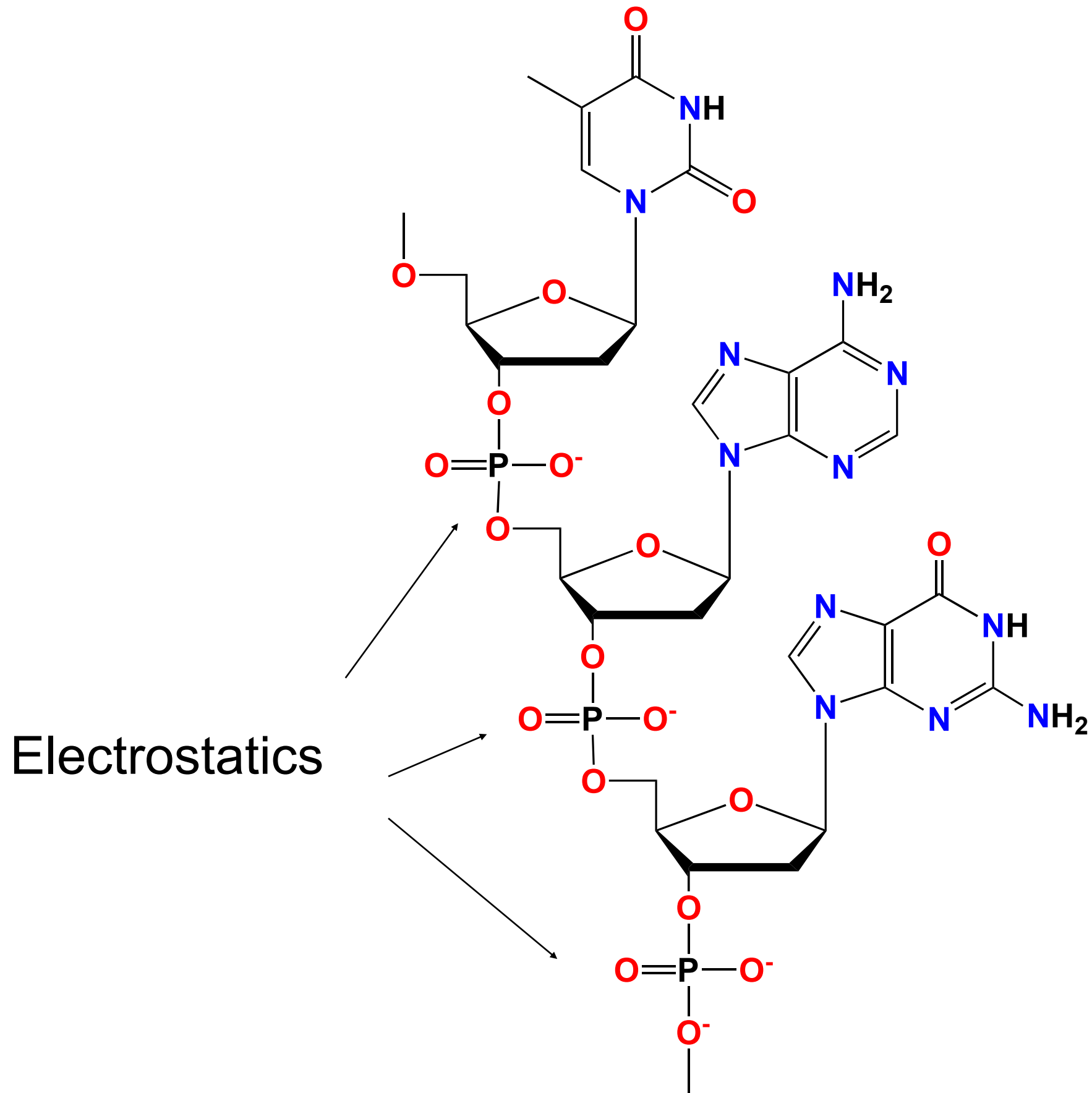
A look at the Chemistry



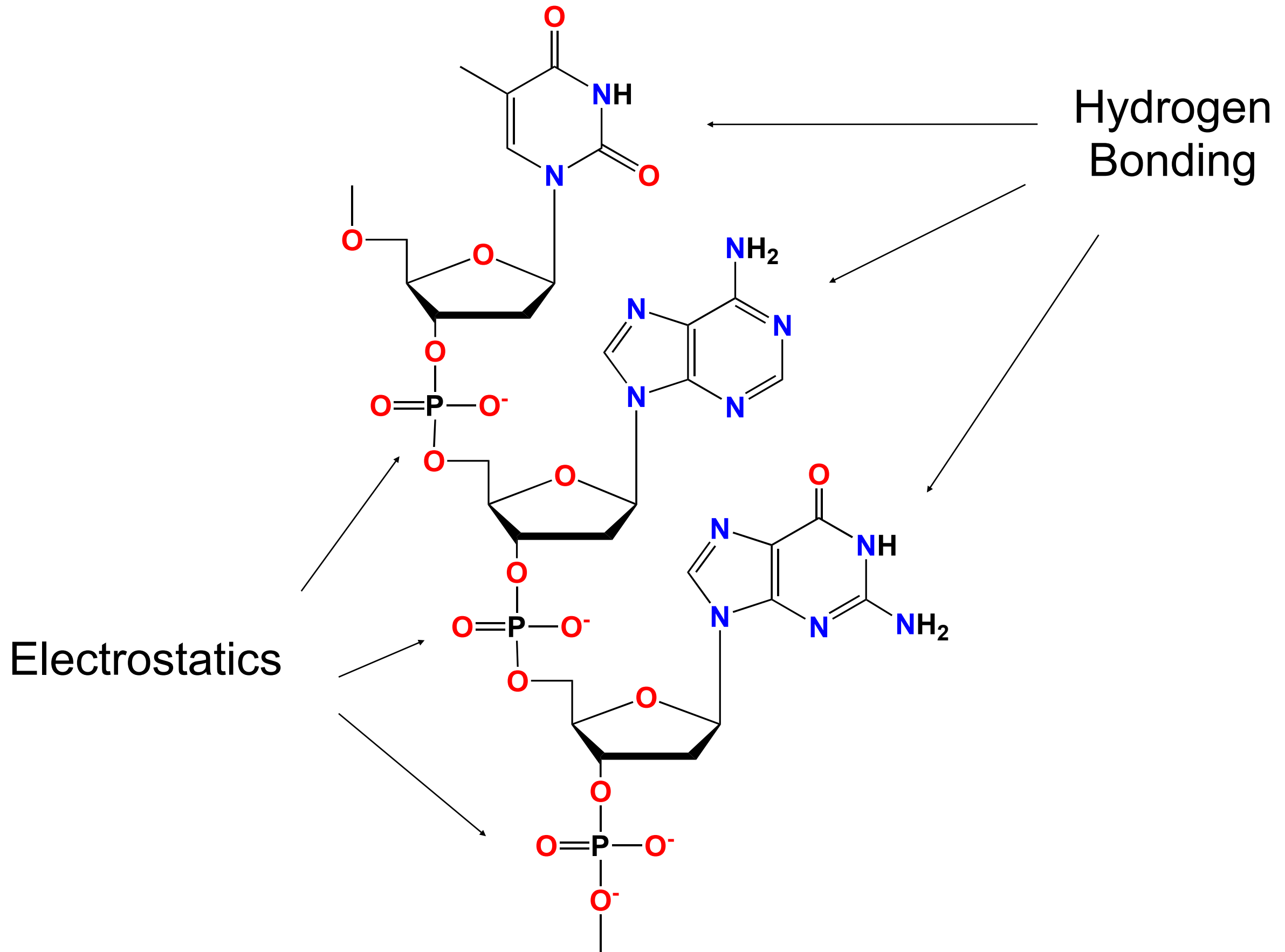
What forces are important?



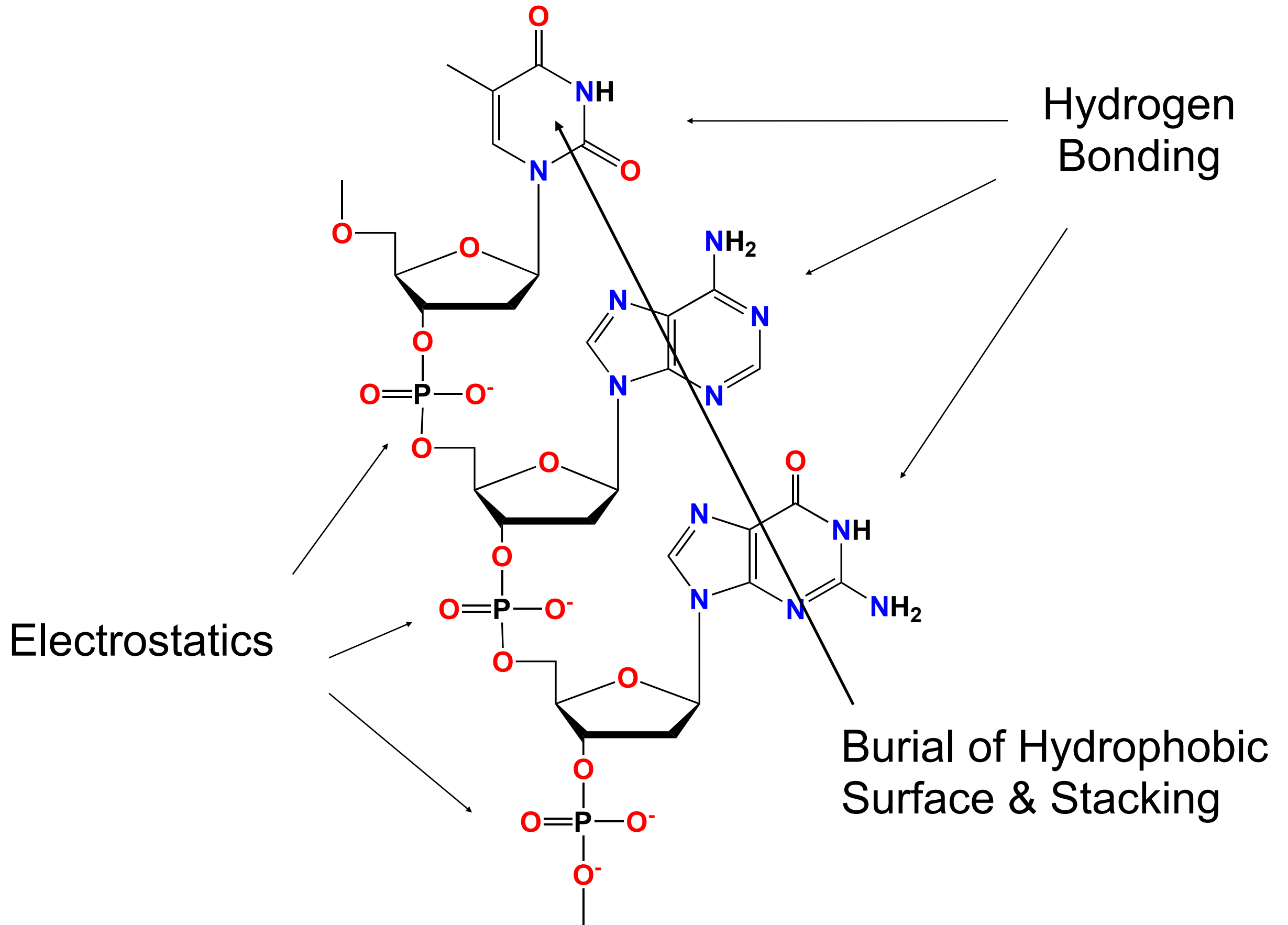
What forces are important?



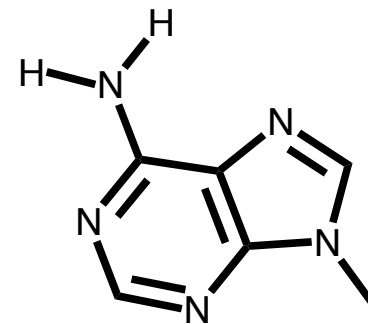
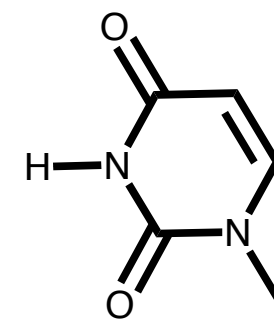
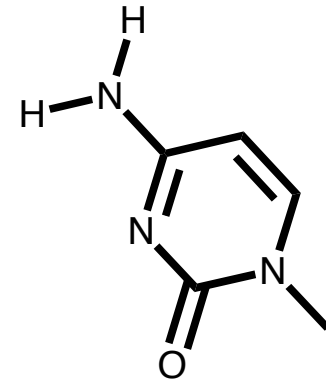
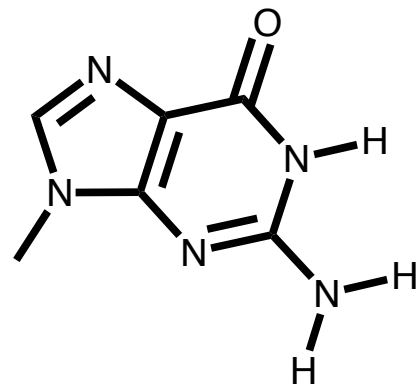
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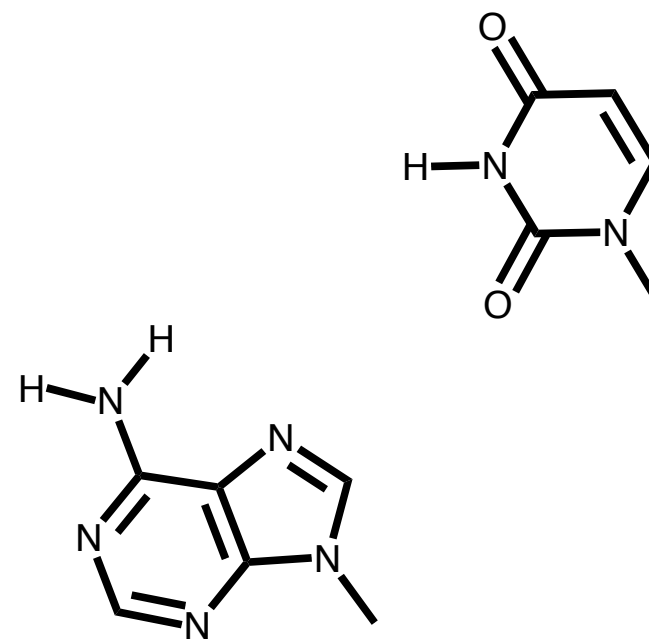
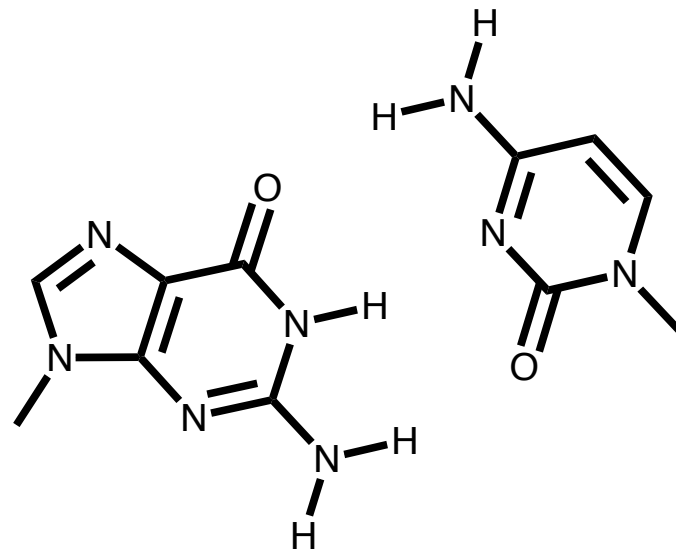
What forces are important?



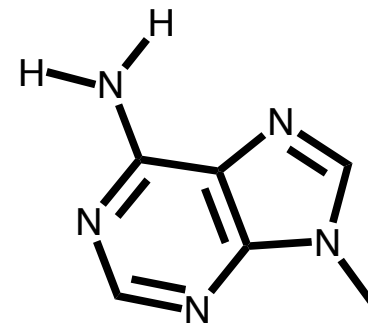
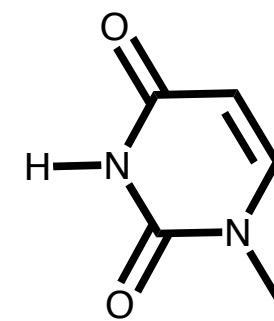
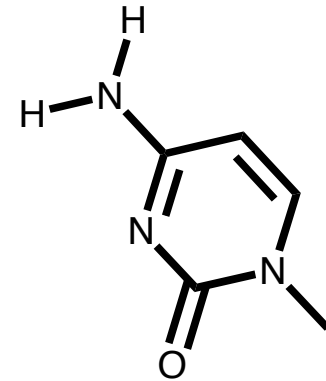
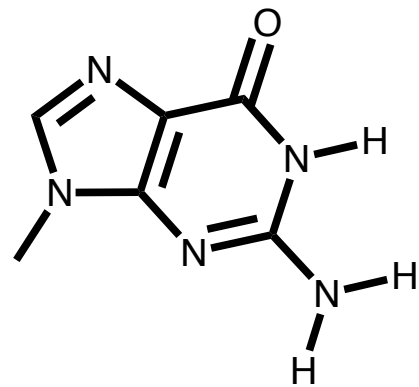
Base Pairs



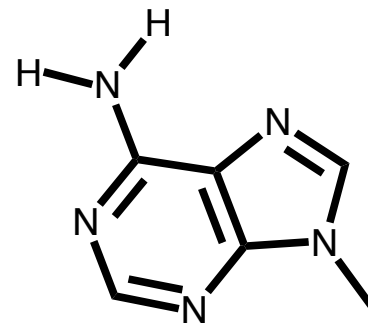
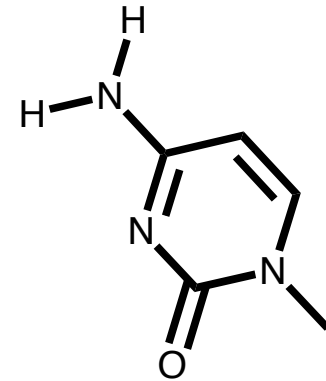
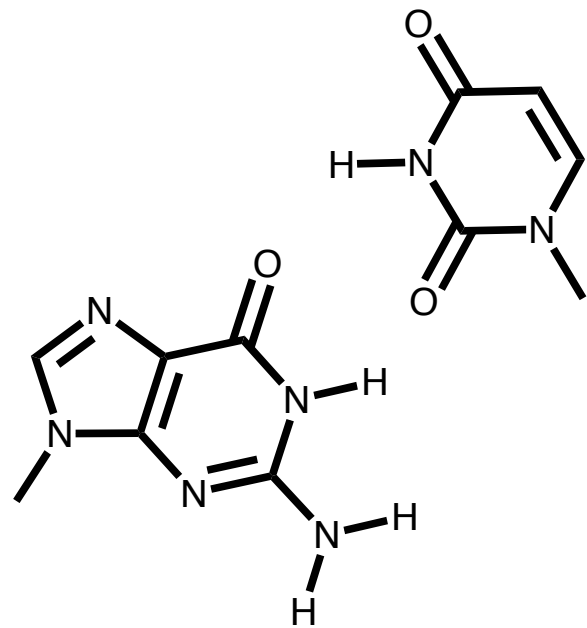
Base Pairs



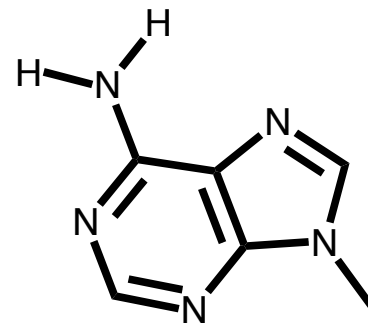
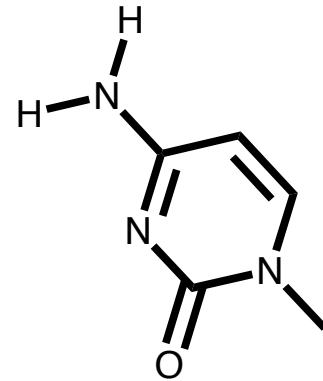
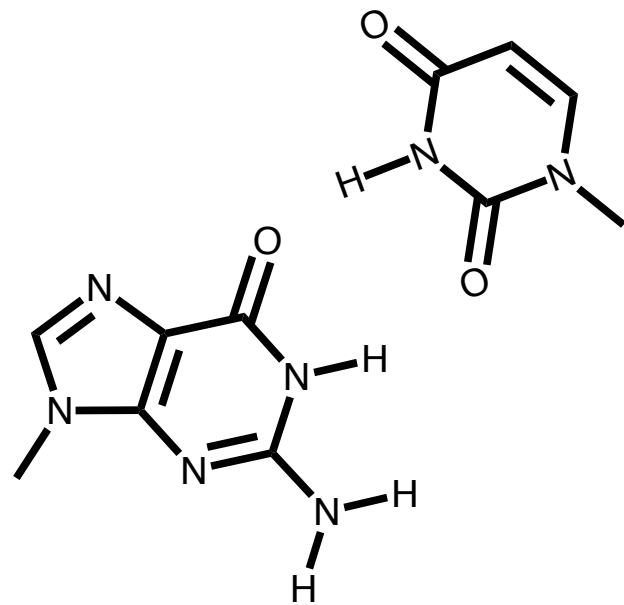
Base Pairs



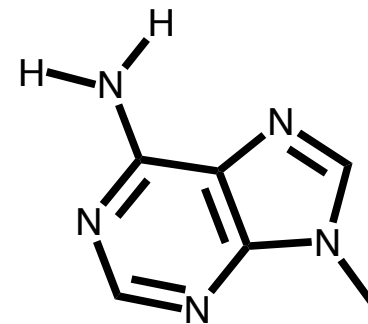
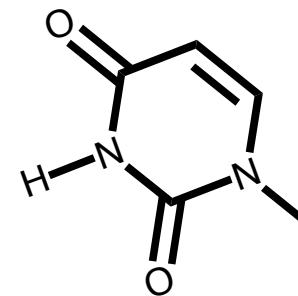
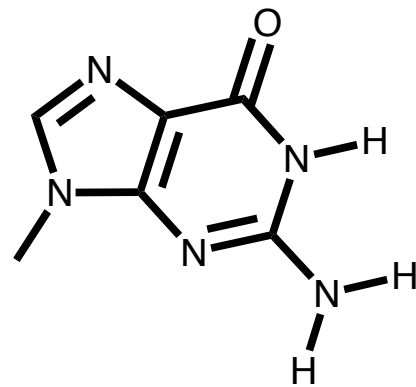
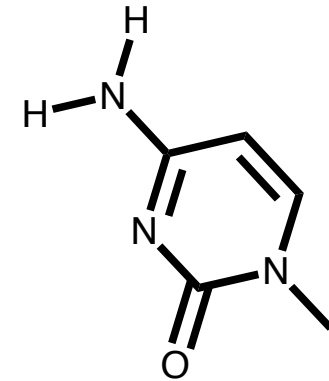
Base Pairs



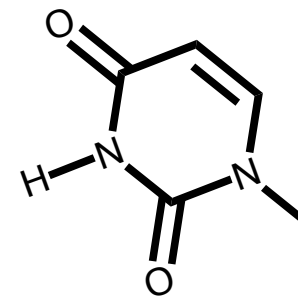
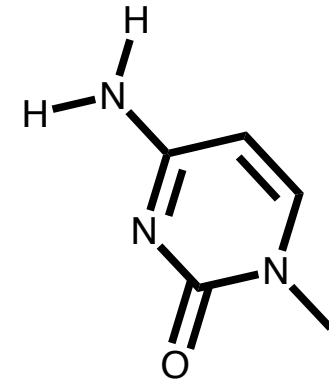
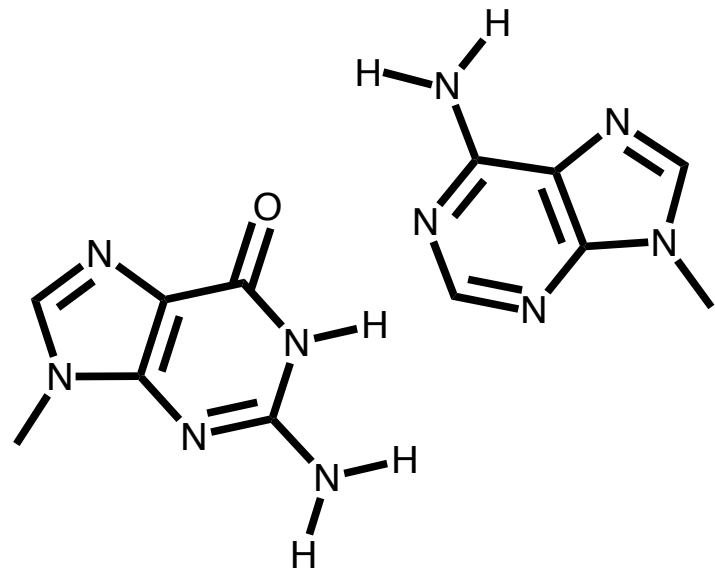
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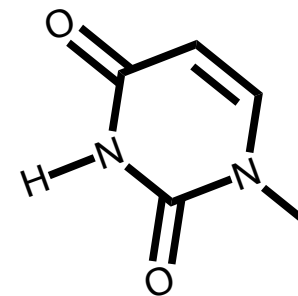
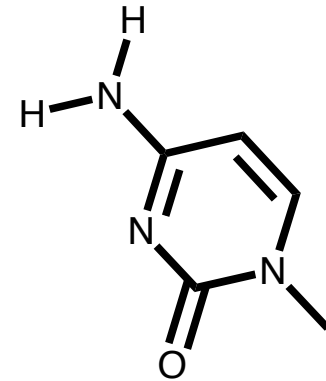
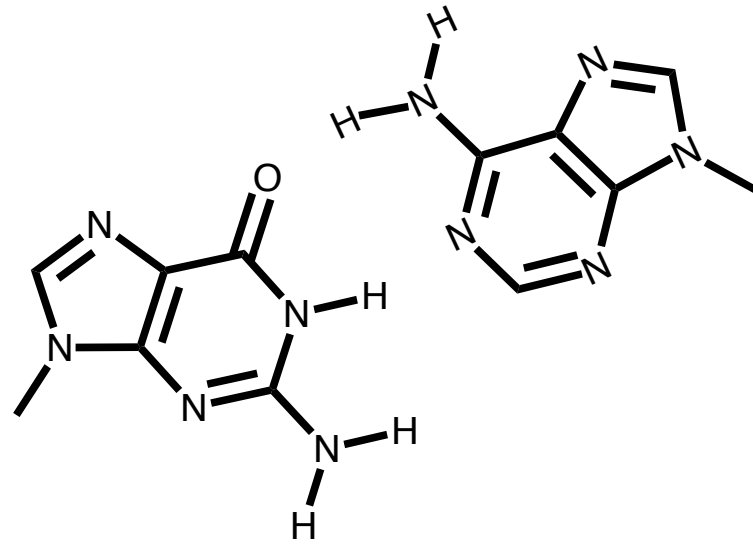
Base Pairs



Base Pairs



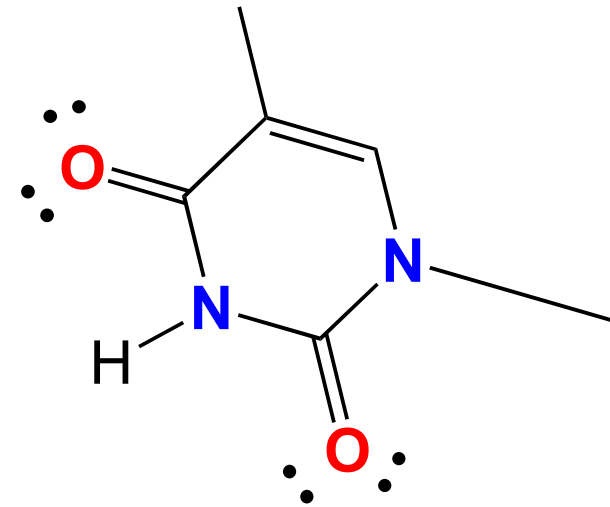
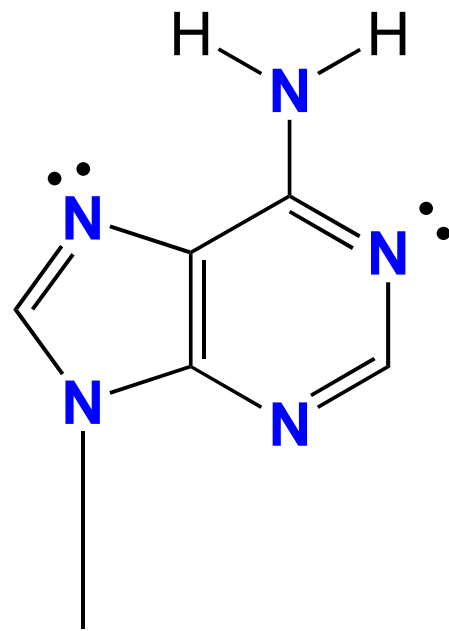
Base Pairs



Base Pairing

(Donors matched to Acceptors)

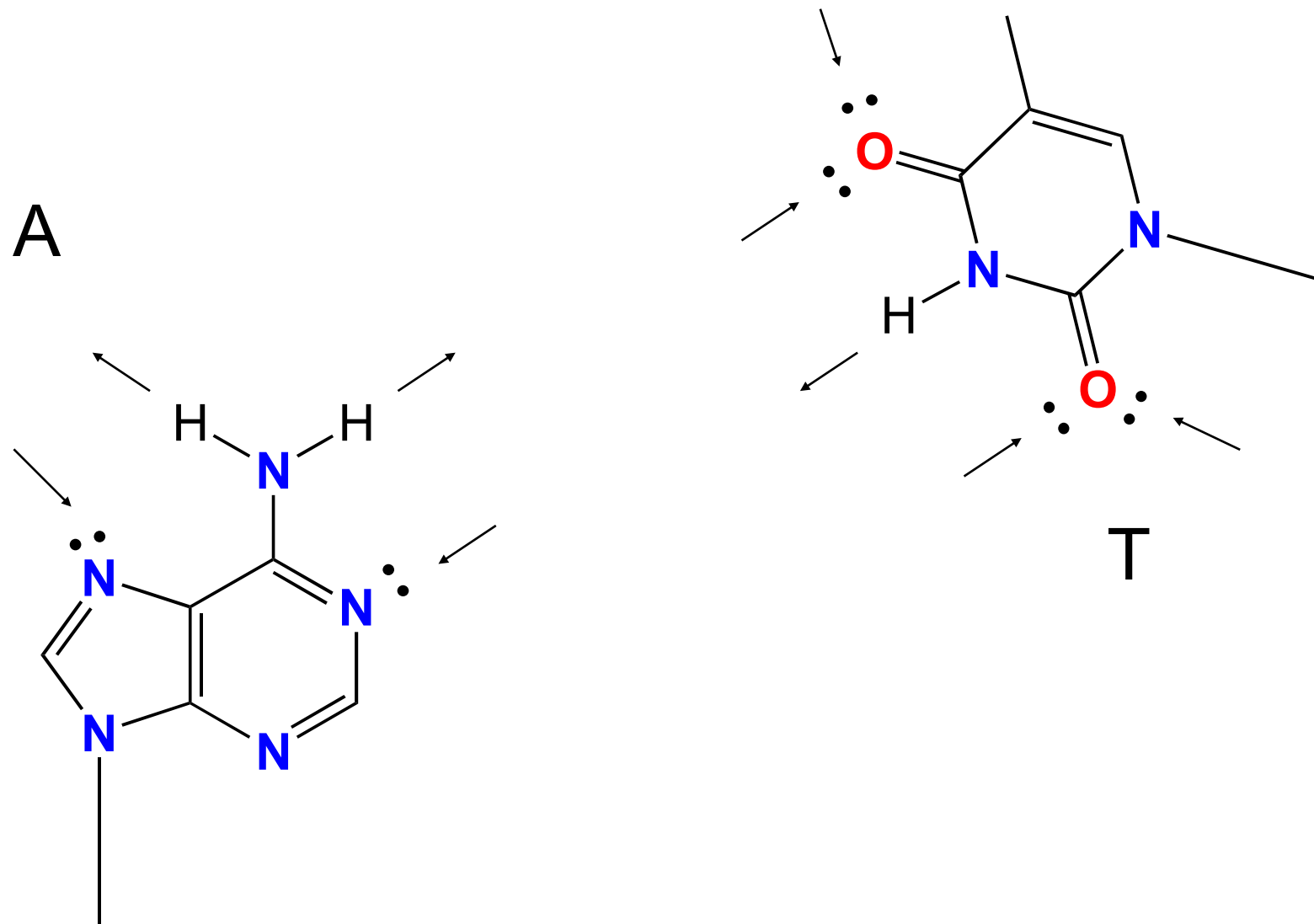
A



T

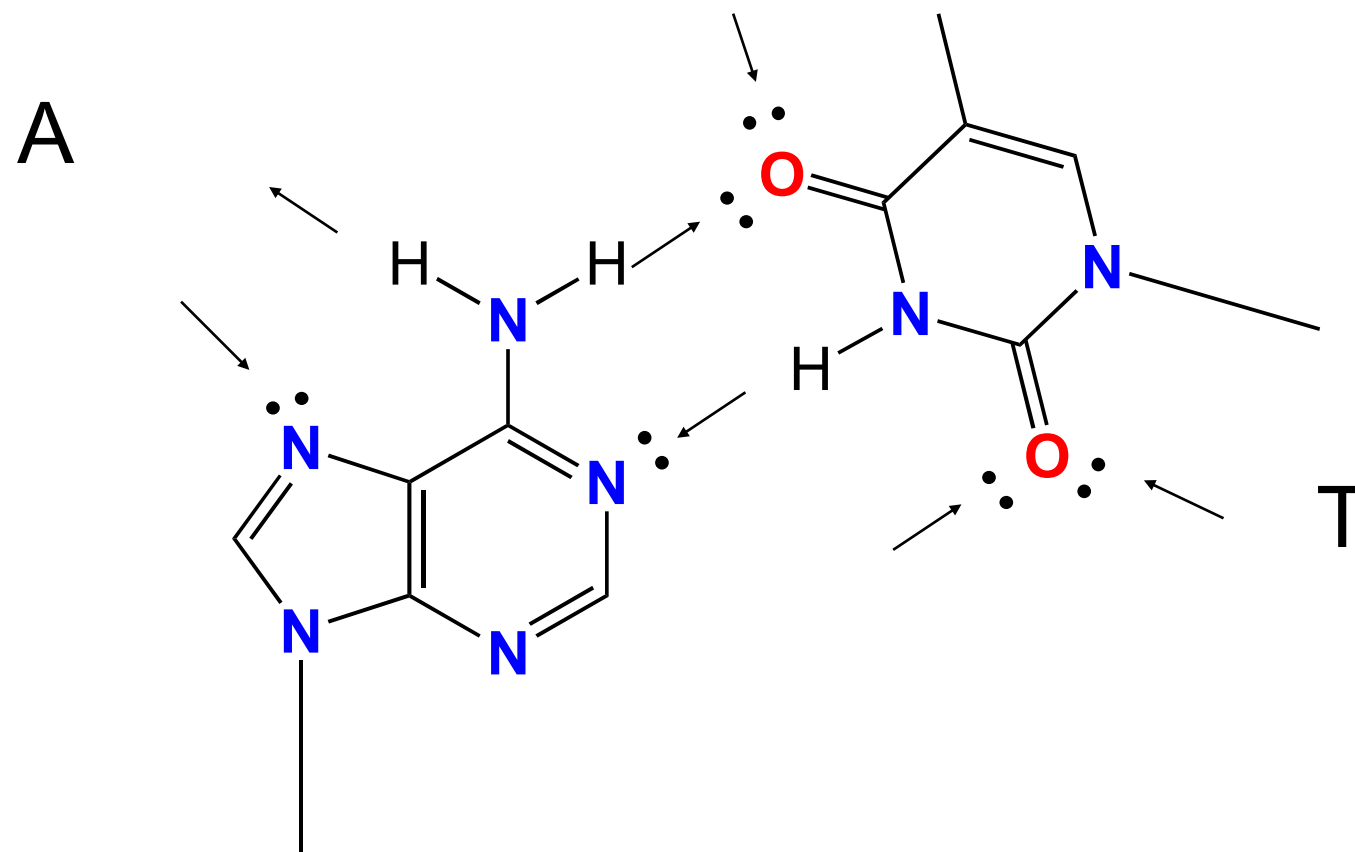
Base Pairing

(Donors matched to Acceptors)



Base Pairing

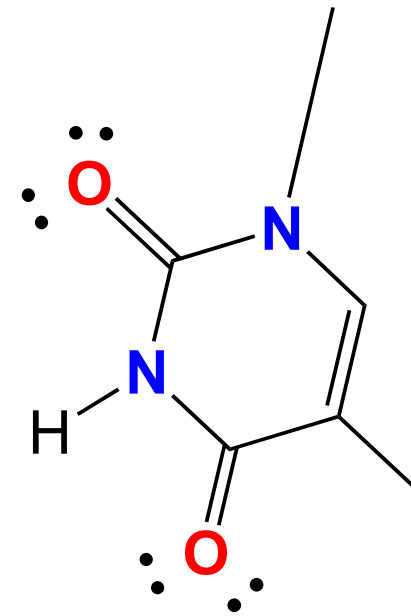
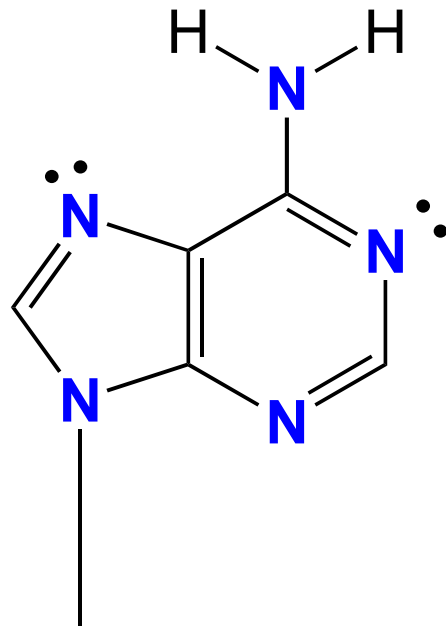
(Donors matched to Acceptors)



Base Pairing

(Donors matched to Acceptors)

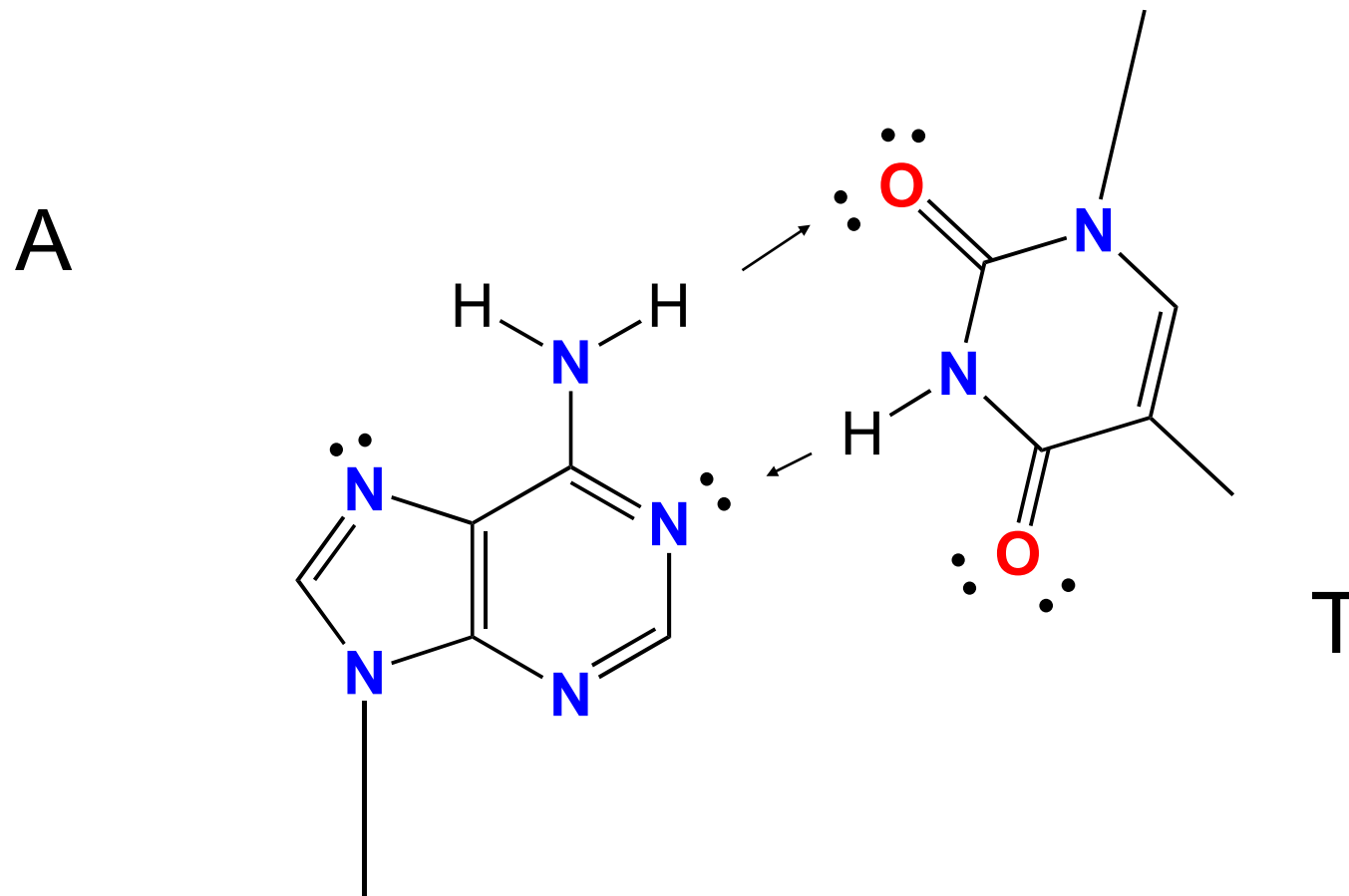
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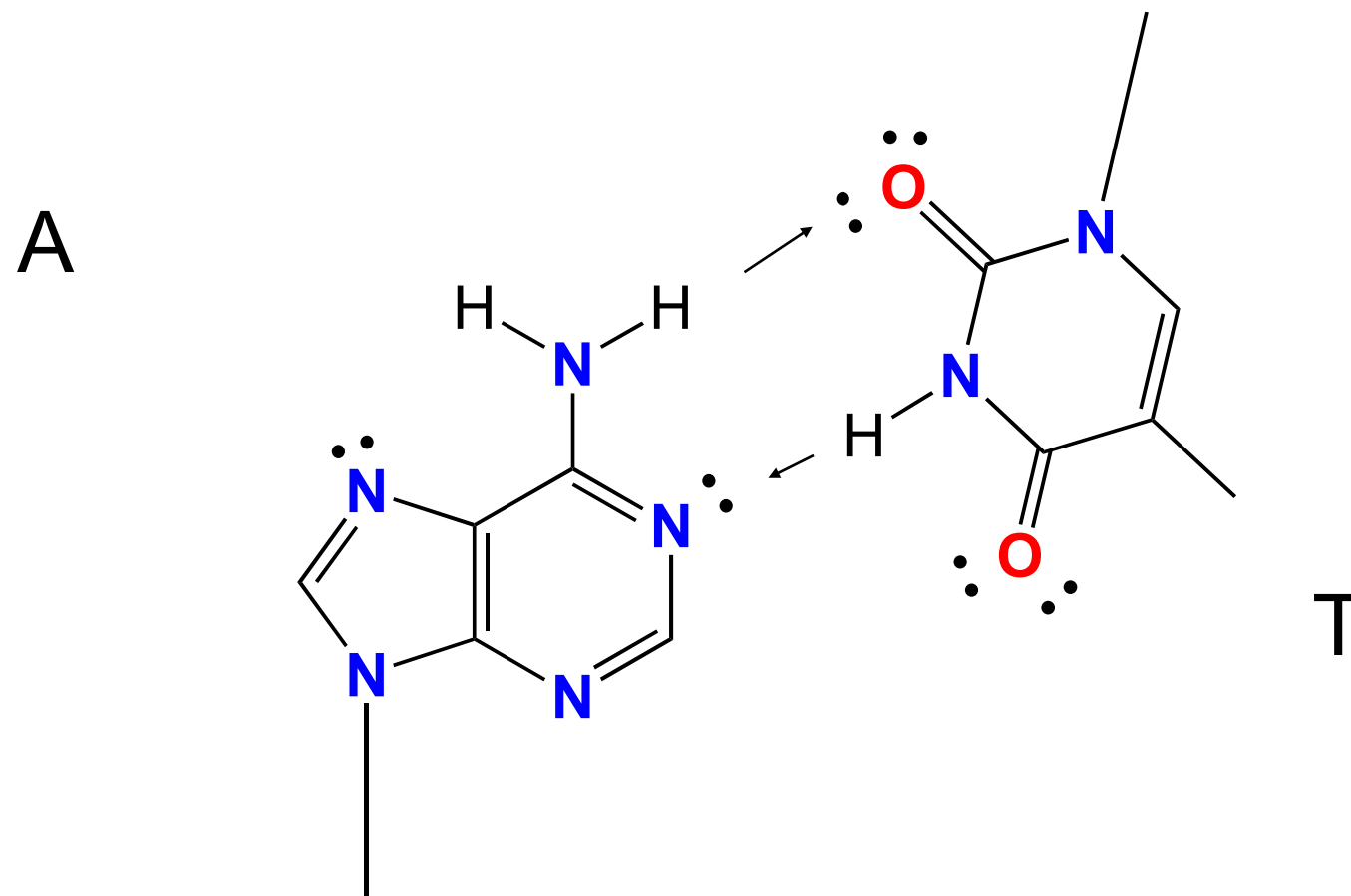
Base Pairing

(Donors matched to Acceptors)



Base Pairing

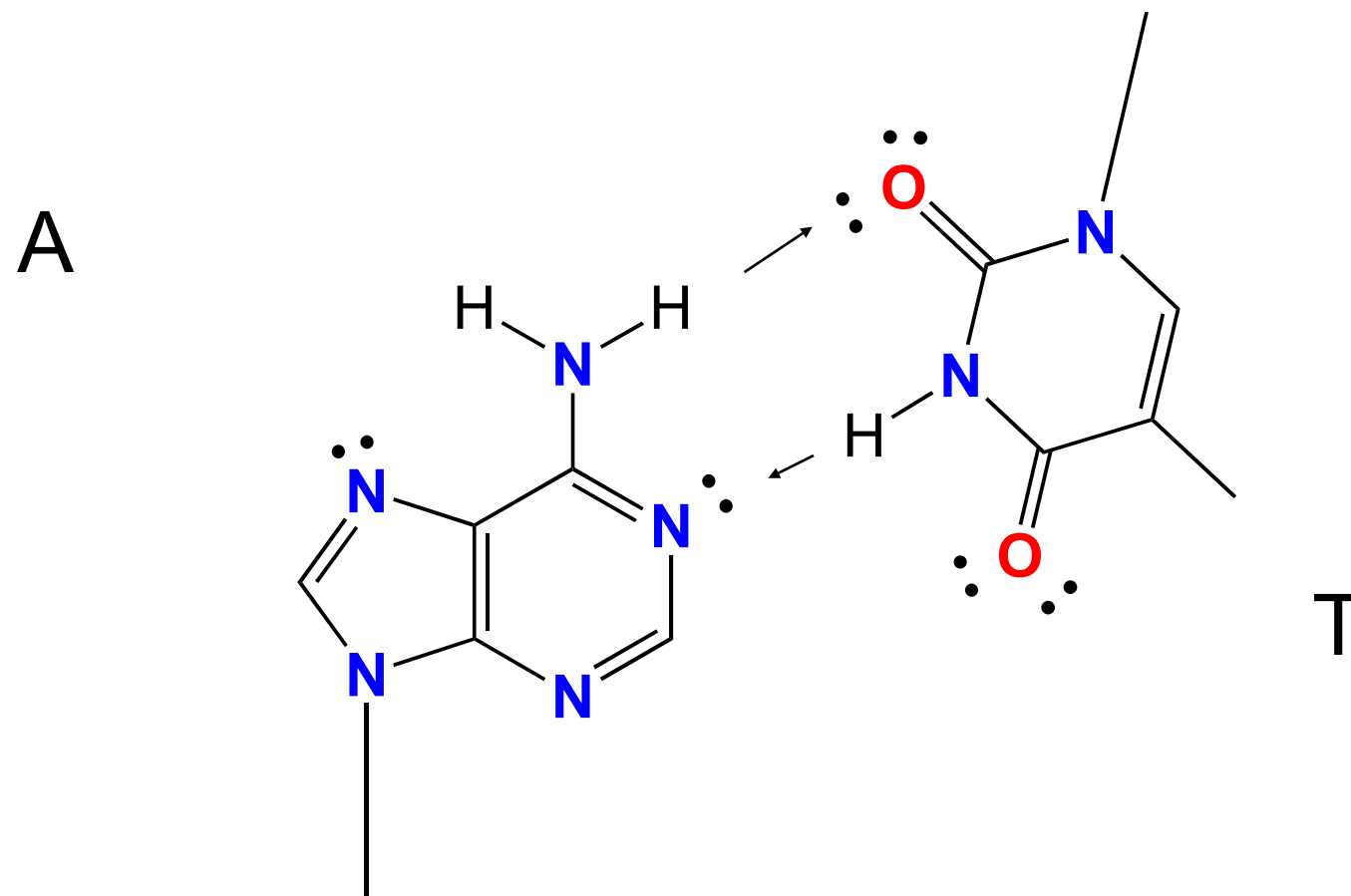
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Good base pairing

Base Pairing

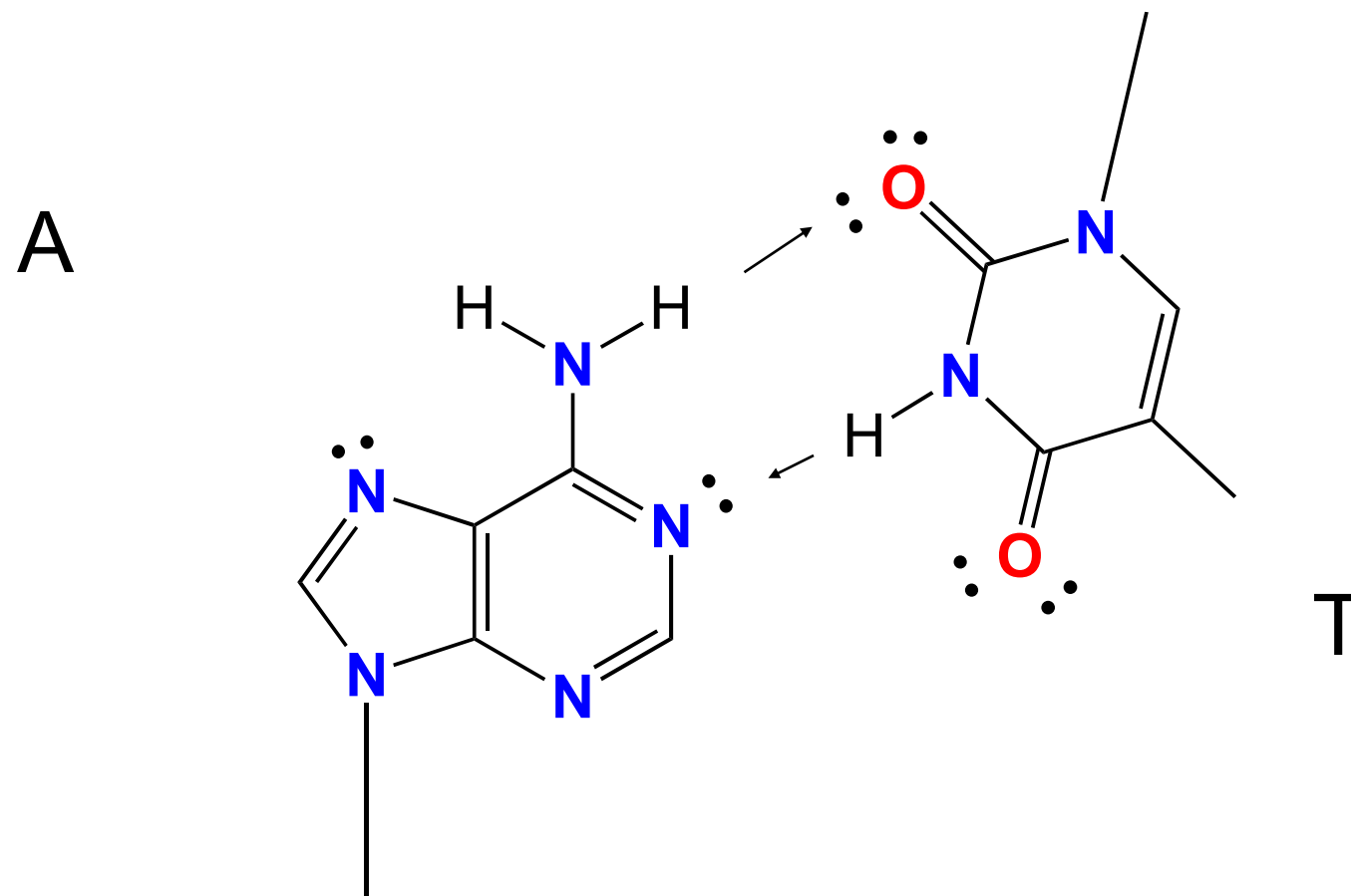
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Good base pairing
Watson-Crick facing

Base Pairing

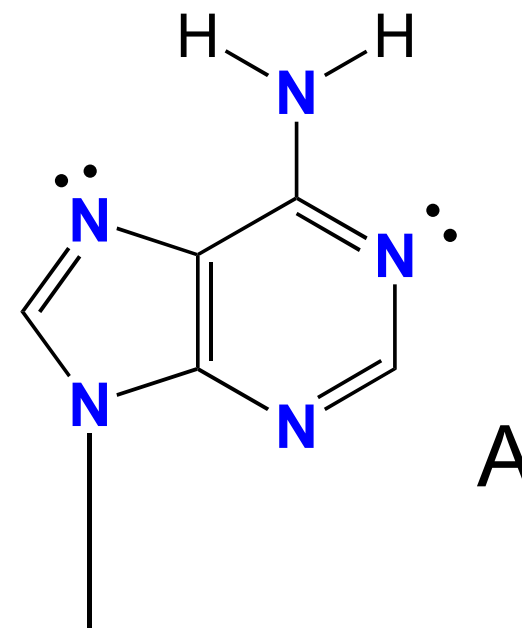
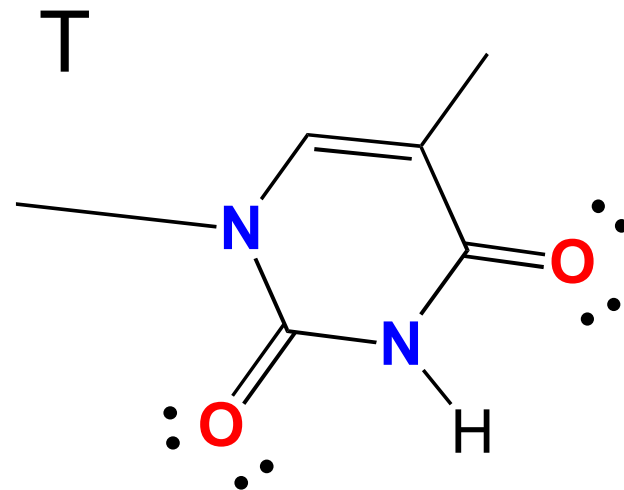
(Donors matched to Acceptors)



Good base pairing
Watson-Crick facing
but *Anti-Watson-Crick* orientation

Base Pairing

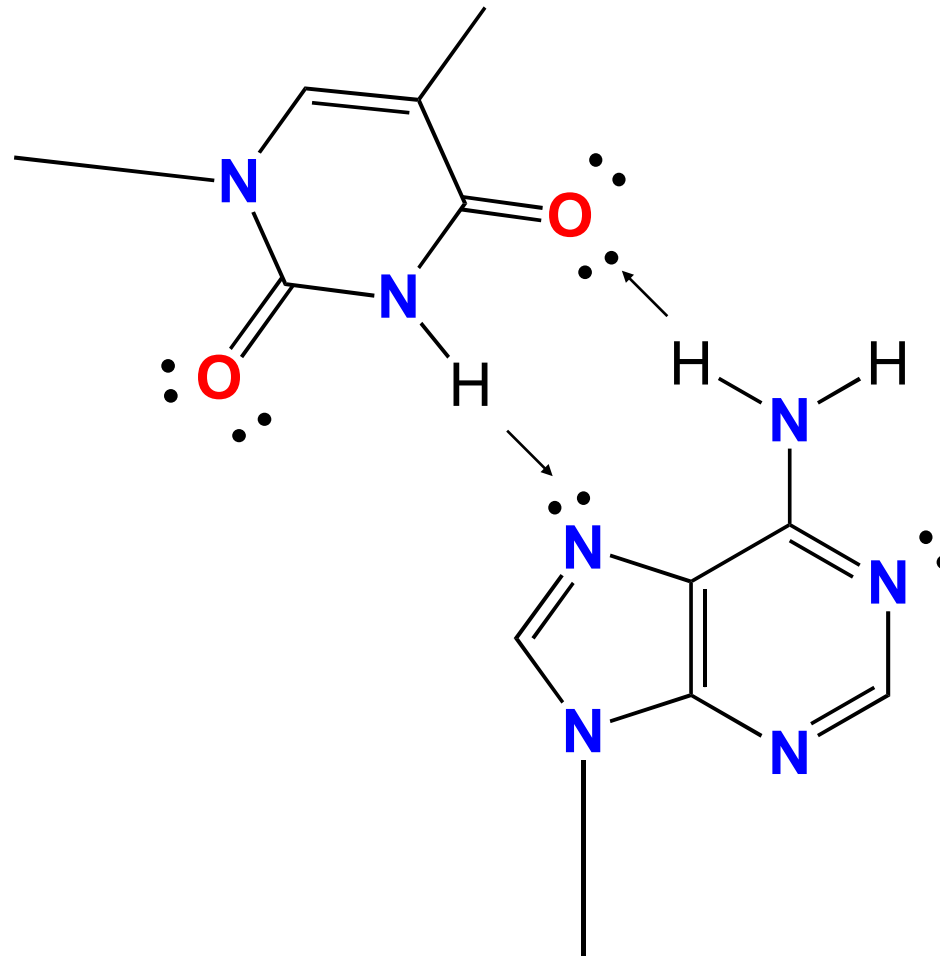
(Donors matched to Acceptors)



Base Pairing

(Donors matched to Acceptors)

T

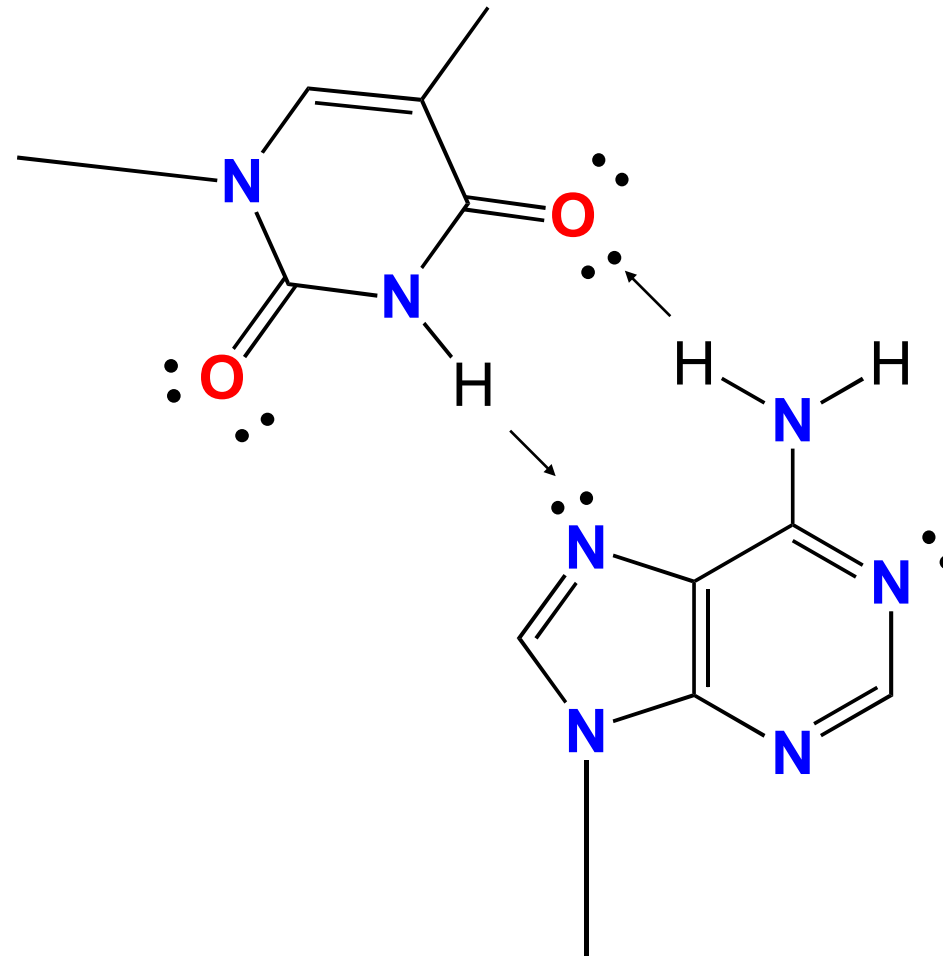


A

Base Pairing

(Donors matched to Acceptors)

T



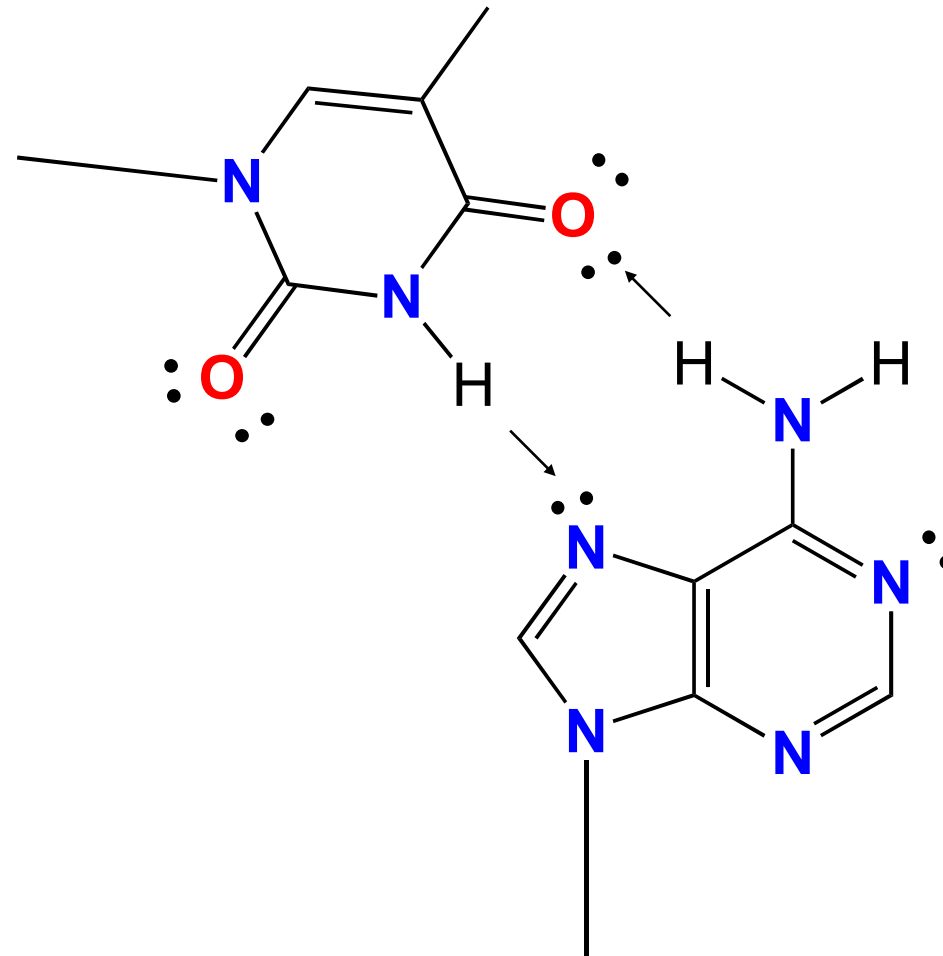
A

Good base pairing

Base Pairing

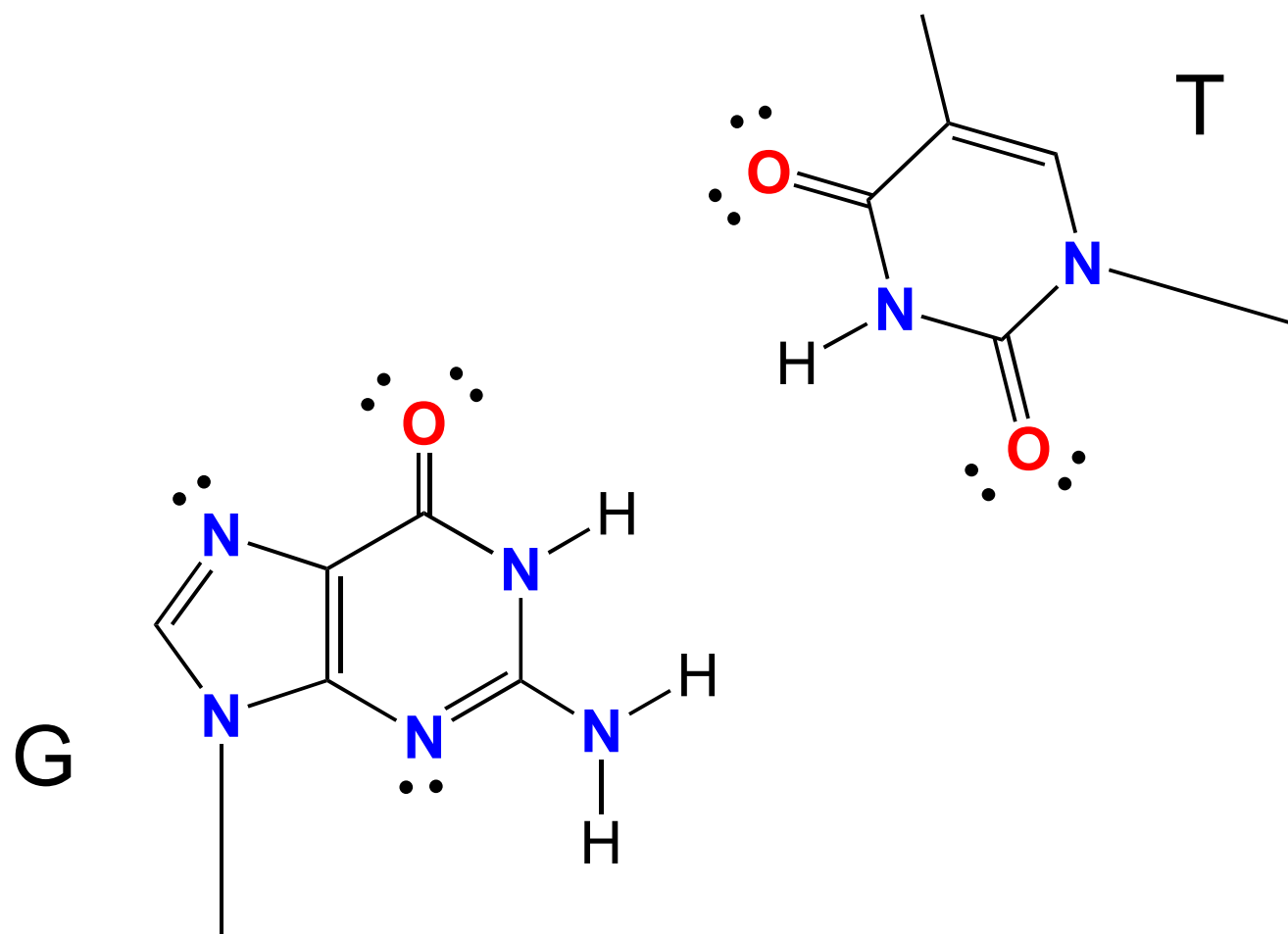
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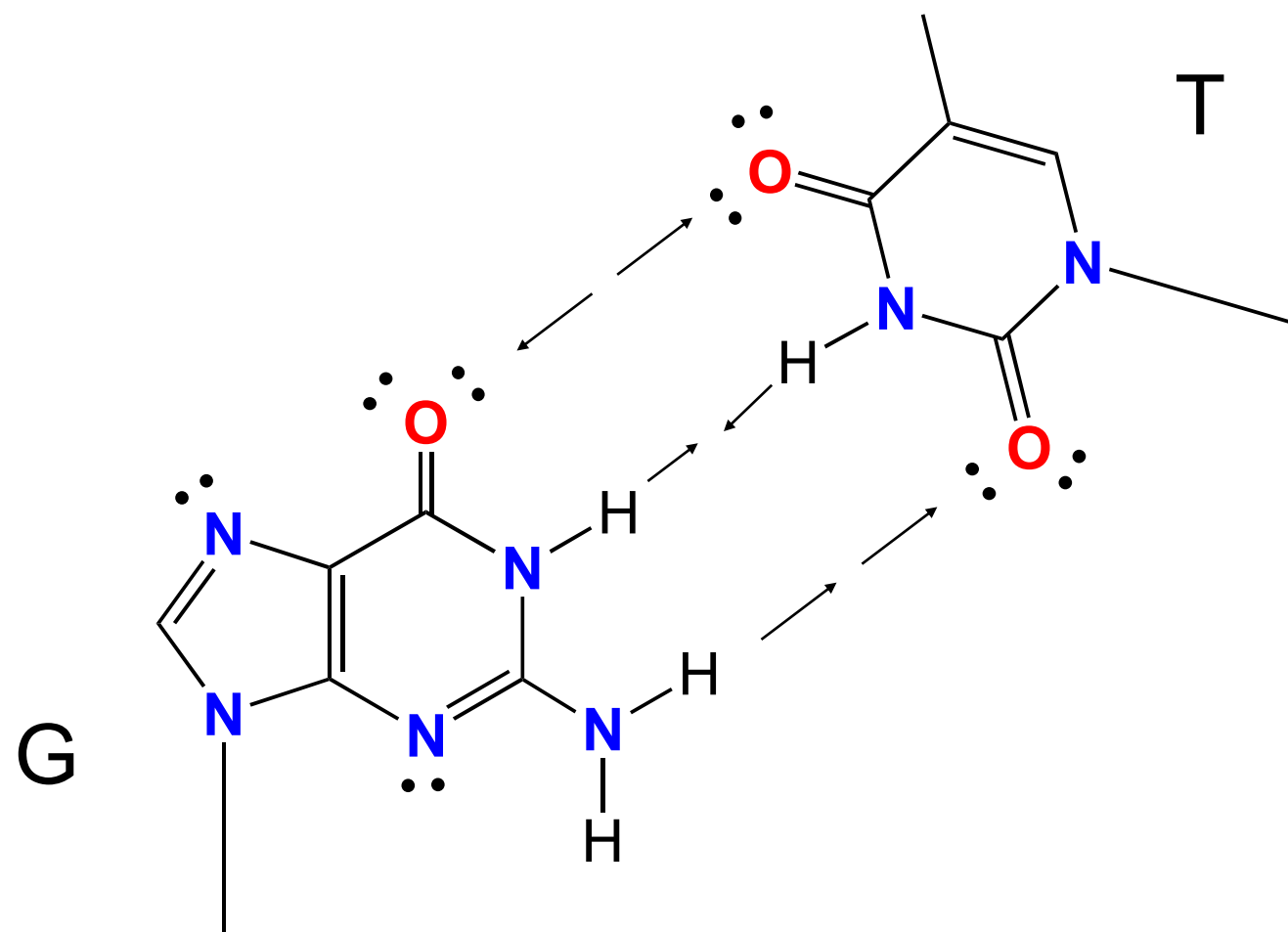
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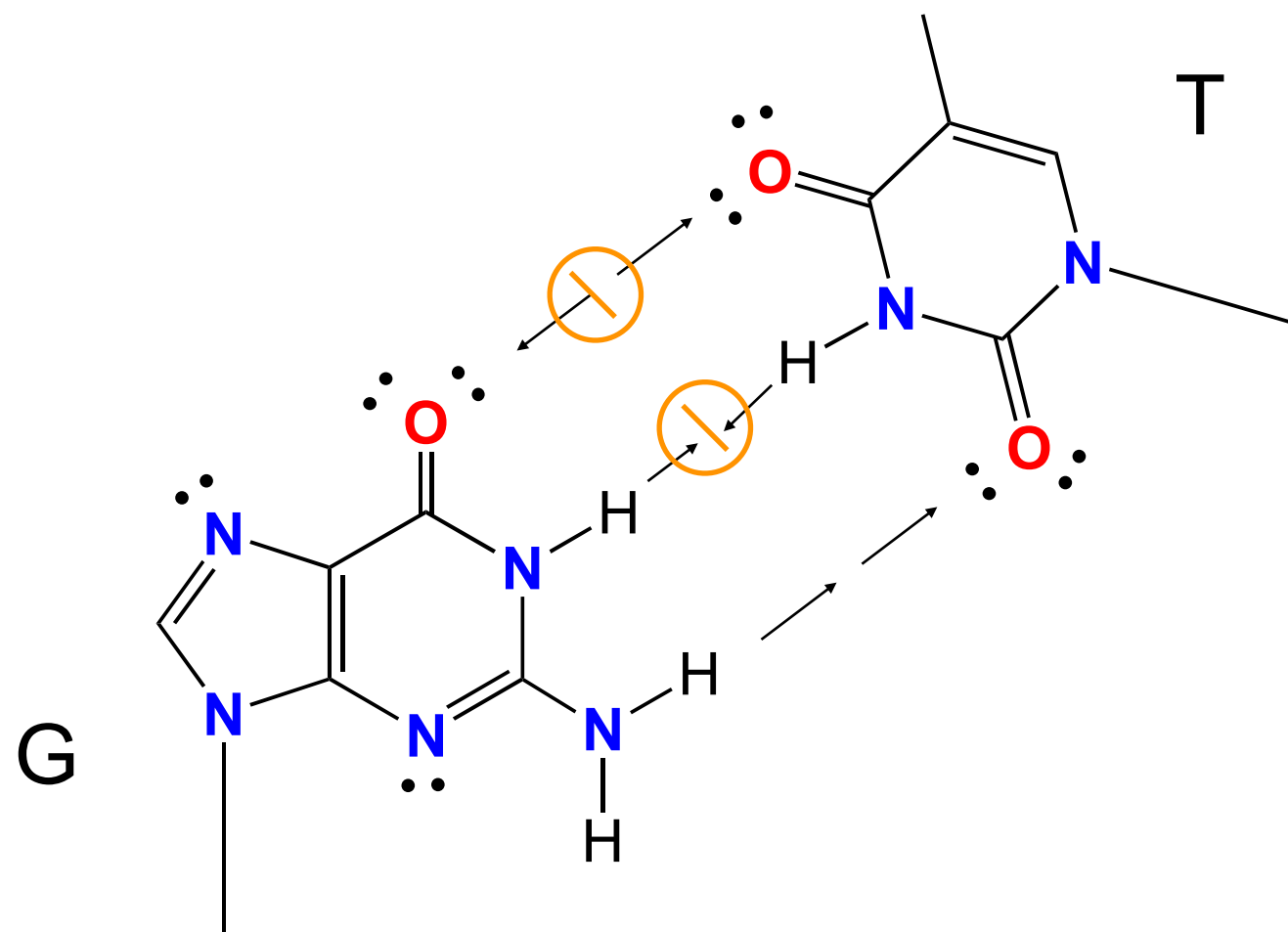


A

Good base pairing
WC-Hoogsteen facing

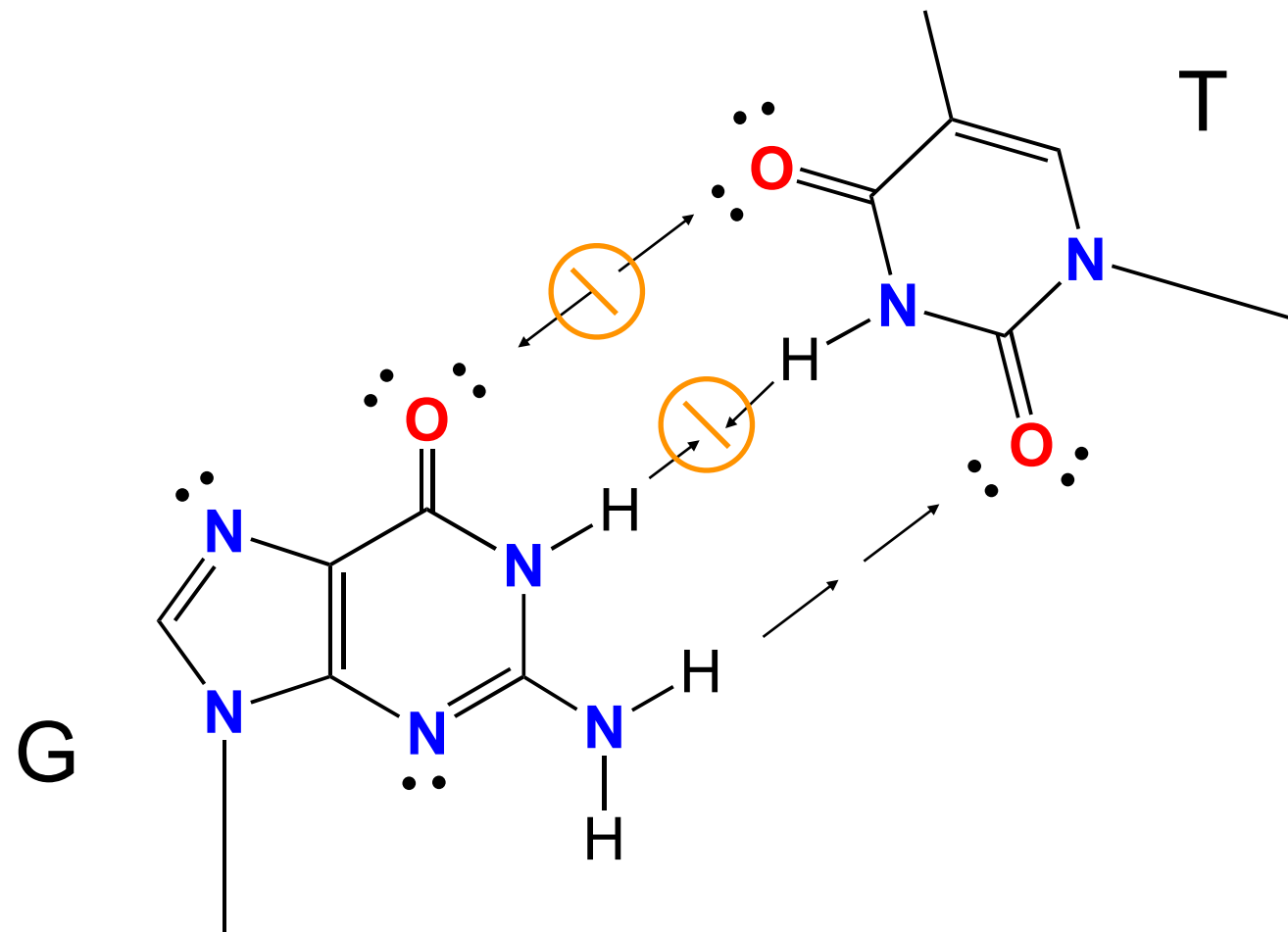


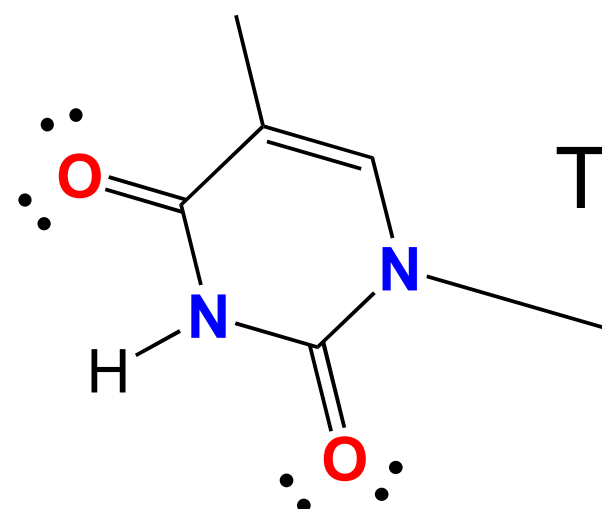
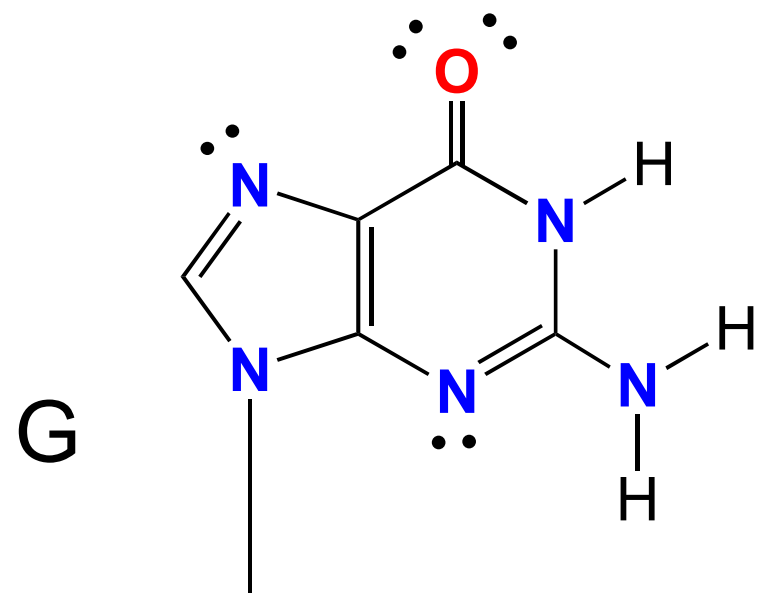


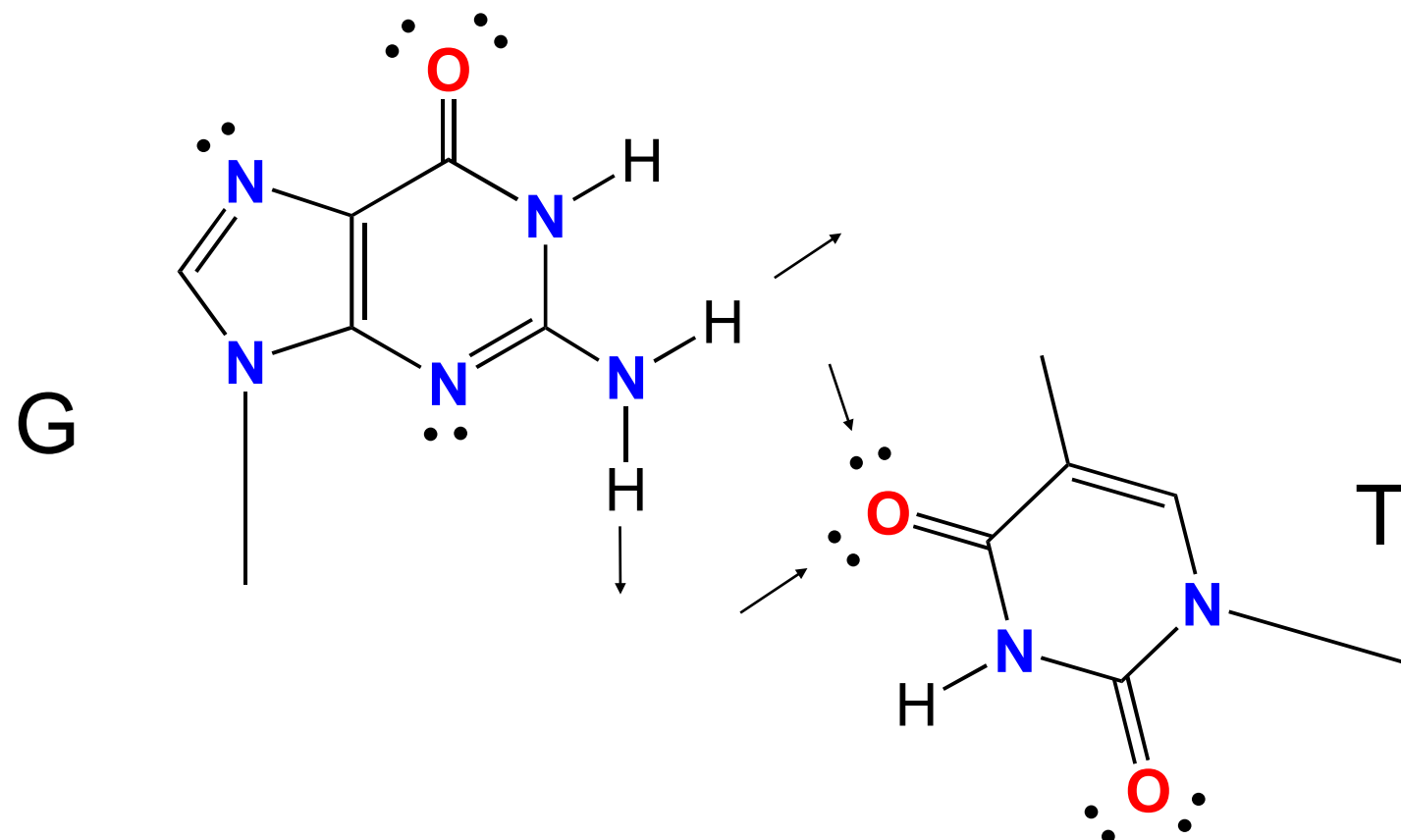


Bad Base Pairing

(Donors ***not*** matched to Acceptors)

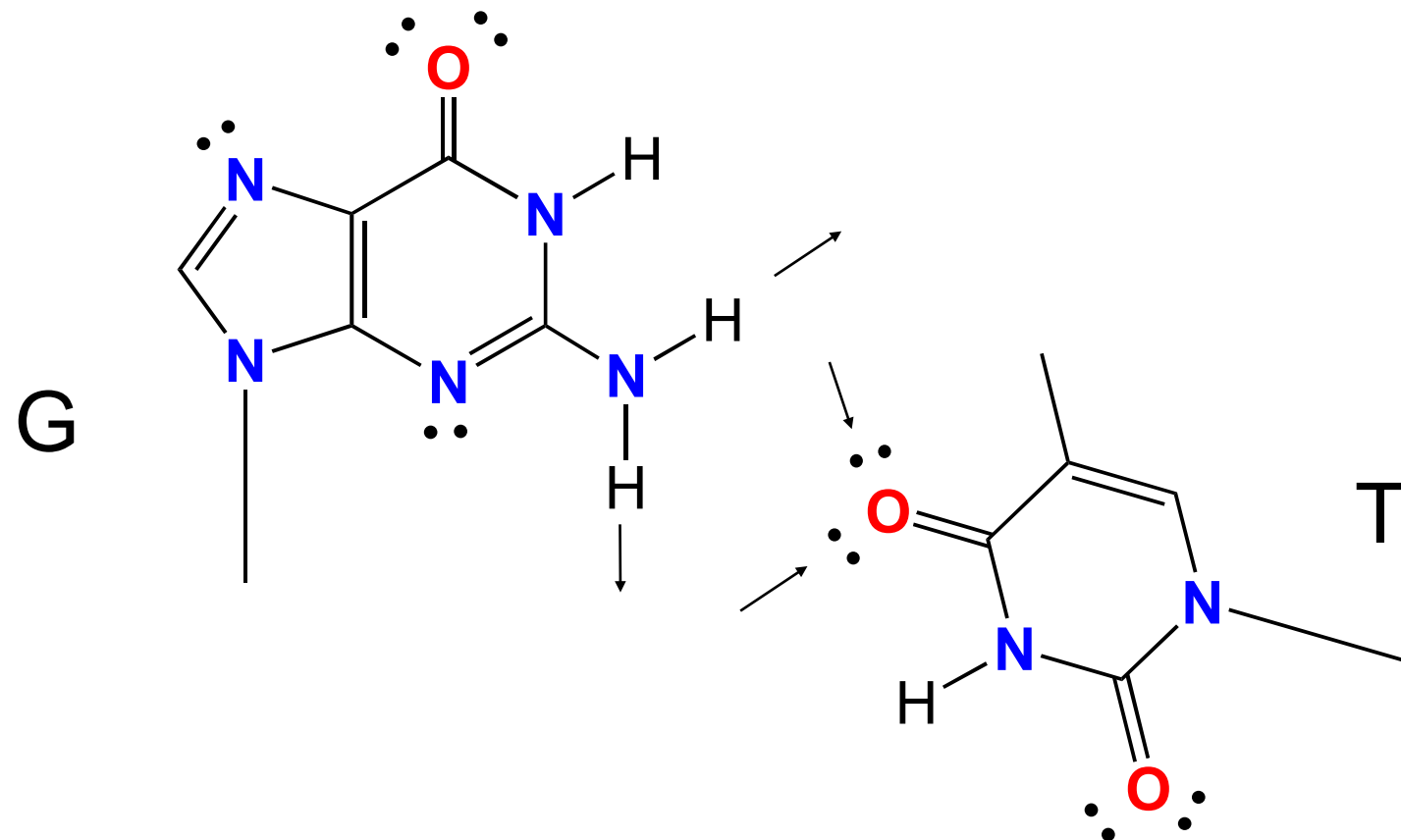




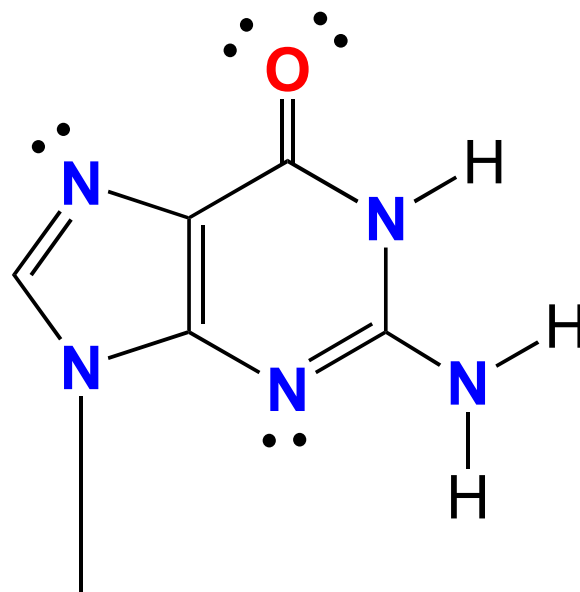


Bad Base Pairing

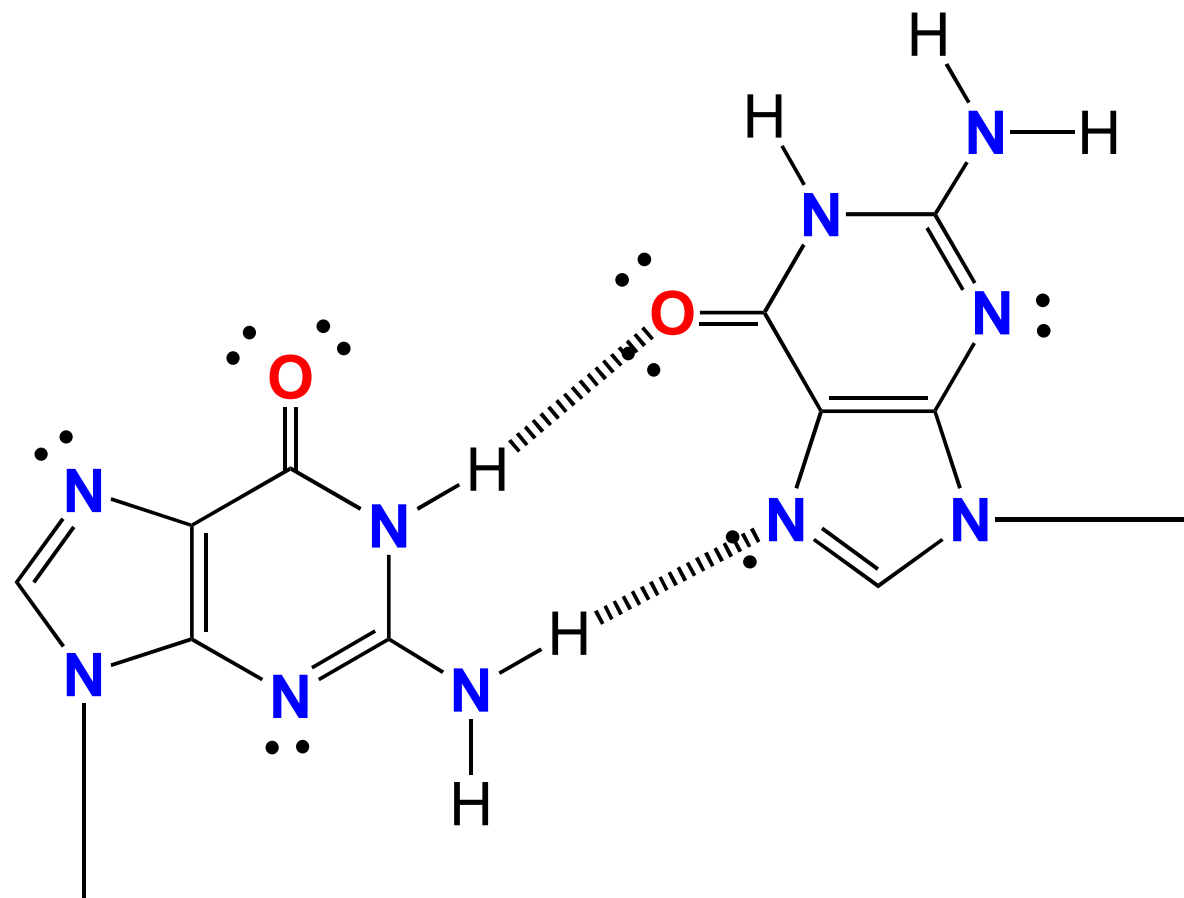
(Donors to Acceptors with *terrible angles*)



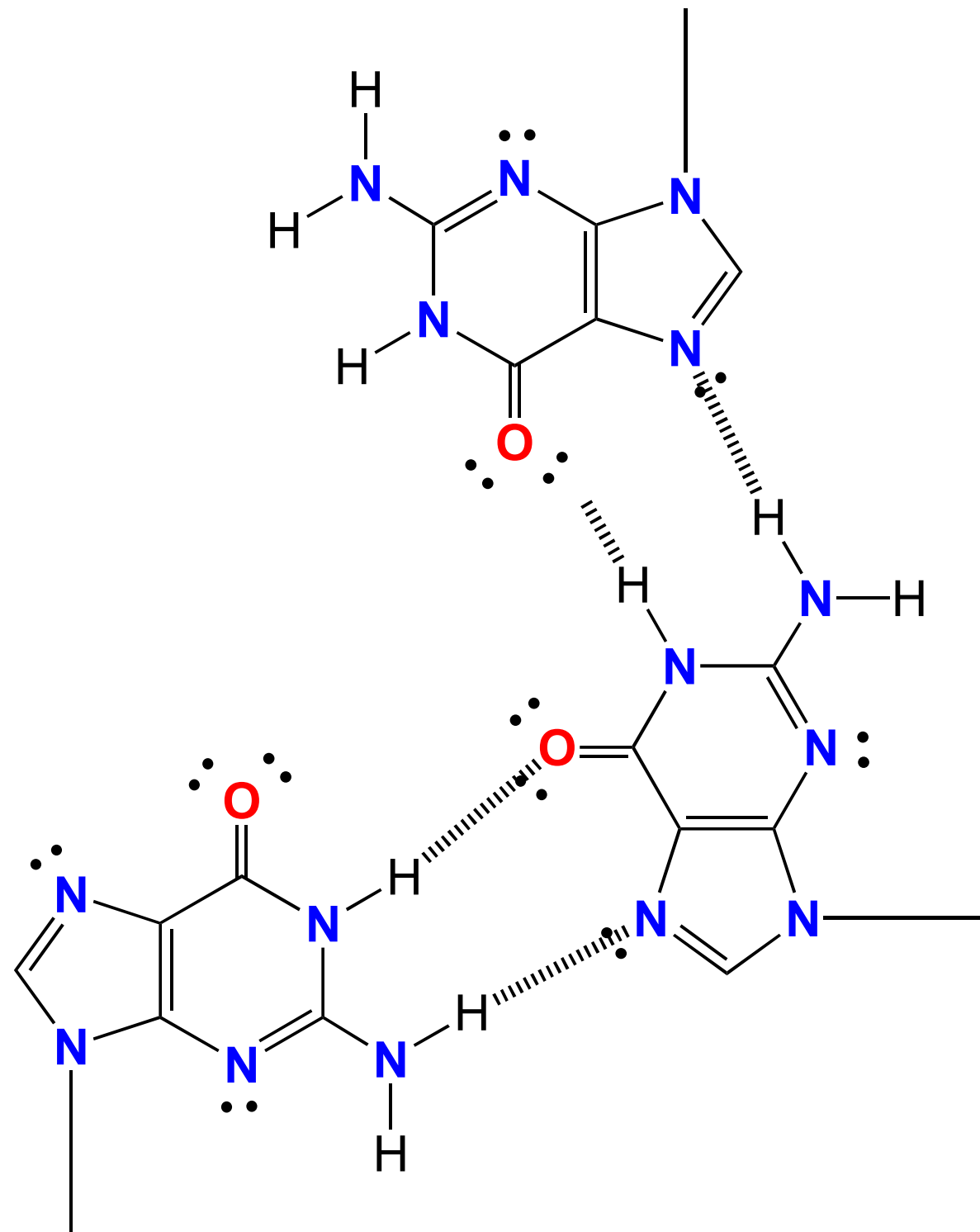
Wild (but good) Base Pairing



Wild (but good) Base Pairing

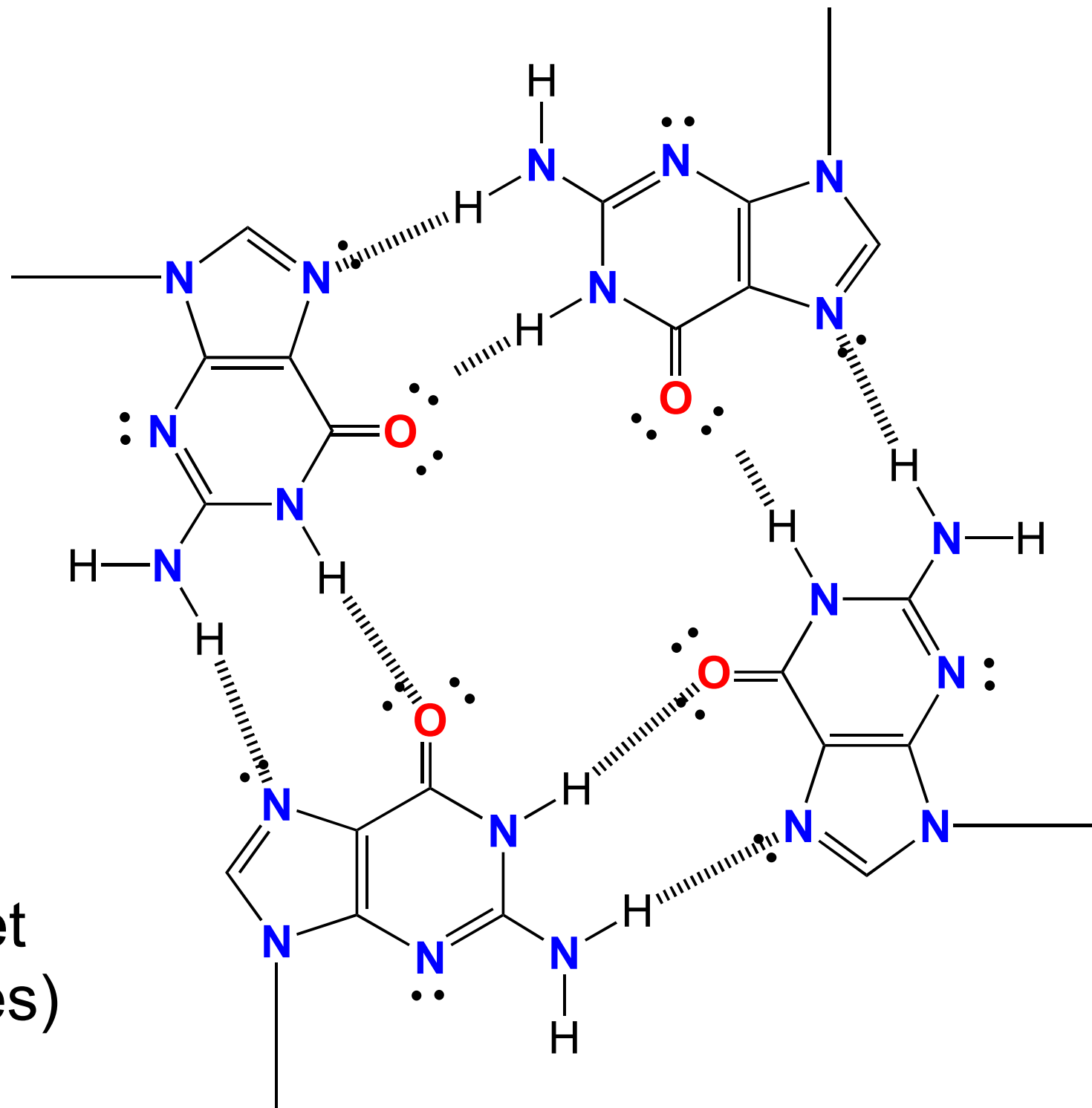


Wild (but good) Base Pairing



Wild (but good) Base Pairing

G-quartet
(Telomeres)

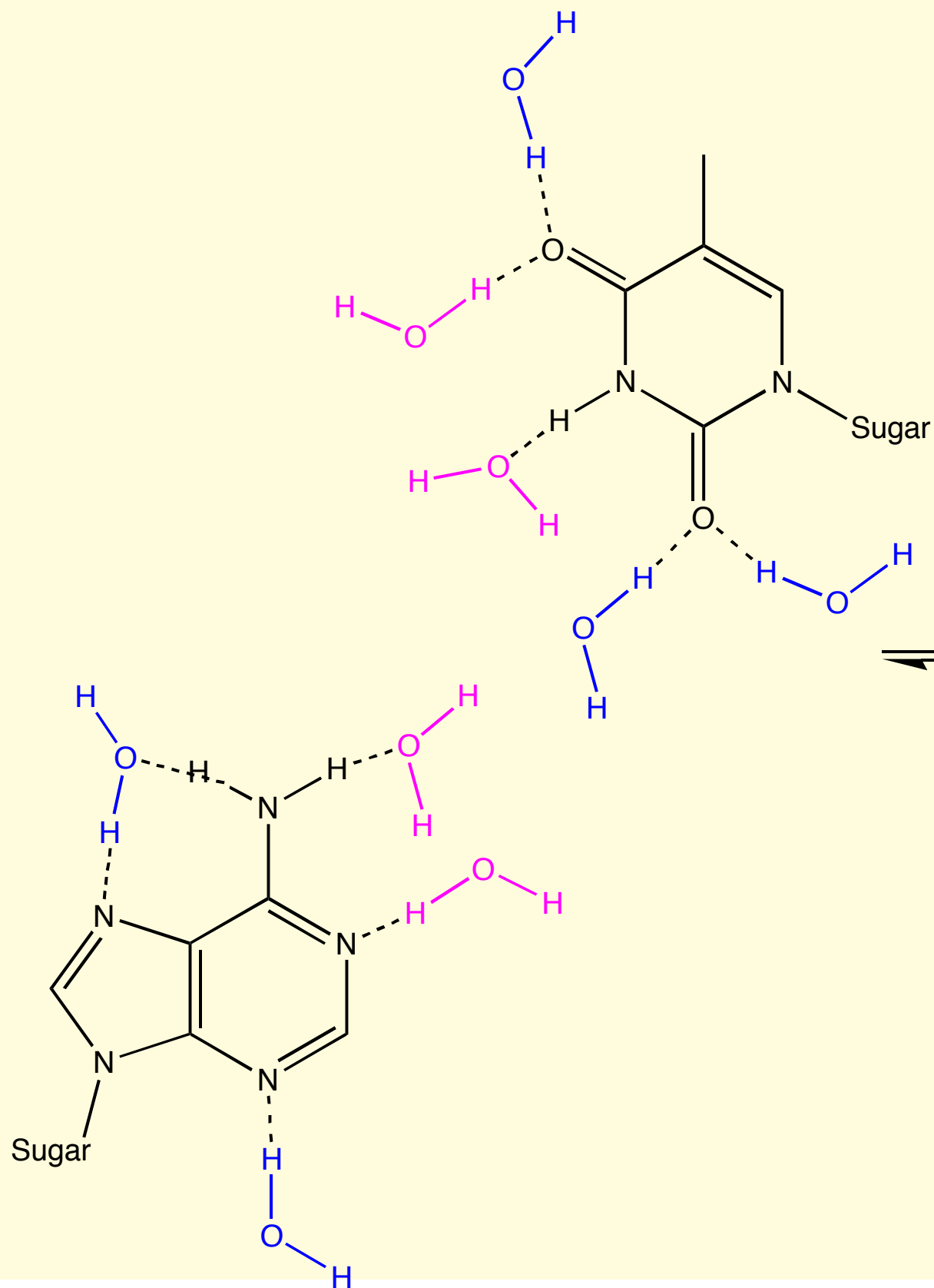


What *drives* double helix formation?

What *drives* double helix formation?

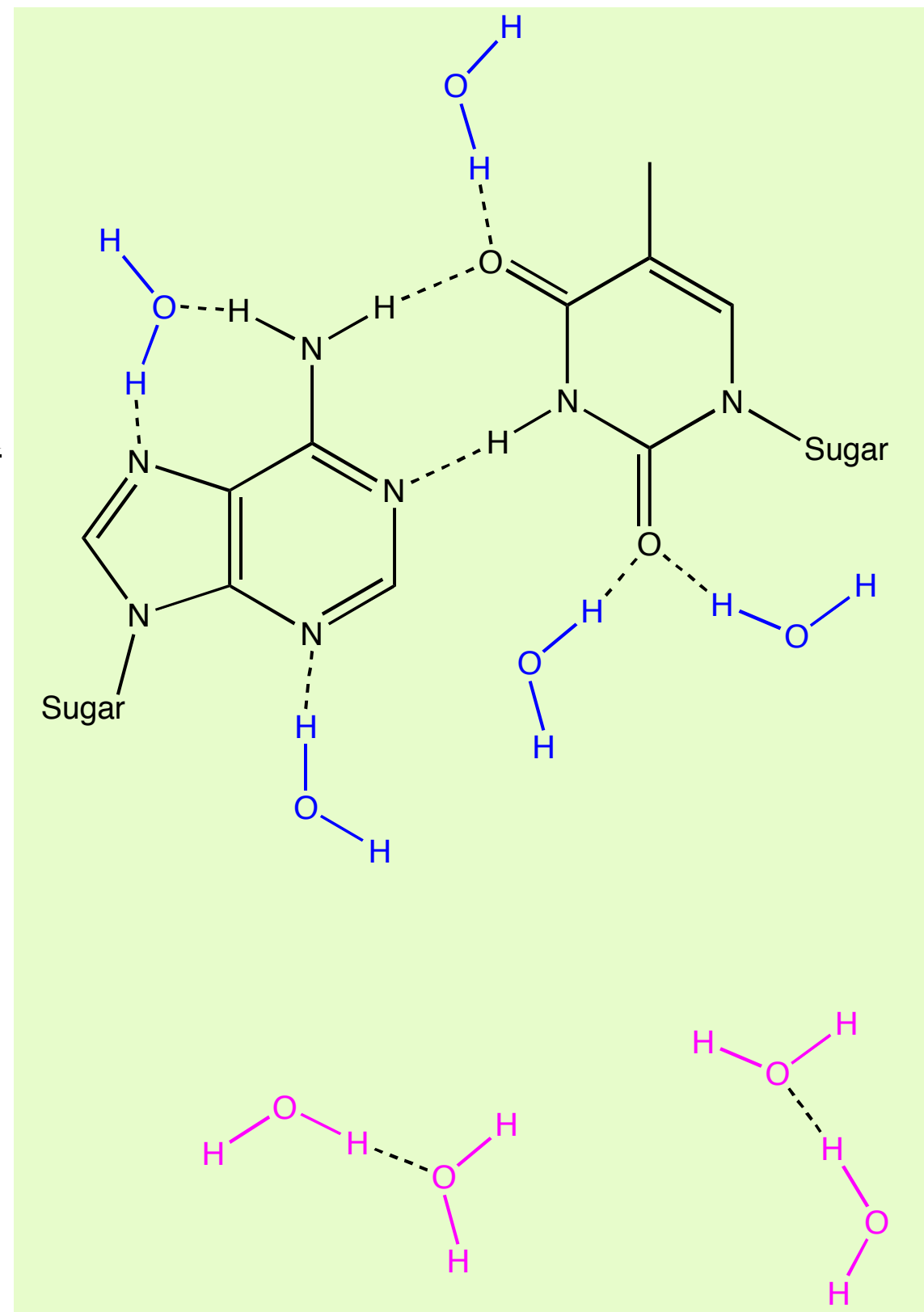
NOT hydrogen bonding

AT Base Pair

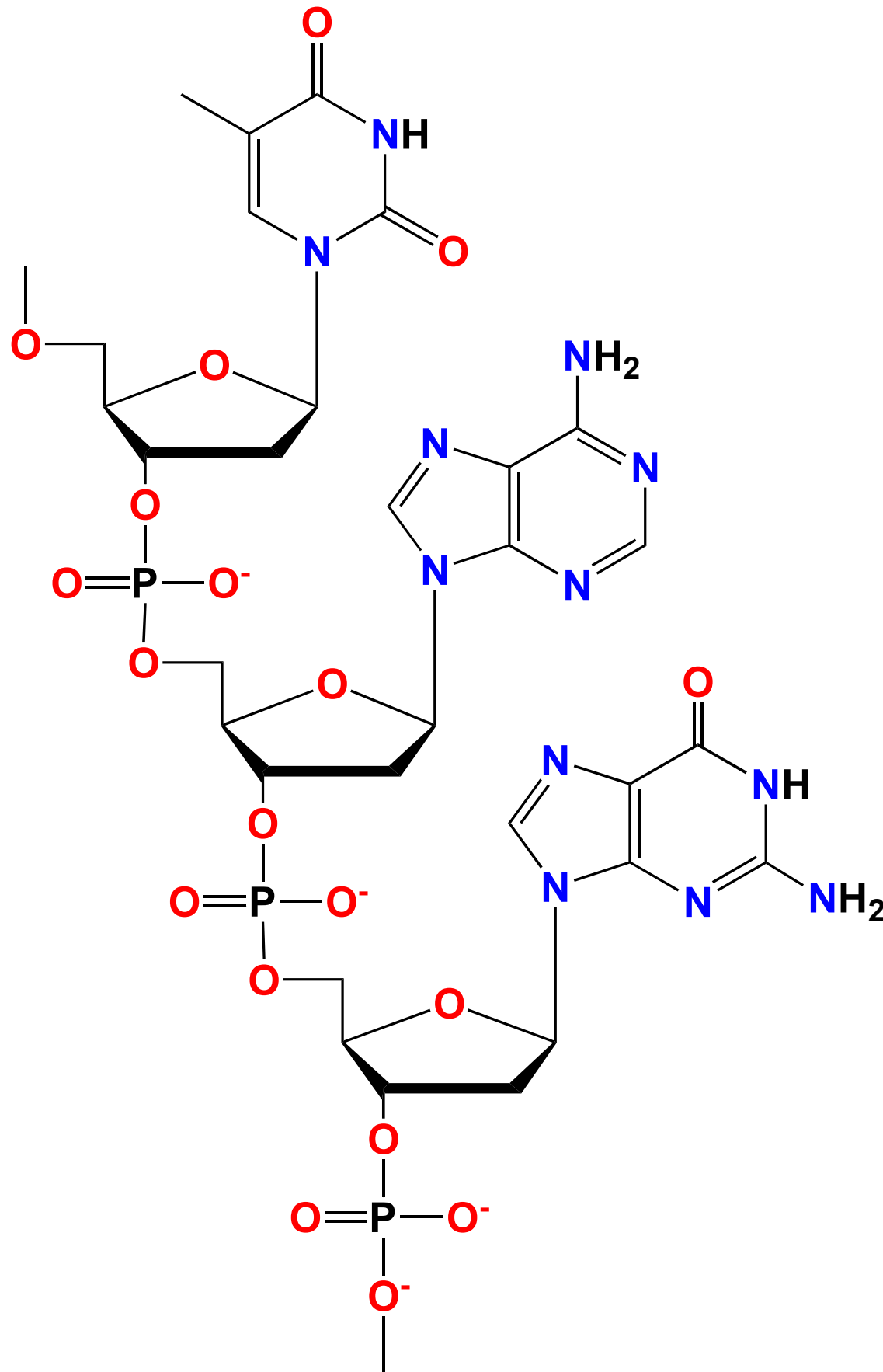


Ten H-Bonds

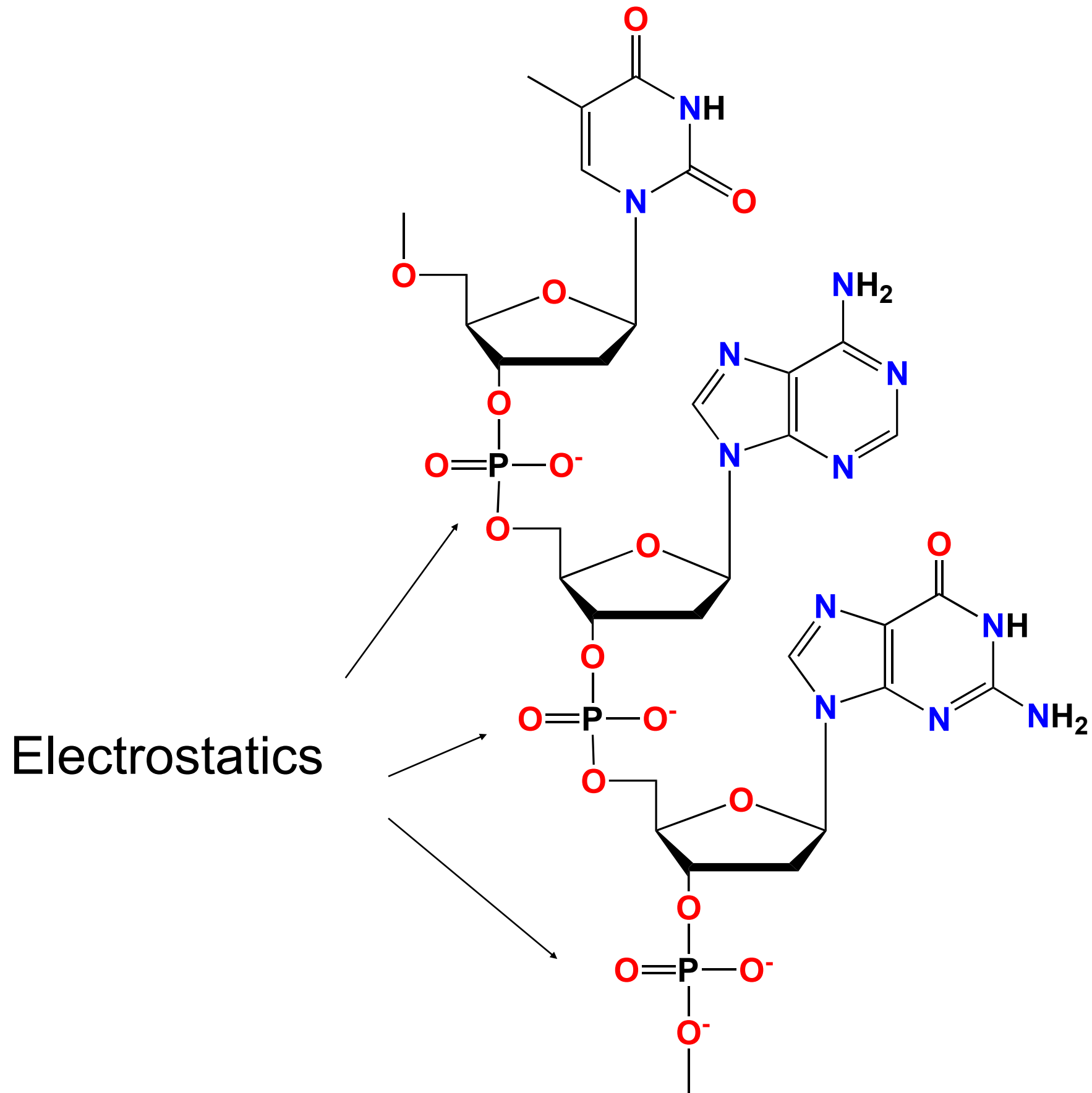
Ten H-Bonds



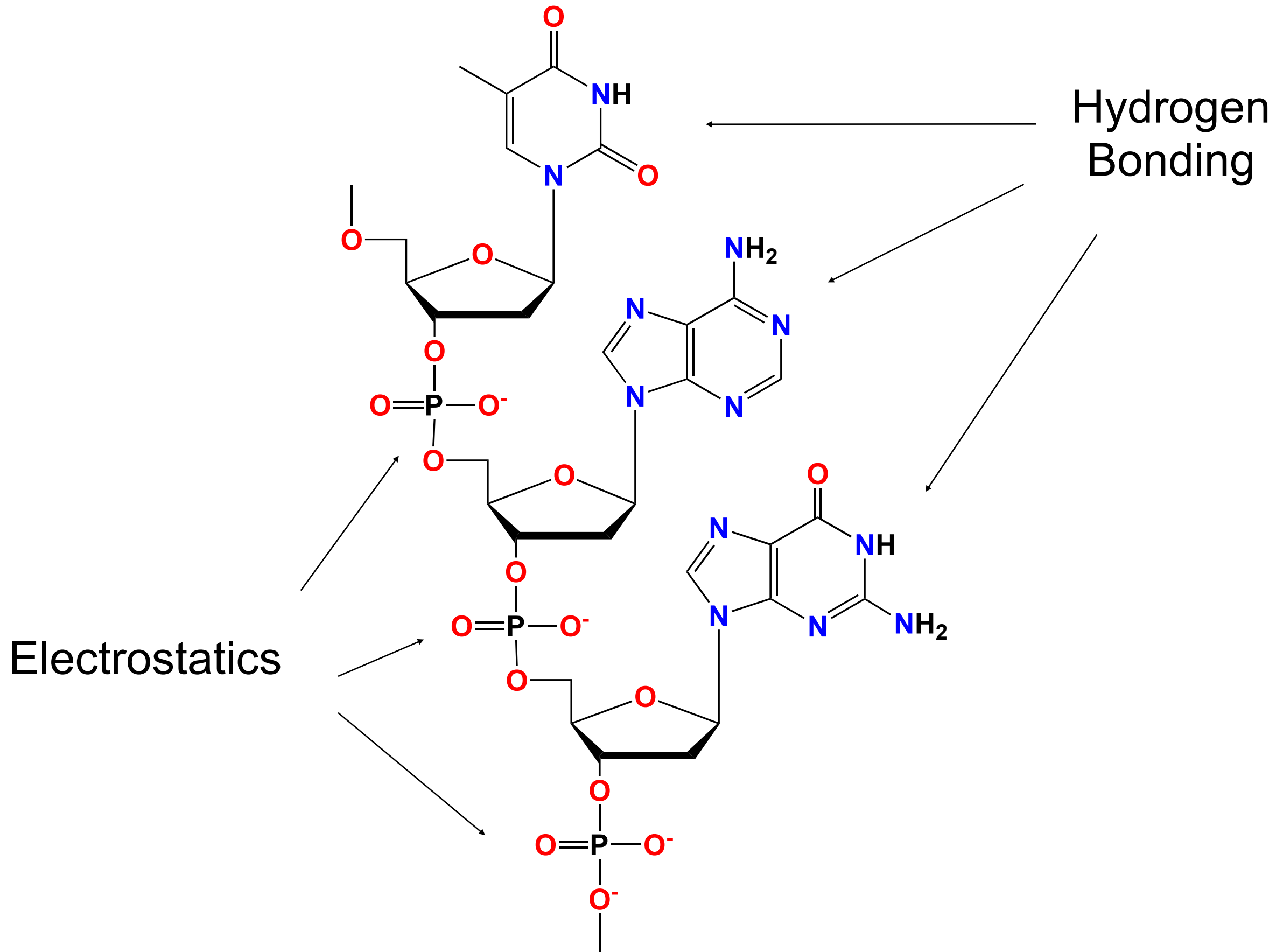
What *drives* double helix formation?



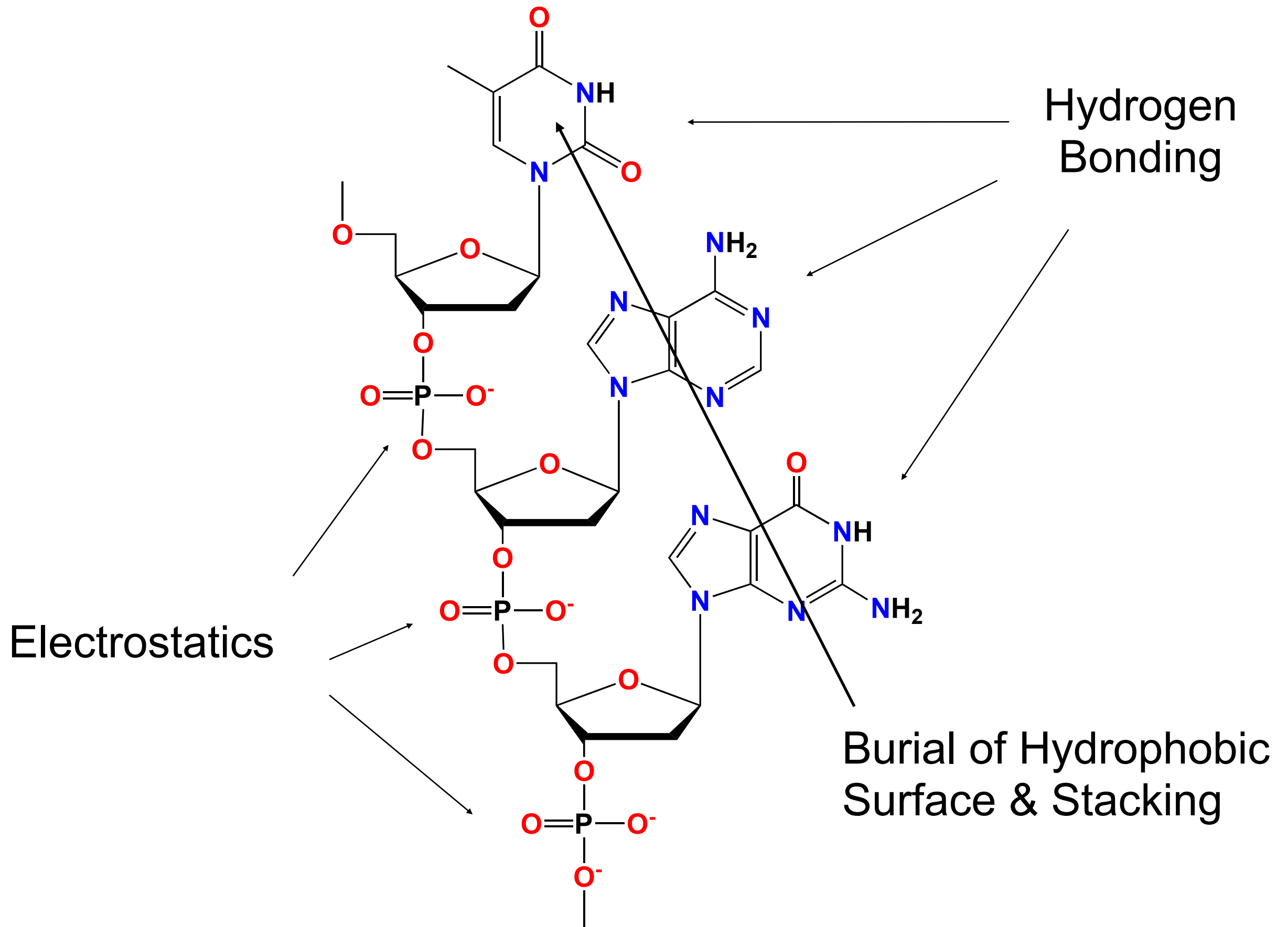
What *drives* double helix formation?



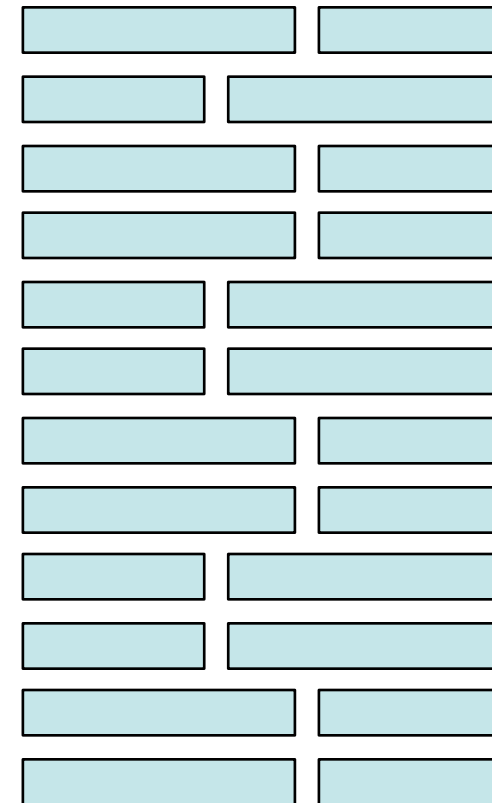
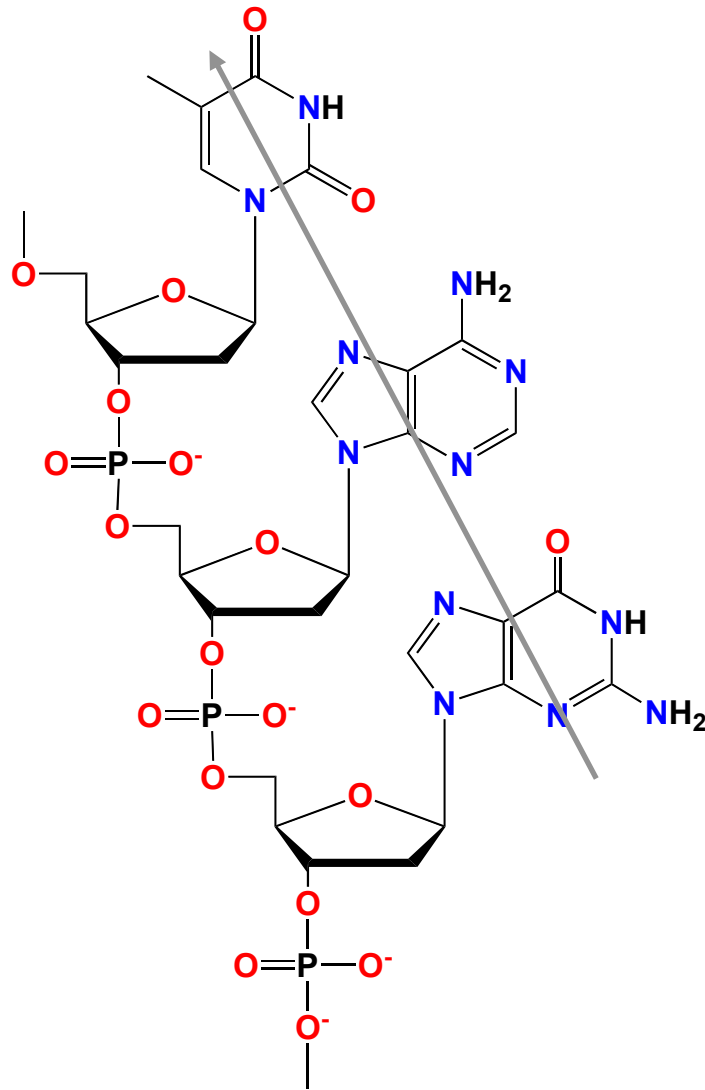
What *drives* double helix formation?



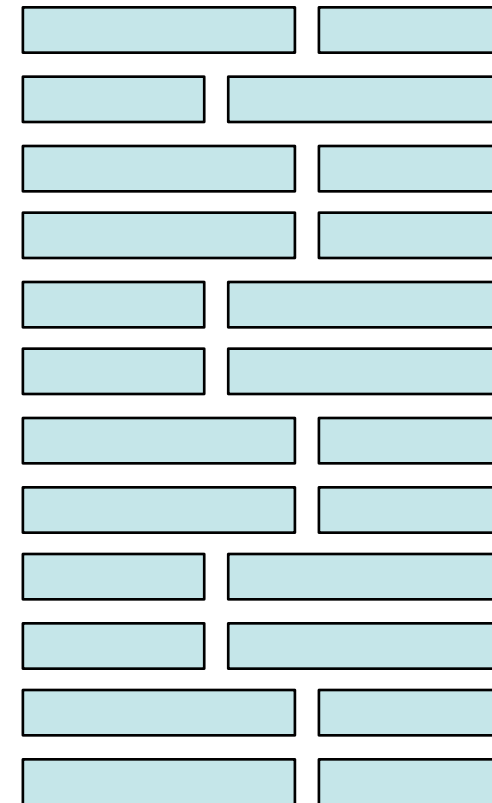
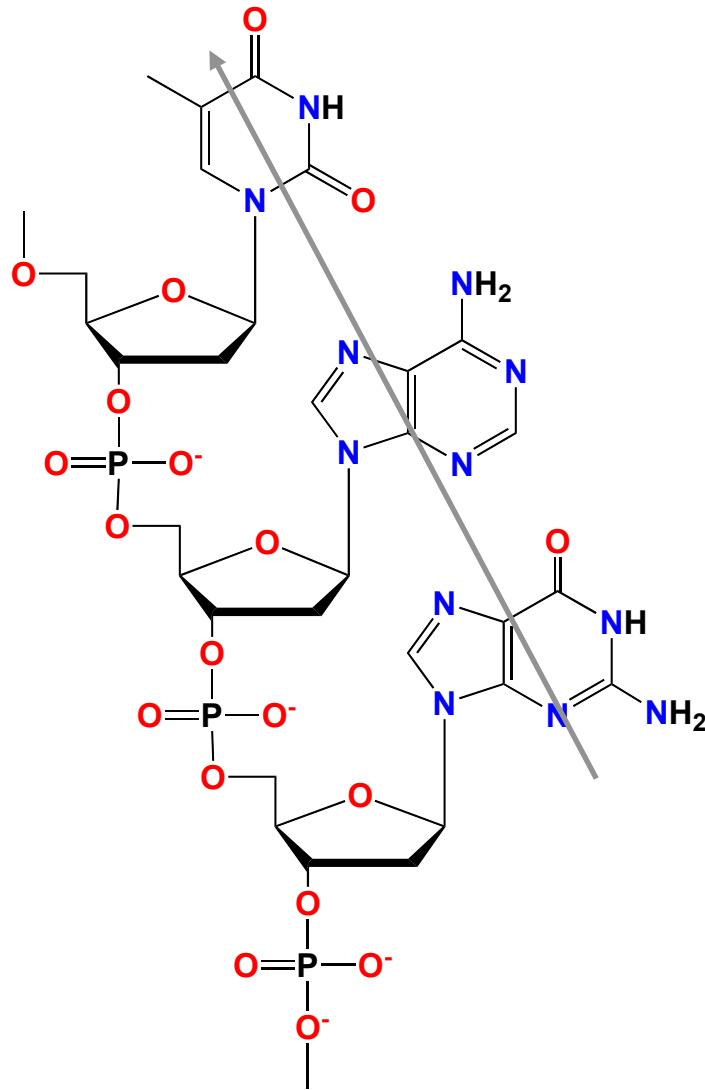
What *drives* double helix formation?



Burial of hydrophobic surface drives helix formation (hydrophobic core / stacking interactions)

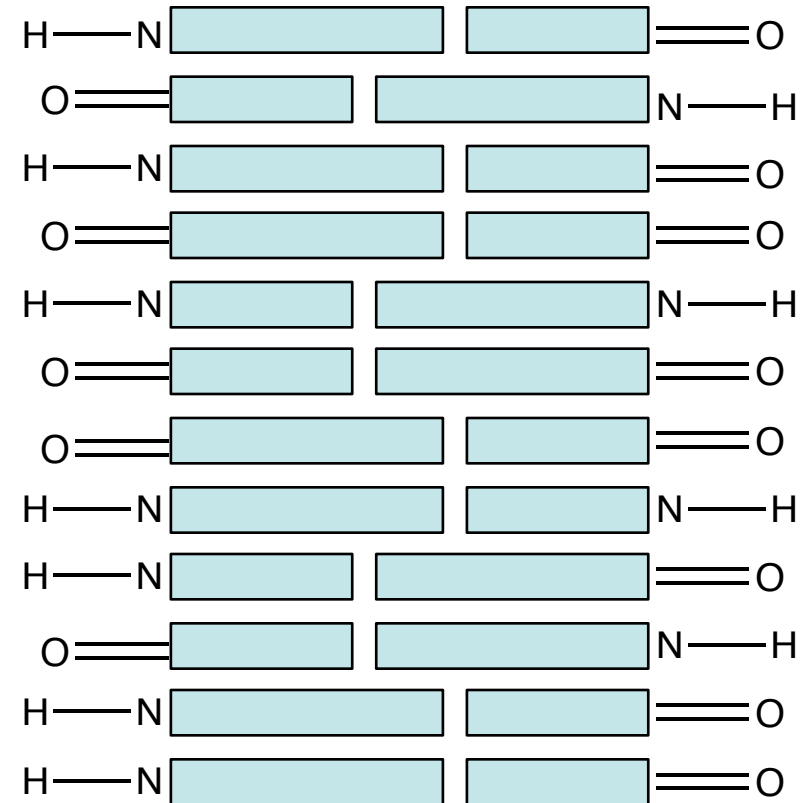
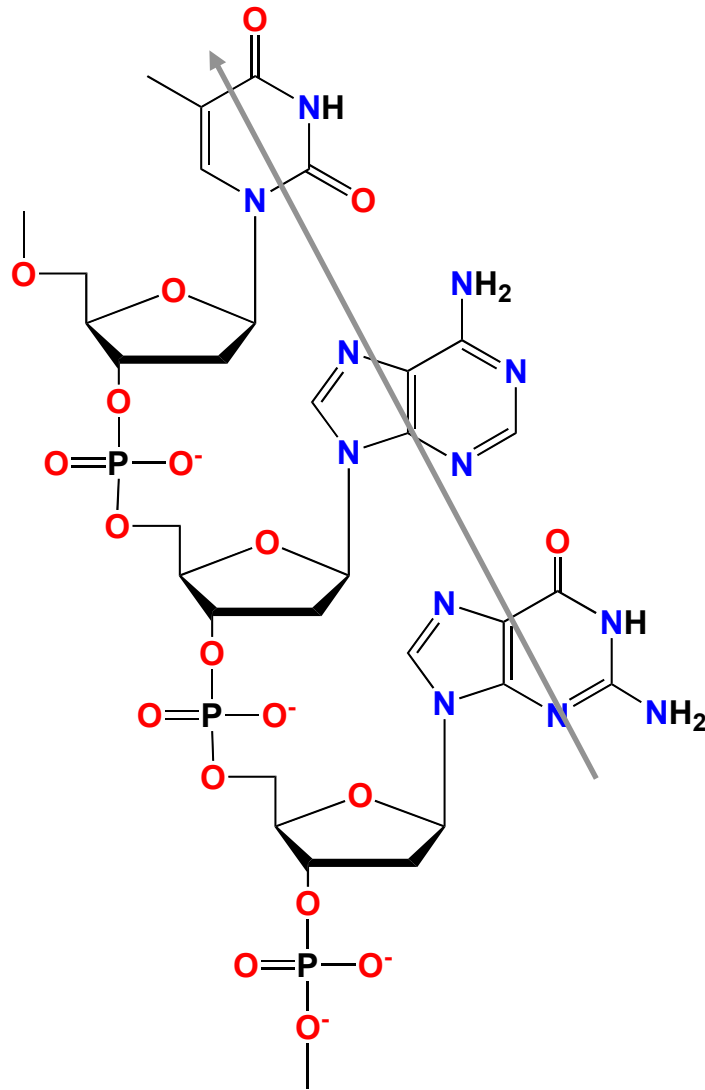


Burial of hydrophobic surface drives helix formation (hydrophobic core / stacking interactions)



Flat faces are nonpolar

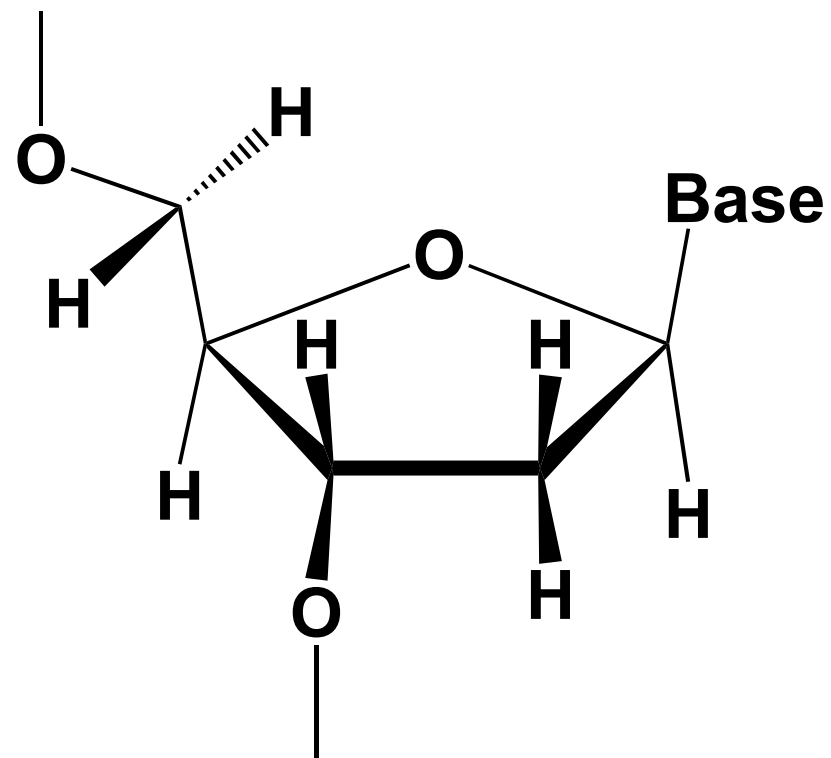
Burial of hydrophobic surface drives helix formation (hydrophobic core / stacking interactions)



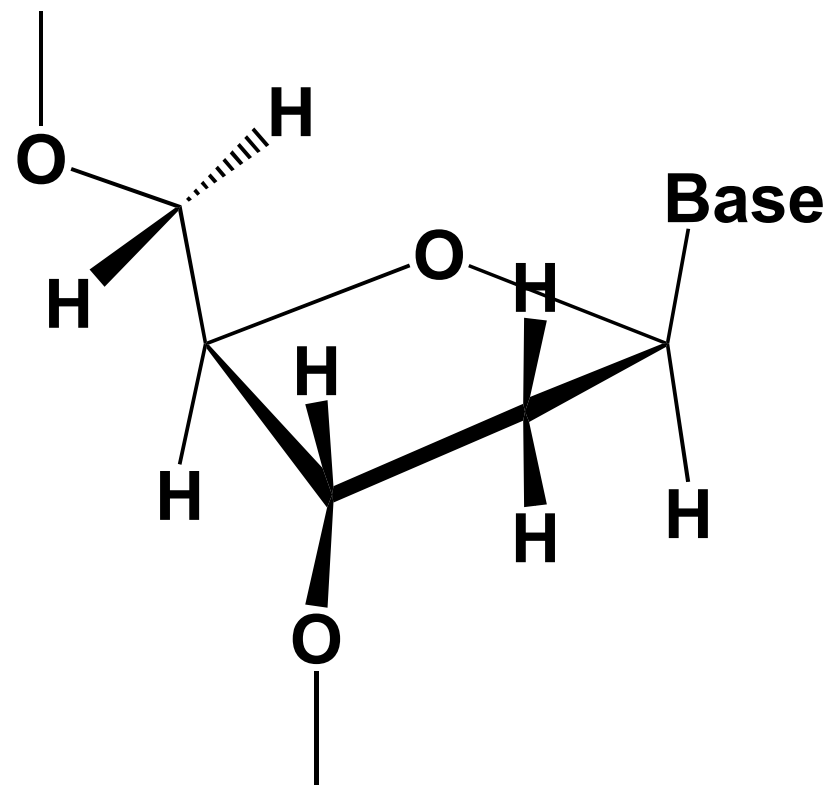
Flat faces are nonpolar
Edges are very polar (can H-bond)

Other chemical constraints

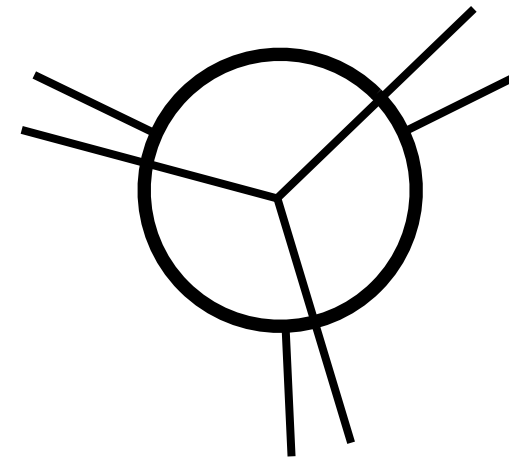
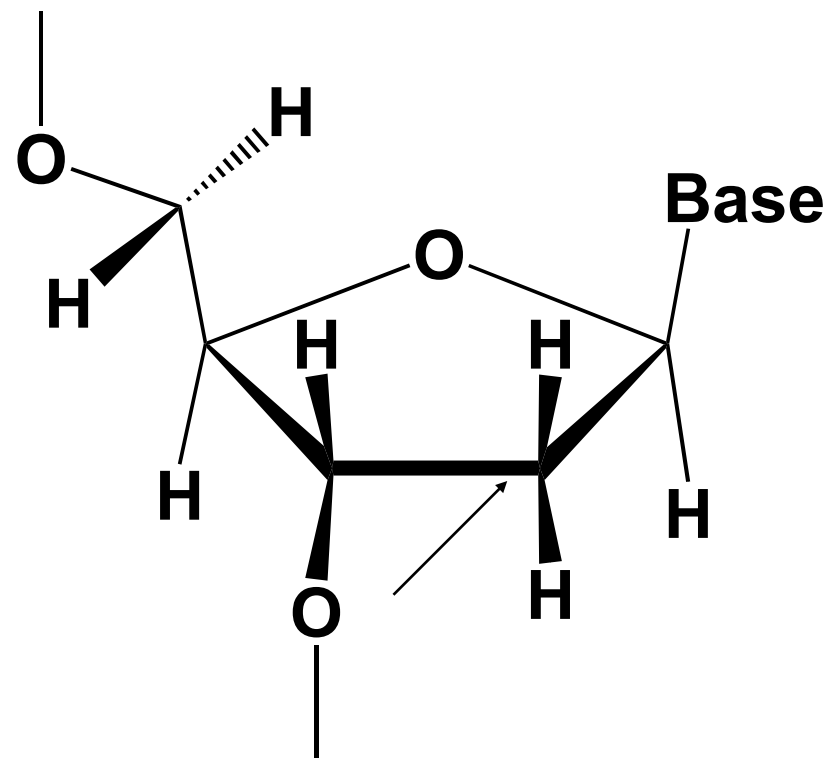
Furanose Sugar Ring



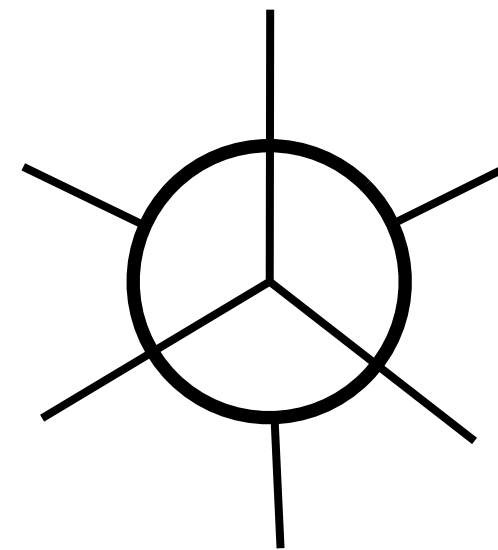
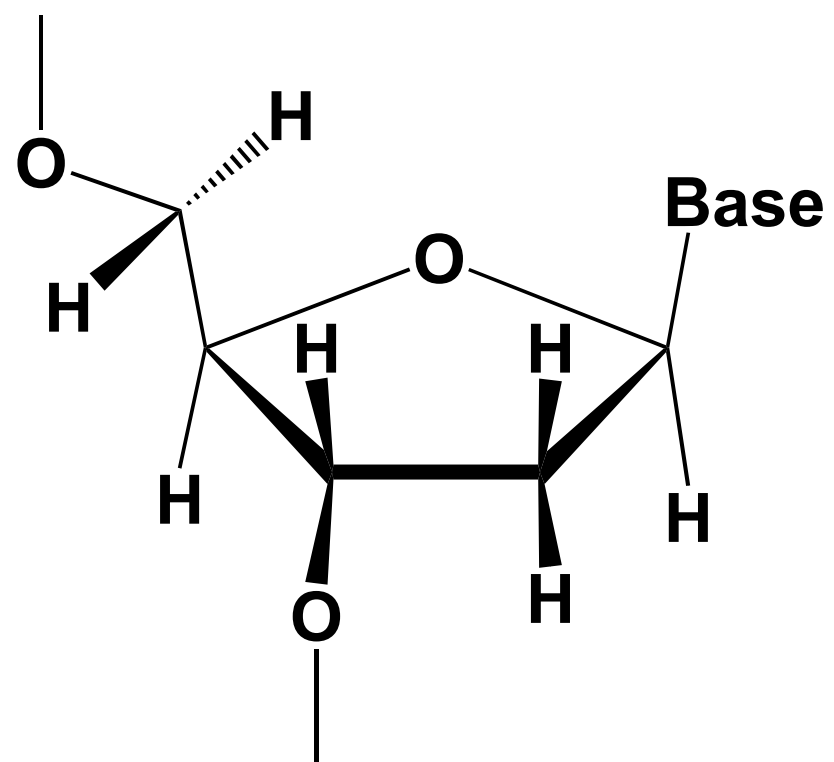
Furanose Sugar Ring



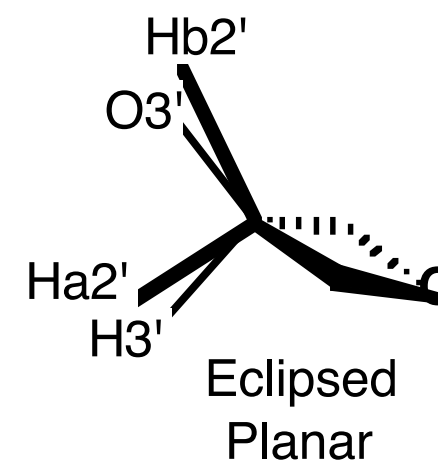
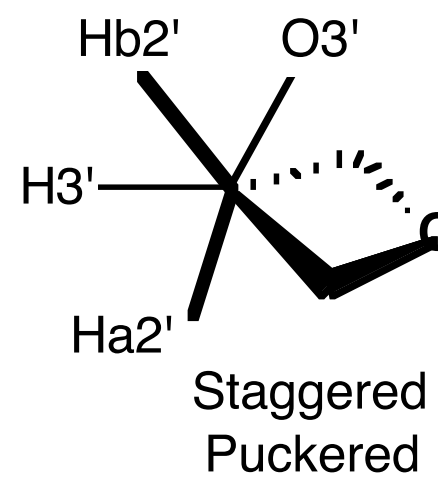
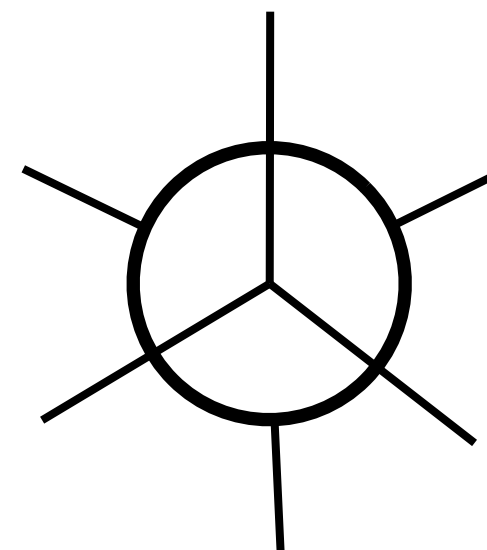
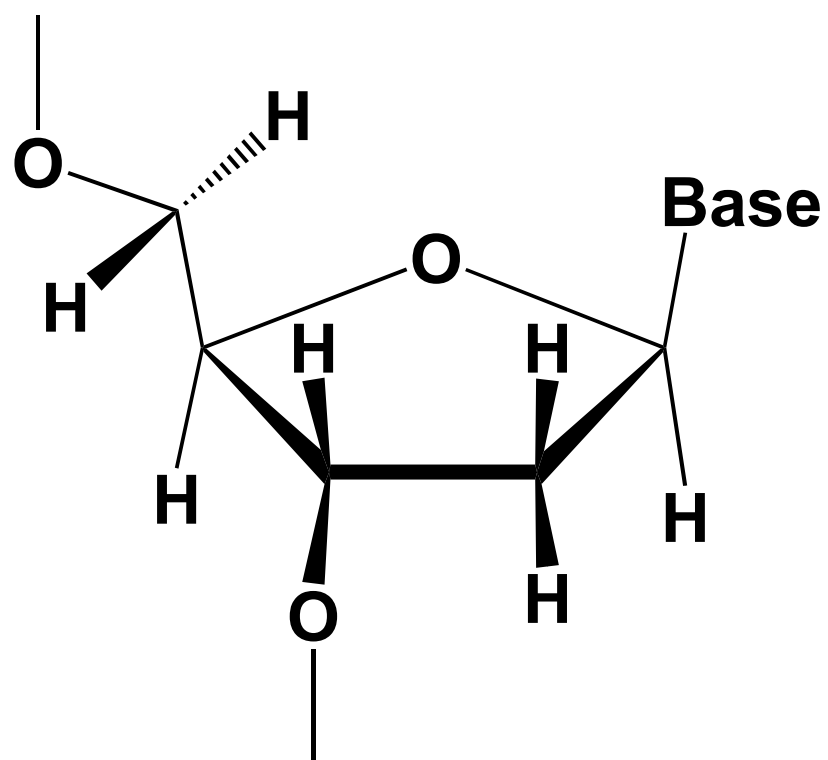
Furanose Sugar Ring



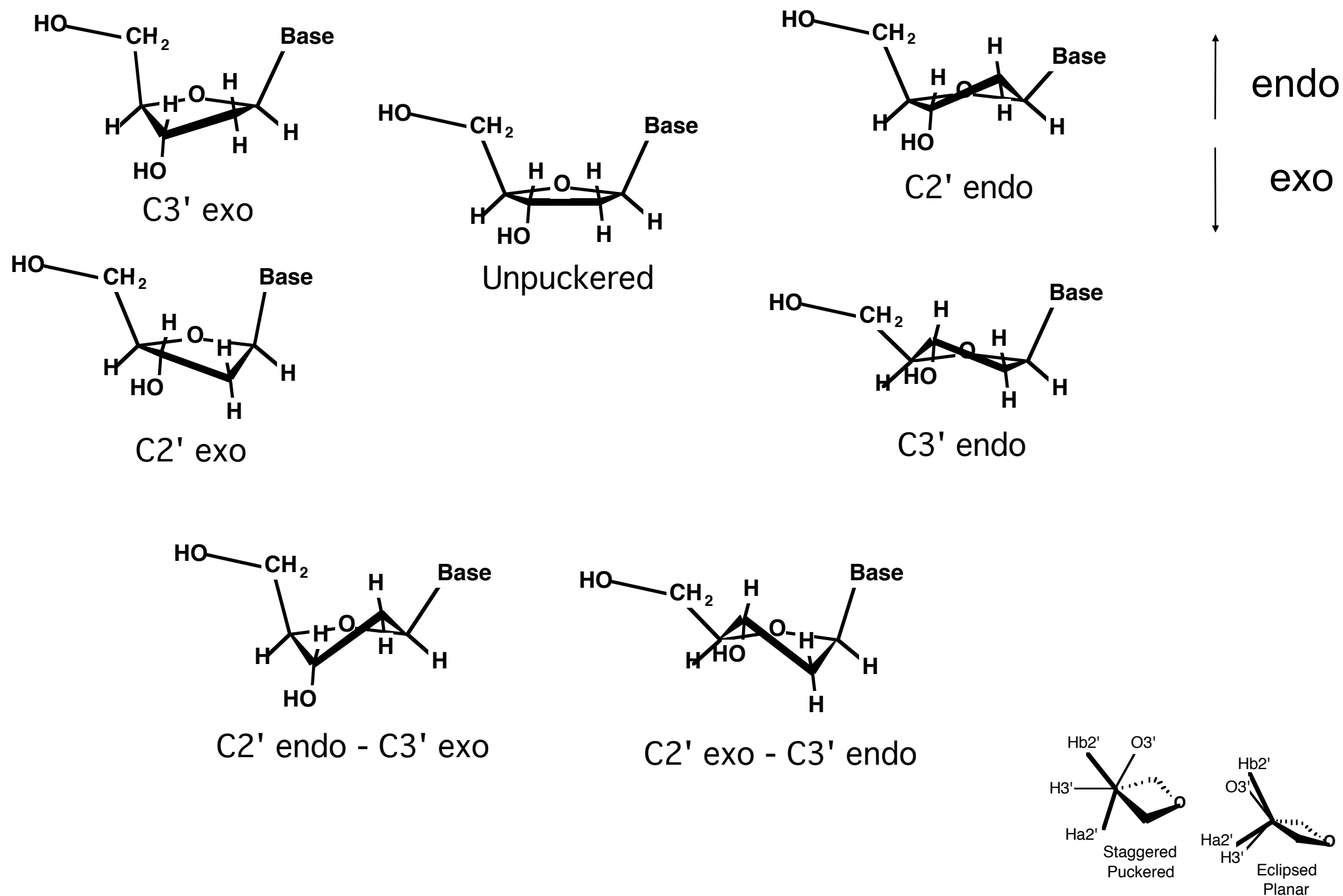
Furanose Sugar Ring

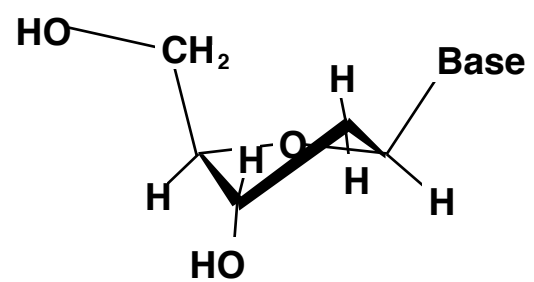


Furanose Sugar Ring

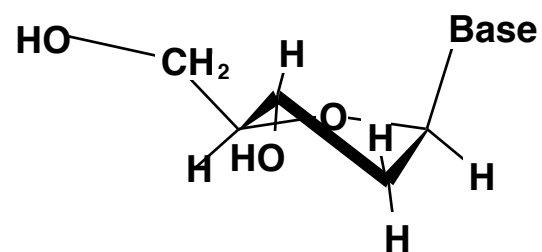


Furanose Sugar Ring



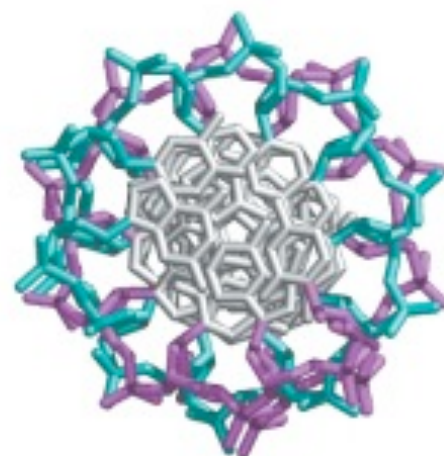
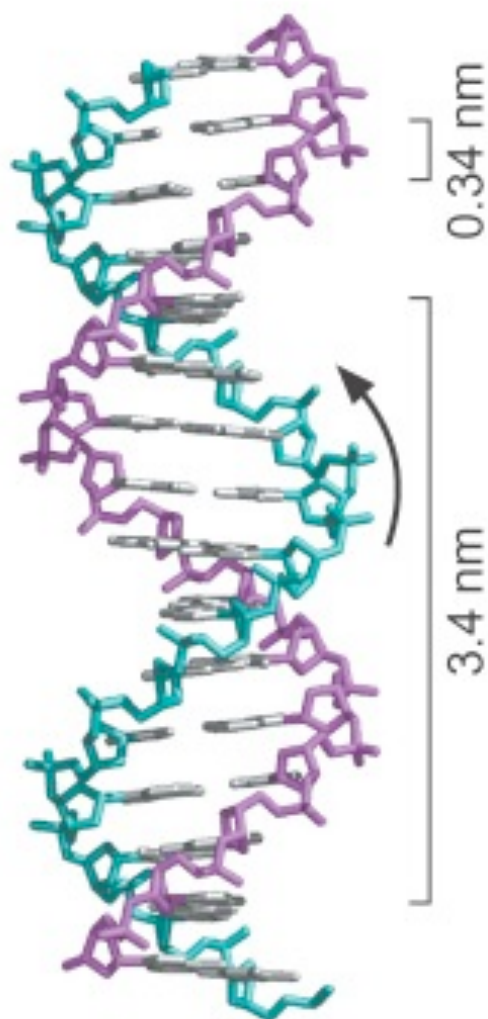


C2' endo - C3' exo

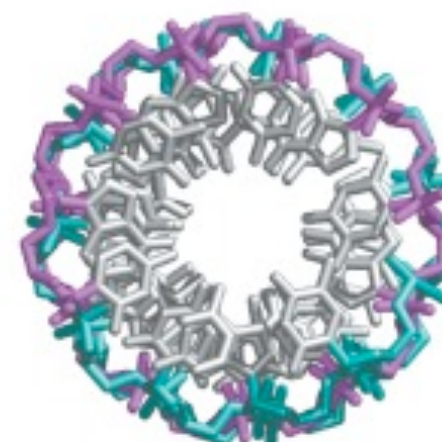
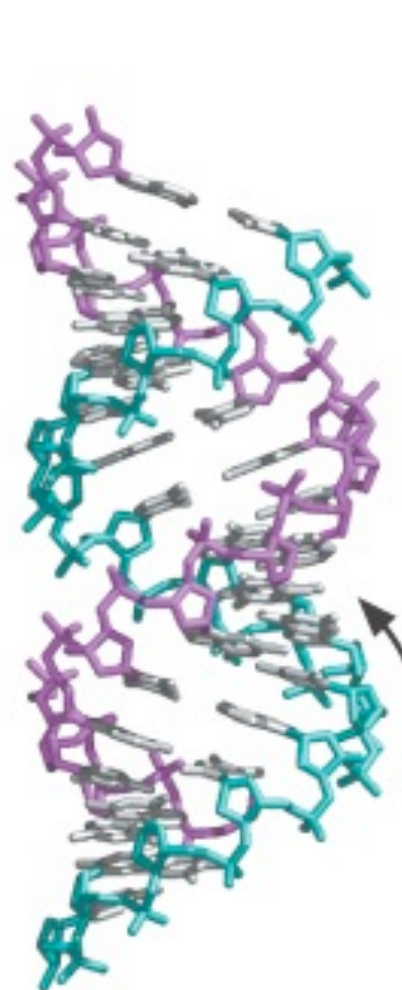


C2' exo - C3' endo

a B DNA



b A DNA



c Z DNA

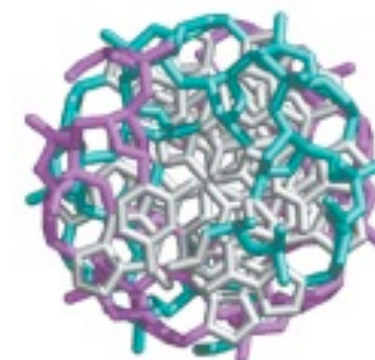
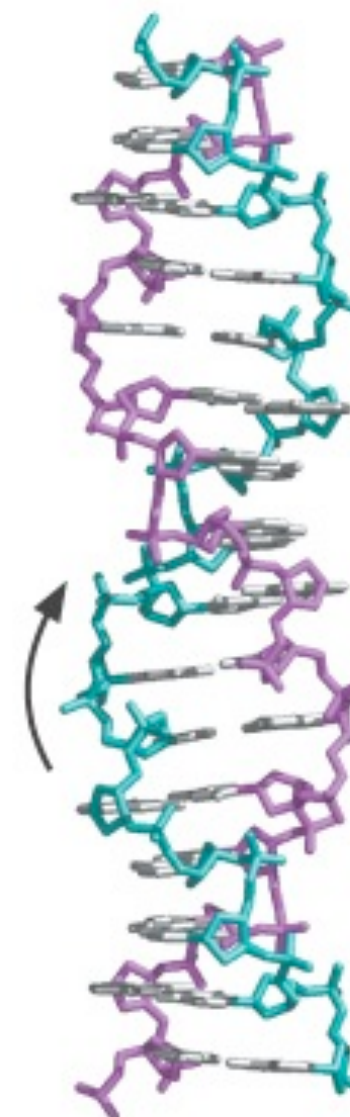


TABLE 6-2 A Comparison of the Structural Properties of A, B, and Z DNAs as Derived from Single-Crystal X-Ray Analysis

	Helix Type		
	A	B	Z
Overall proportions	Short and broad	Longer and thinner	Elongated and slim
Rise per base pair	2.3 Å	3.32 Å	3.8 Å
Helix-packing diameter	25.5 Å	23.7 Å	18.4 Å
Helix rotation sense	Right-handed	Right-handed	Left-handed
Base pairs per helix repeat	1	1	2
Base pairs per turn of helix	~11	~10	12
Rotation per base pair	33.6°	35.9°	−60° per 2 bp
Pitch per turn of helix	24.6 Å	33.2 Å	45.6 Å
Tilt of base normals to helix axis	+19°	−1.2°	−9°
Base-pair mean propeller twist	+18°	+16°	~0°
Helix axis location	Major groove	Through base pairs	Minor groove
Major-groove proportions	Extremely narrow but very deep	Wide and of intermediate depth	Flattened out on helix surface
Minor-groove proportions	Very broad but shallow	Narrow and of intermediate depth	Extremely narrow but very deep
Glycosyl-bond conformation	<i>anti</i>	<i>anti</i>	<i>anti</i> at C, <i>syn</i> at G

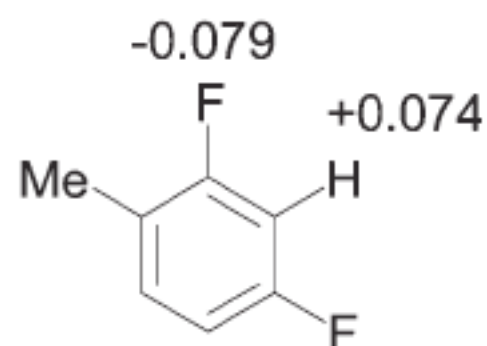
Source: Adapted from Dickerson R. E. et al. 1982. *CSHSQB* 47: 14. Copyright © 1982 Cold Spring Harbor Laboratory Press. Used with permission.

DifluoroToluene

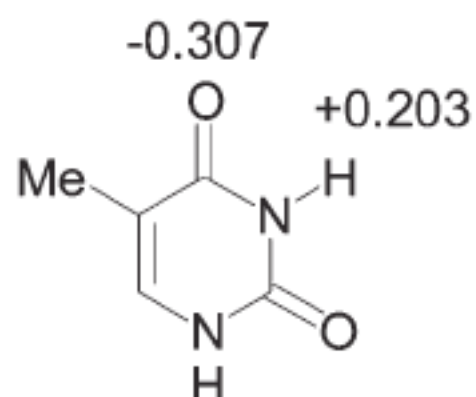
The difluorotoluene debate—a decade later

Eric T. Kool* and Herman O. Sintim

Chem. Commun., 2006, 3665–3675



Difluorotoluene



Thymine

Table 2 Pairing stabilities of 12 bp DNA duplexes containing difluorotoluene or thymine paired opposite natural bases

5'-C T T T T C X T T C T T

3'-G A A A A G Y A A G A A

X.Y	$T_m/^\circ\text{C}$	$\Delta G^\circ_{37}/\text{kcal mol}^{-1}$	X.Y	$T_m/^\circ\text{C}$	$\Delta G^\circ_{37}/\text{kcal mol}^{-1}$
T.A	42.4	−9.7	A.T	42.4	−9.6
T.G	33.3	−7.5	G.T	34.7	−7.9
T.C	29.5	−6.6	C.T	25.7	−6.0
T.T	29.1	−6.8	T.T	29.1	−6.8
F.A	26.2	−6.2	A.F	27.5	−6.6
F.G	23.6	−6.0	G.F	25.1	−6.4
F.C	23.7	−5.8	C.F	25.1	−6.4
F.T	24.0	−5.9	T.F	24.8	−6.4

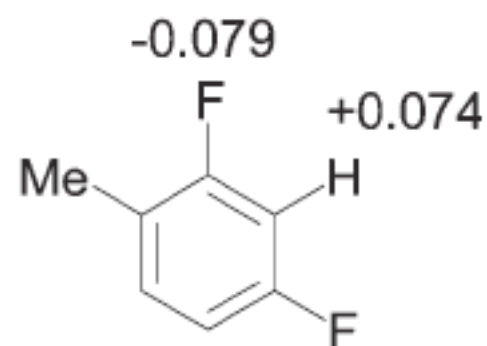
^a Conditions: 1 M NaCl, 10 mM phosphate buffer (pH 7.0) with 0.1 mM EDTA, 2.5 μM each strand, monitored at 260 nm. Error in T_m is estimated at $\pm 0.5^\circ\text{C}$, and in free energy, $\pm 10\%$.

DifluoroToluene

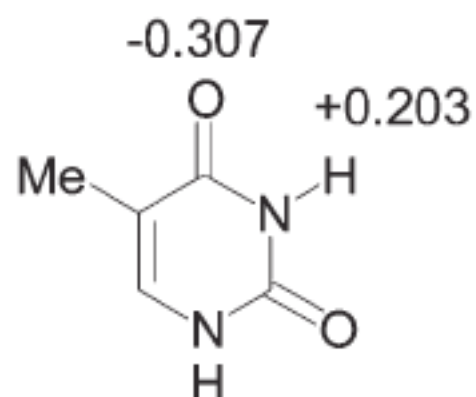
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F.A	26.2	-6.2	A.F	27.5	-6.6
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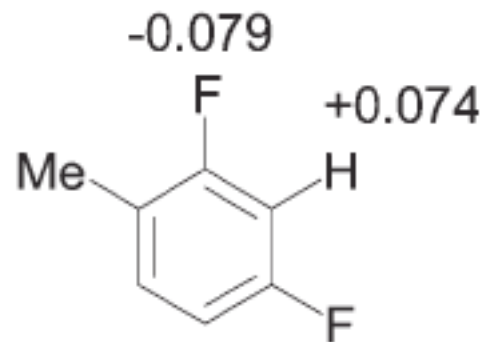
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DifluoroToluene

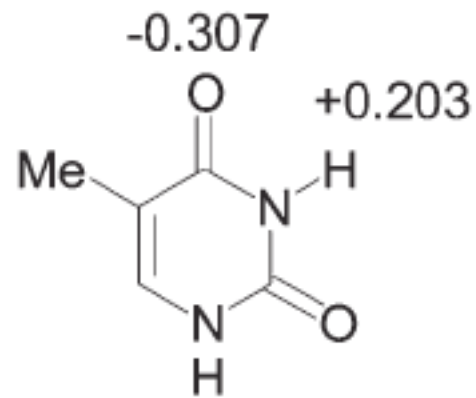
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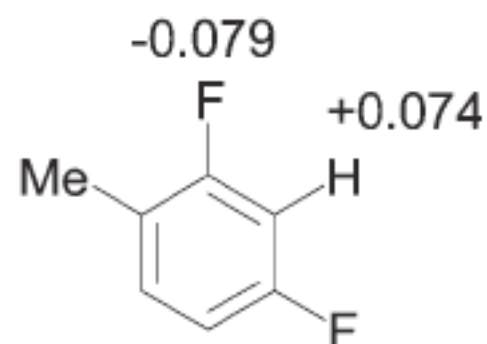
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DifluoroToluene

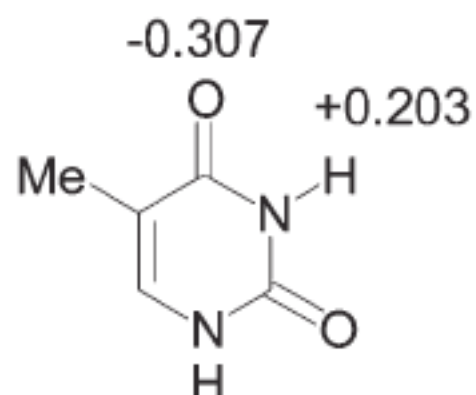
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- Difluorotoluene pairs poorly with Adenine, in contrast to the normal partner Thymine
- DNA polymerase incorporates T opposite A
- DNA polymerase does not incorporate G, C, or A opposite A

DifluoroToluene

The difluorotoluene debate—a decade later

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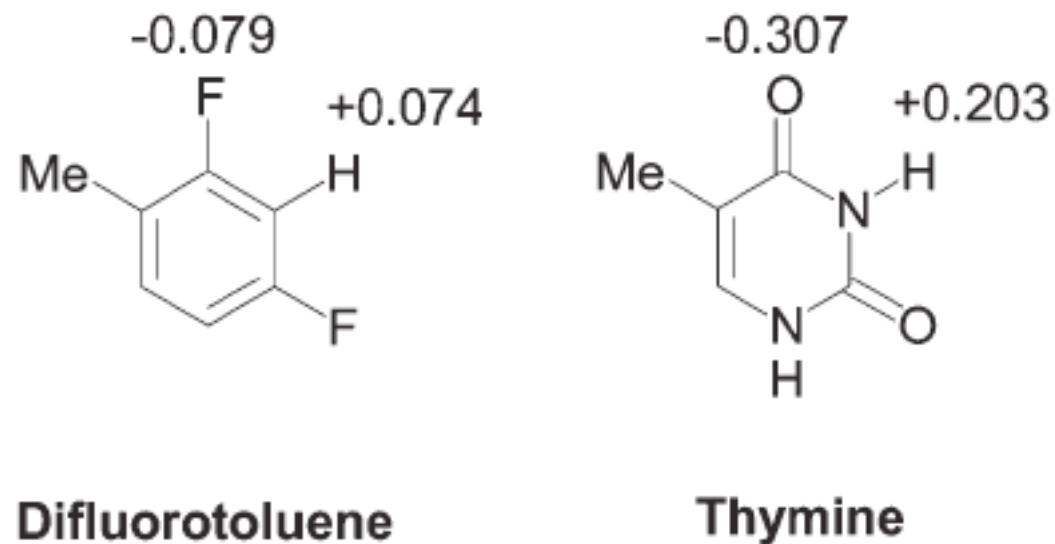


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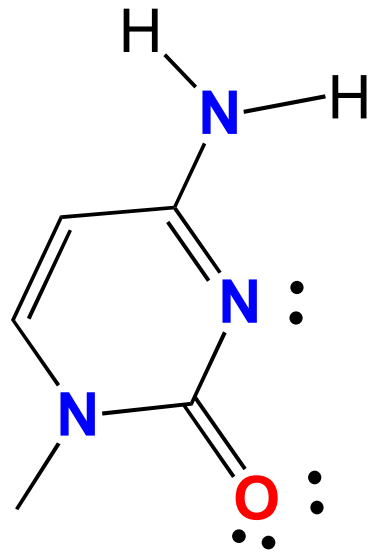
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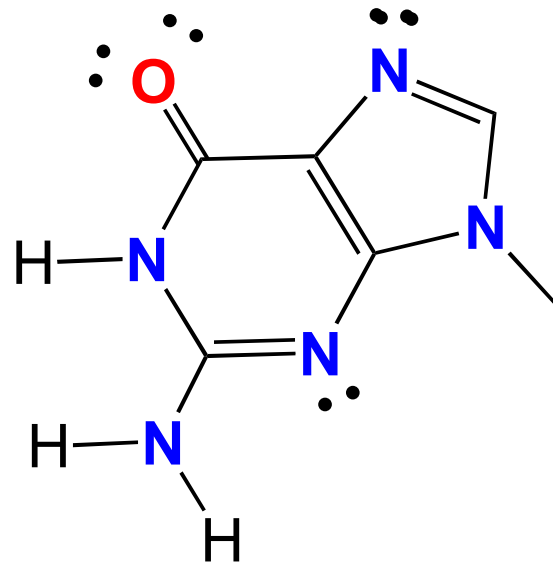
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- DNA polymerase incorporates Difluorotoluene opposite A

Why is Watson-Crick so good?

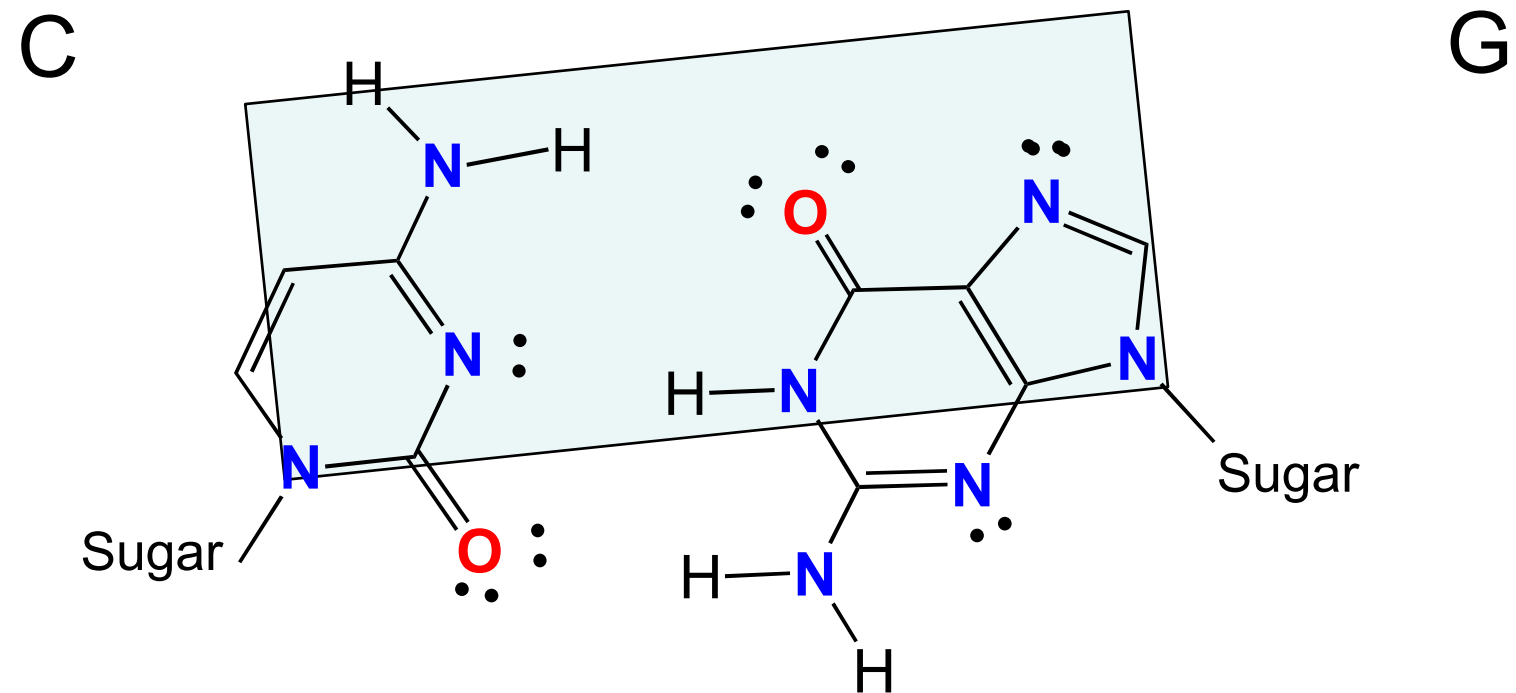
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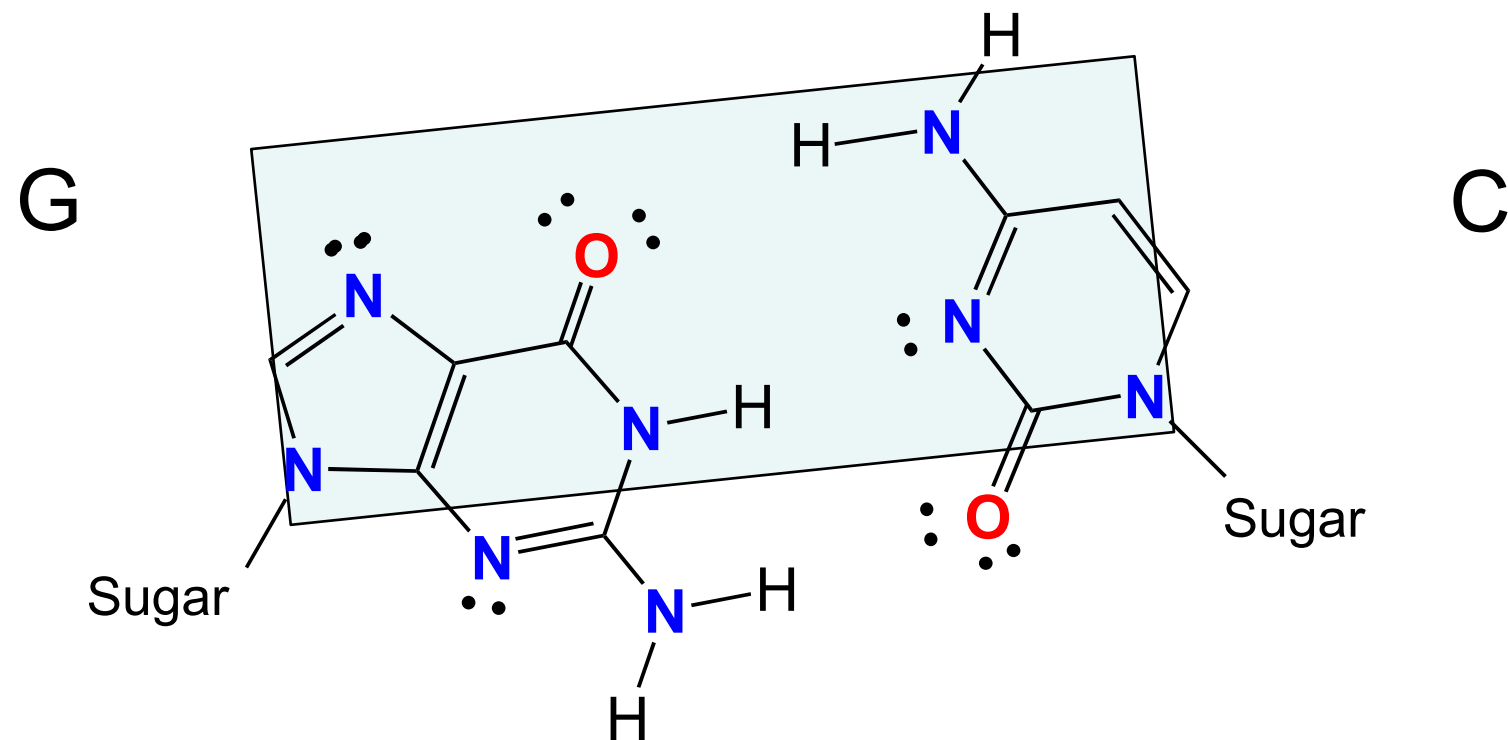
G



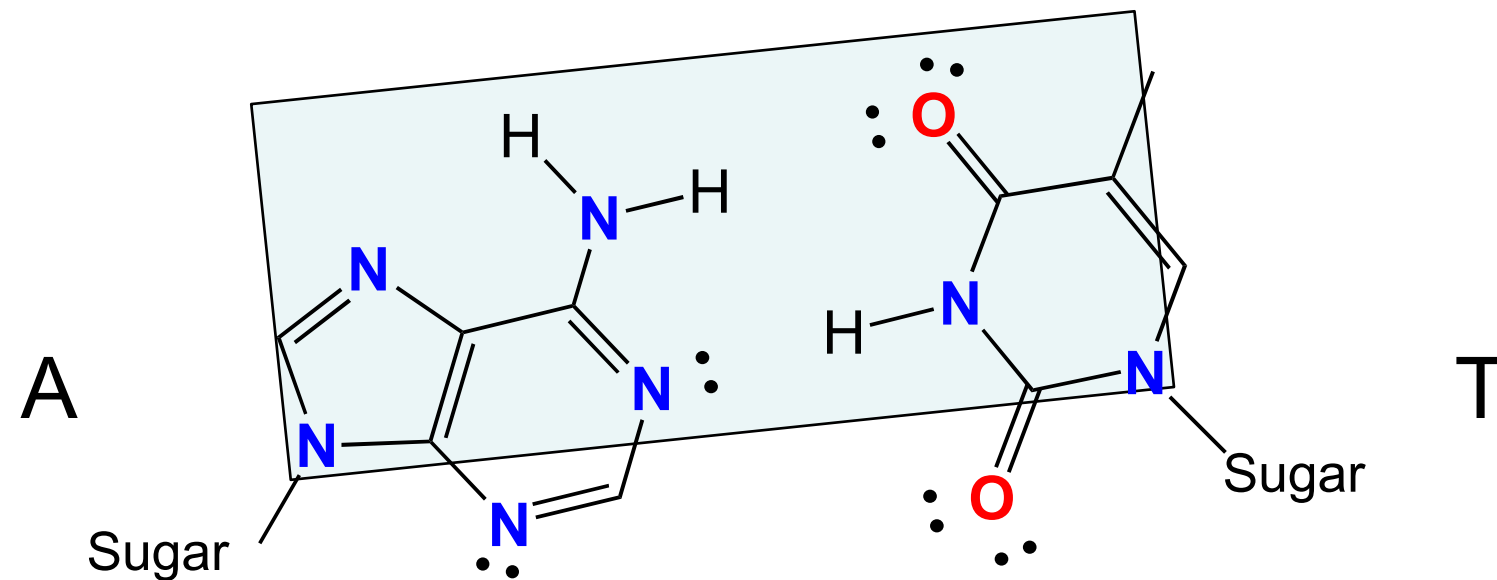
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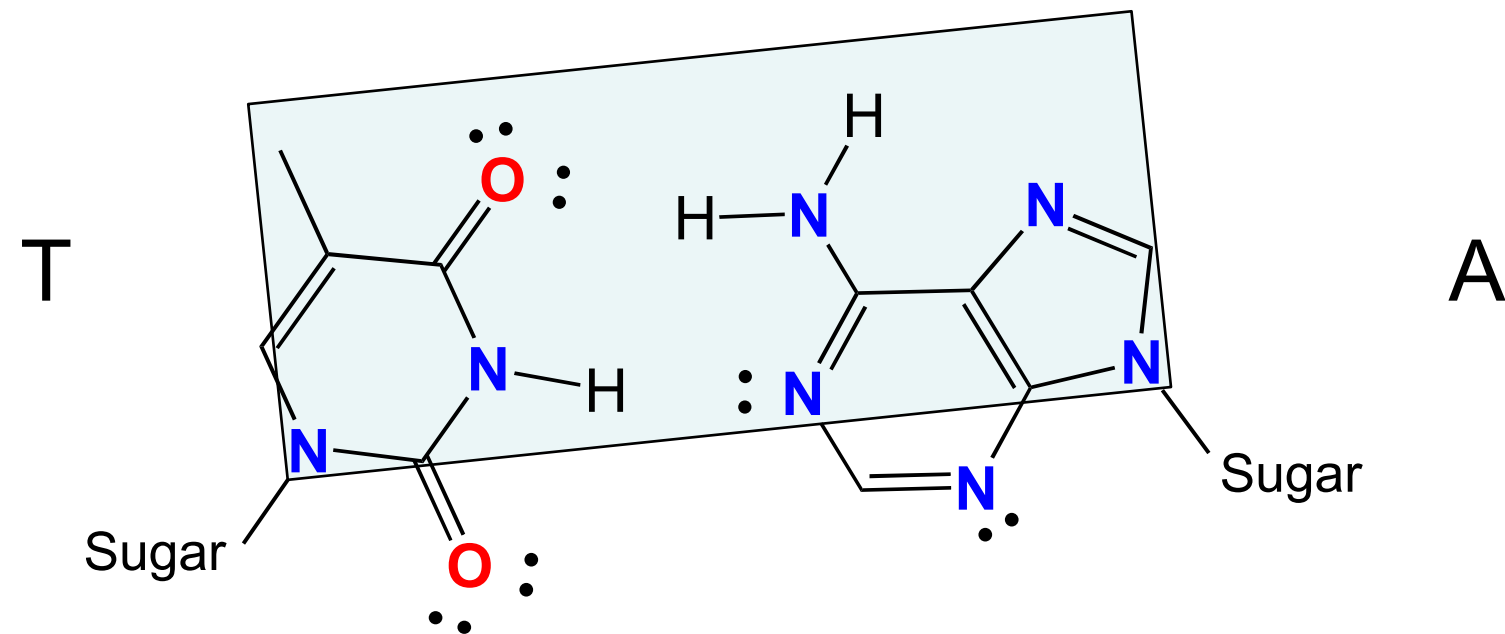
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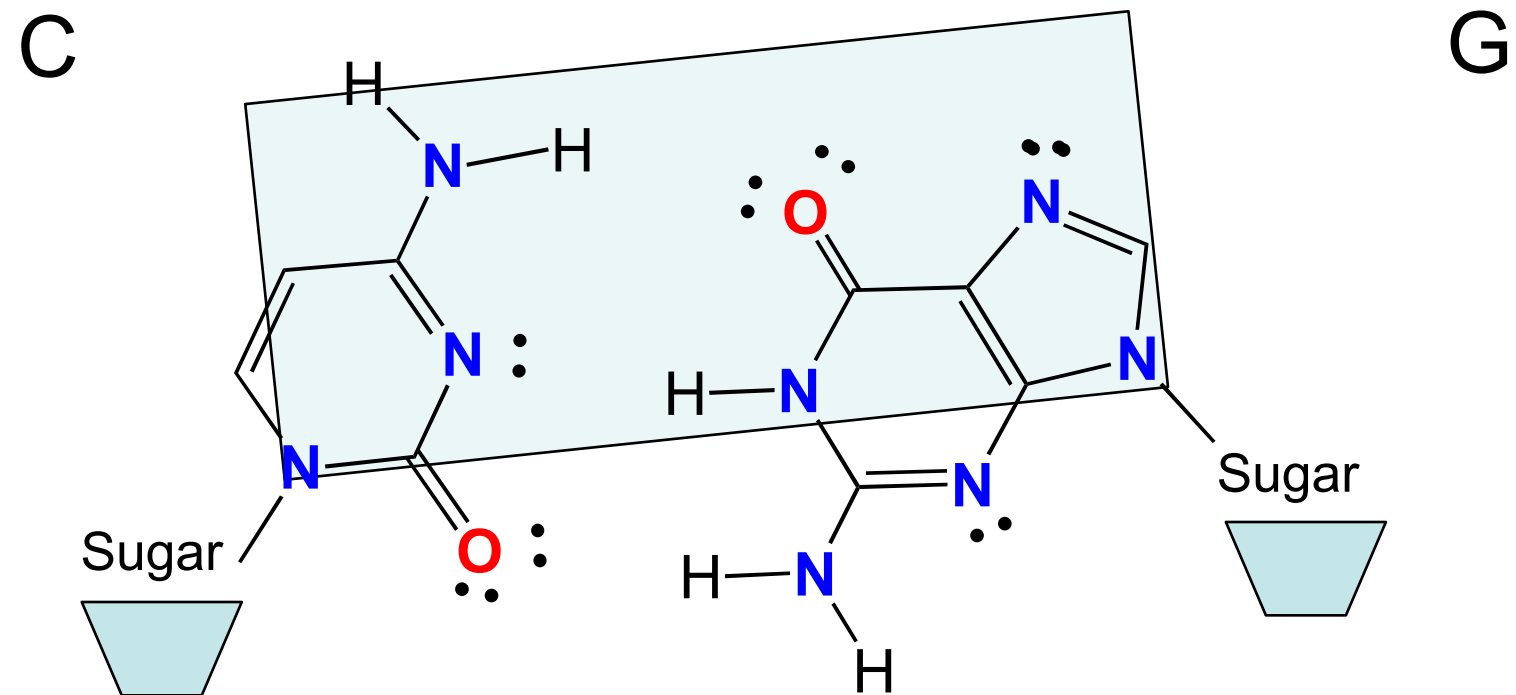
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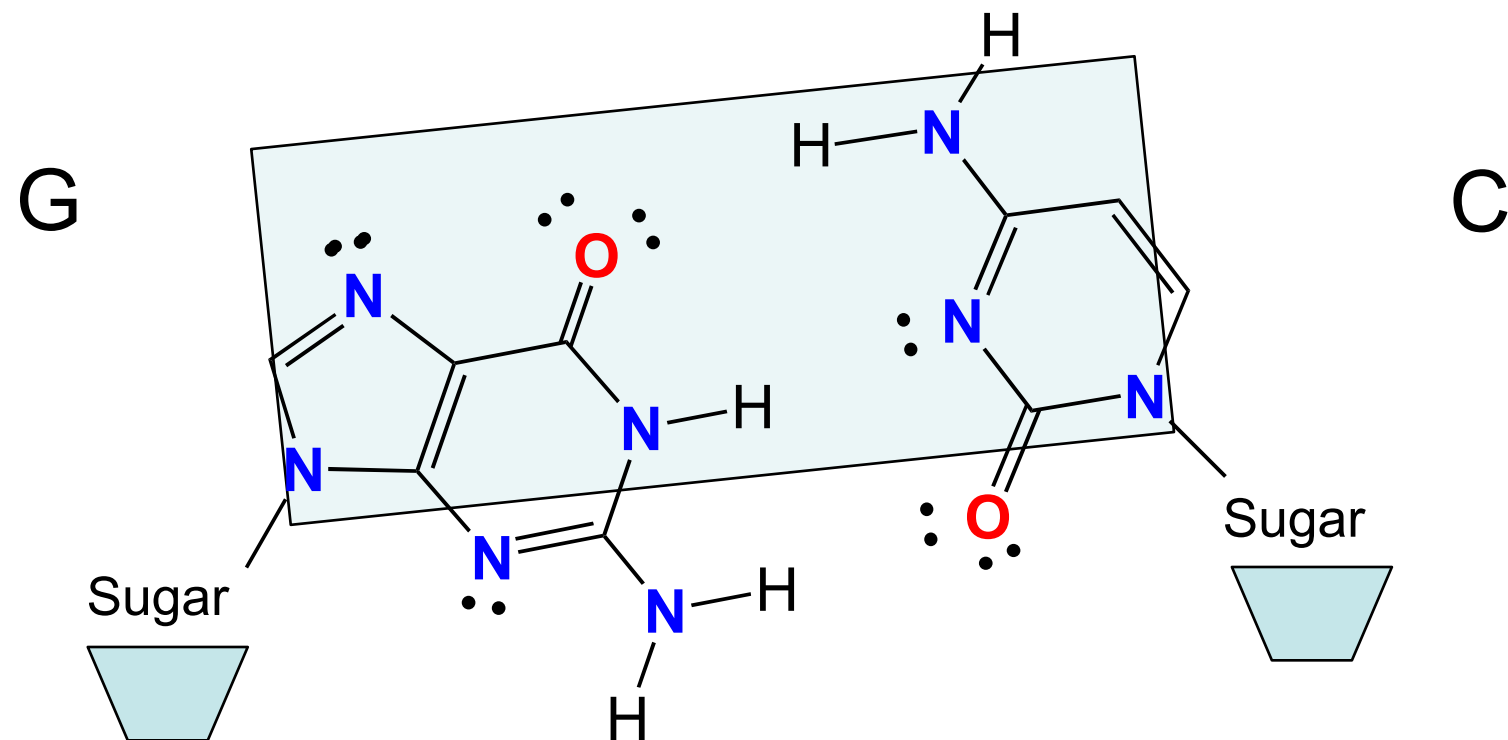
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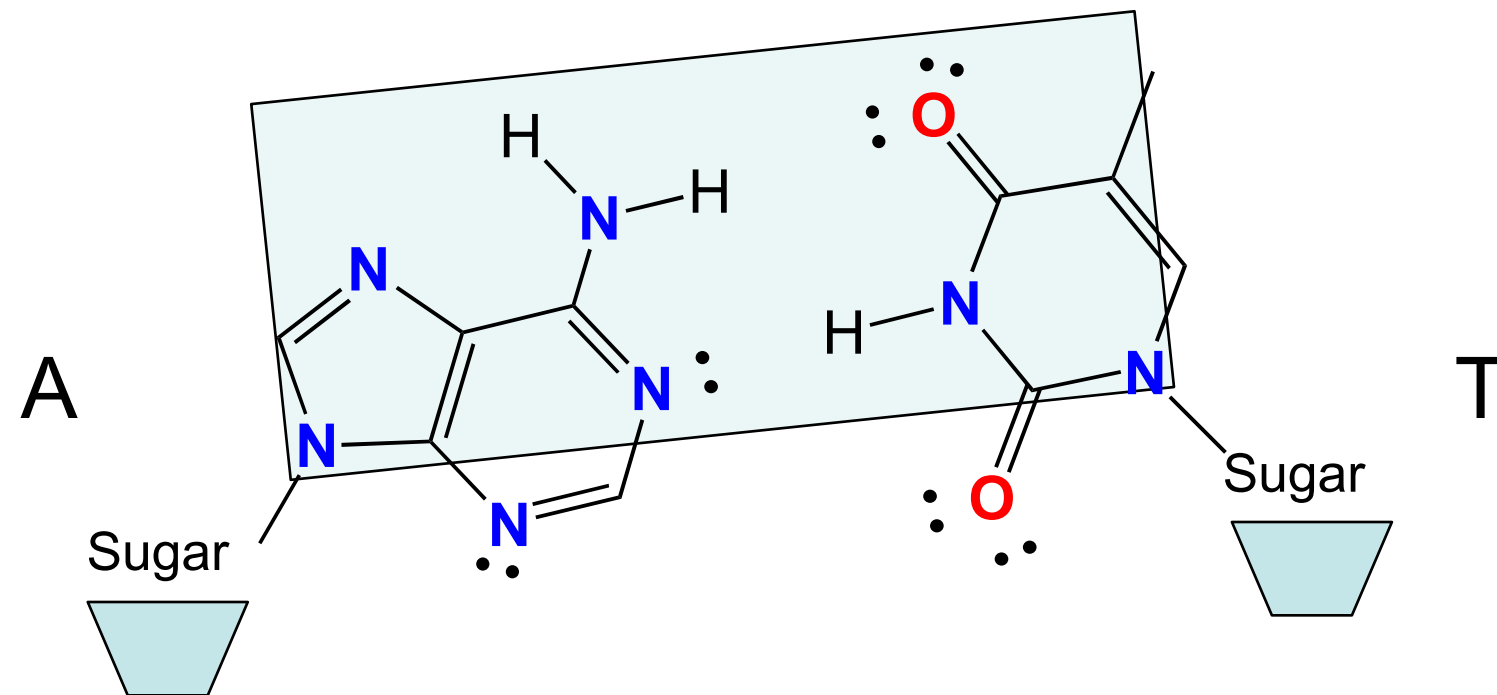
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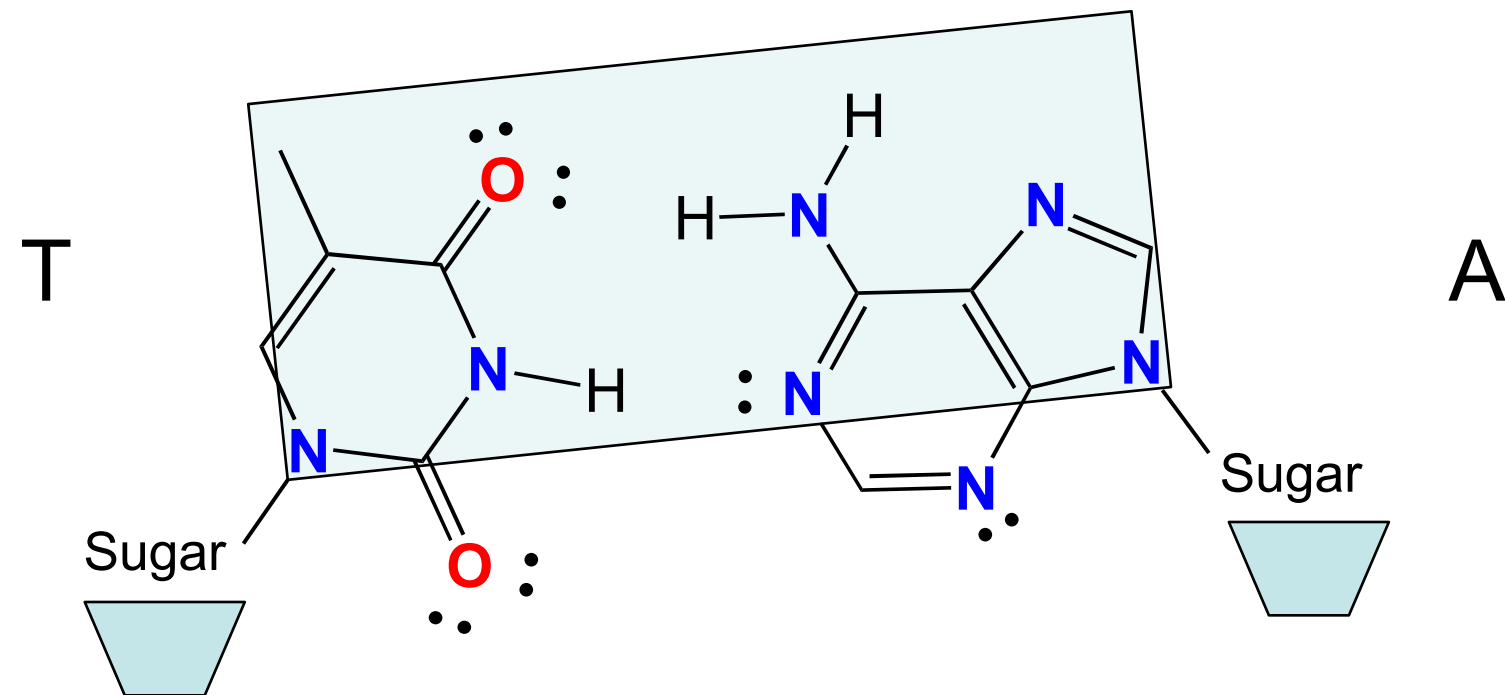
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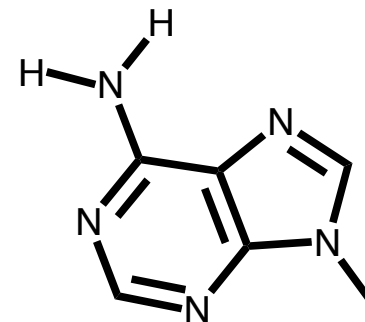
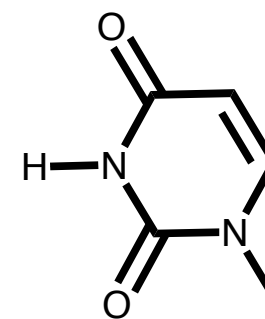
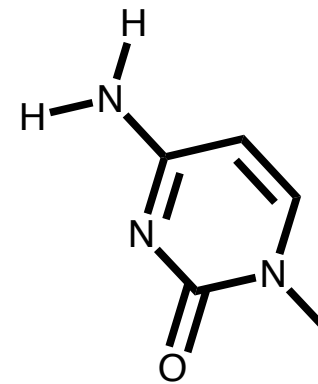
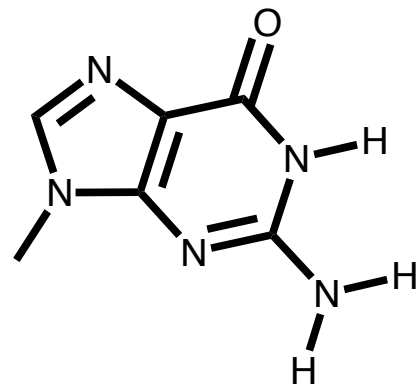
Why is Watson-Crick so good?



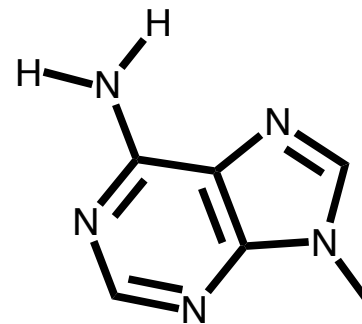
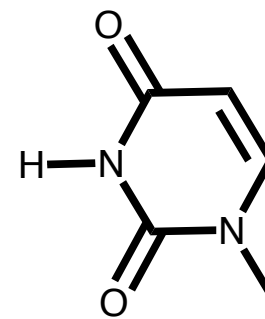
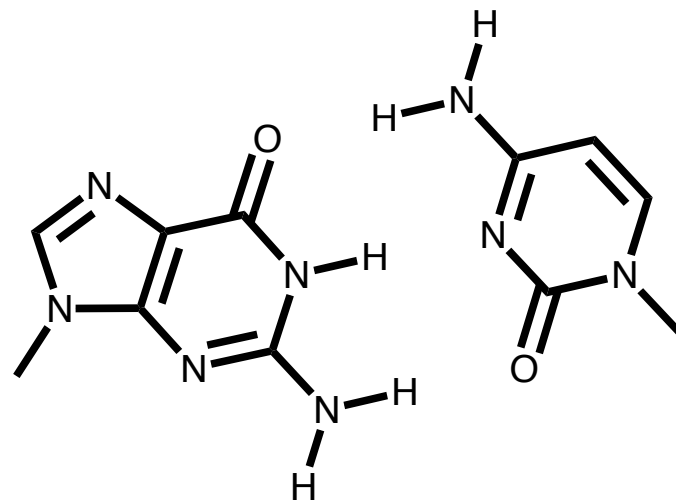
Why is Watson-Crick so good?



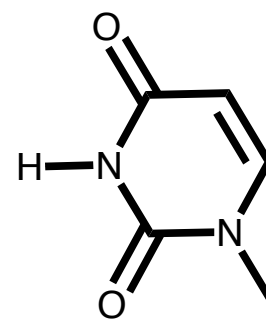
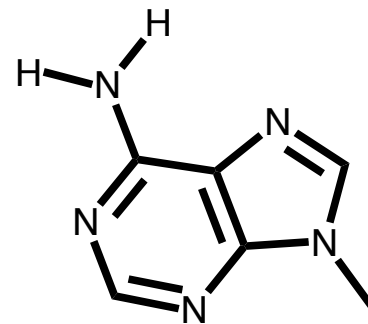
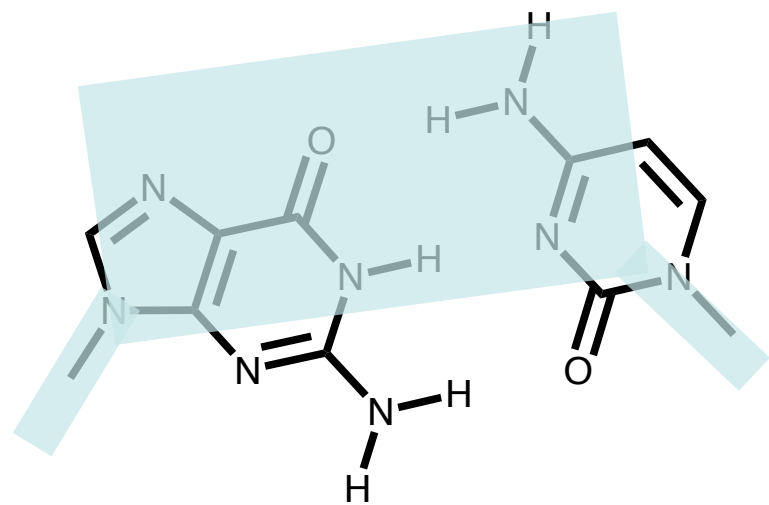
Base Pairs



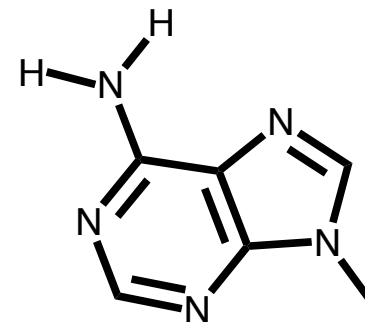
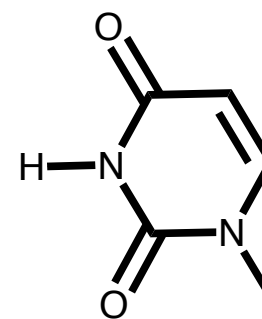
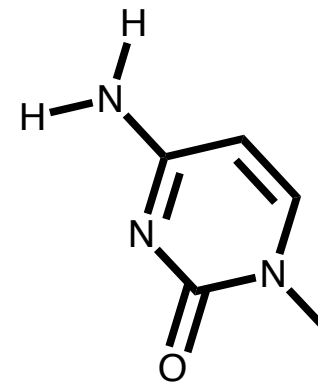
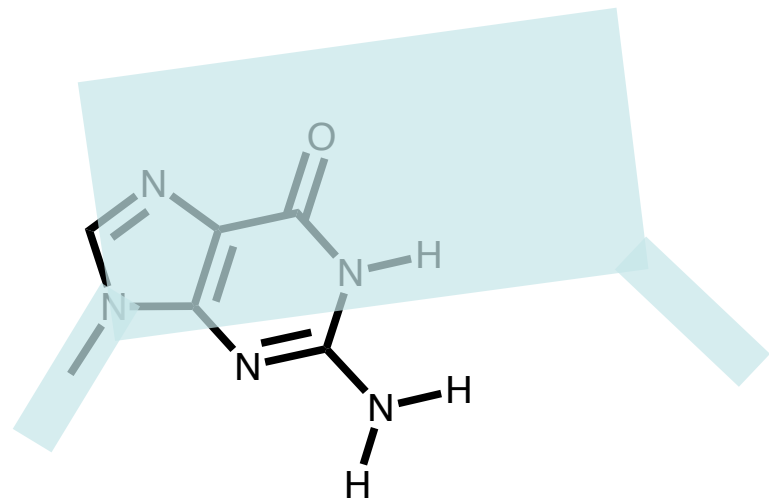
Base Pairs



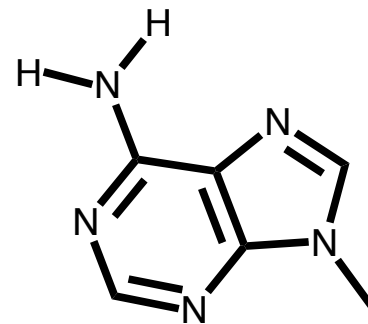
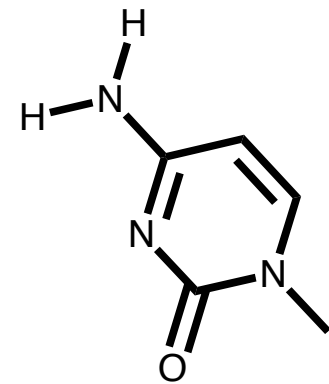
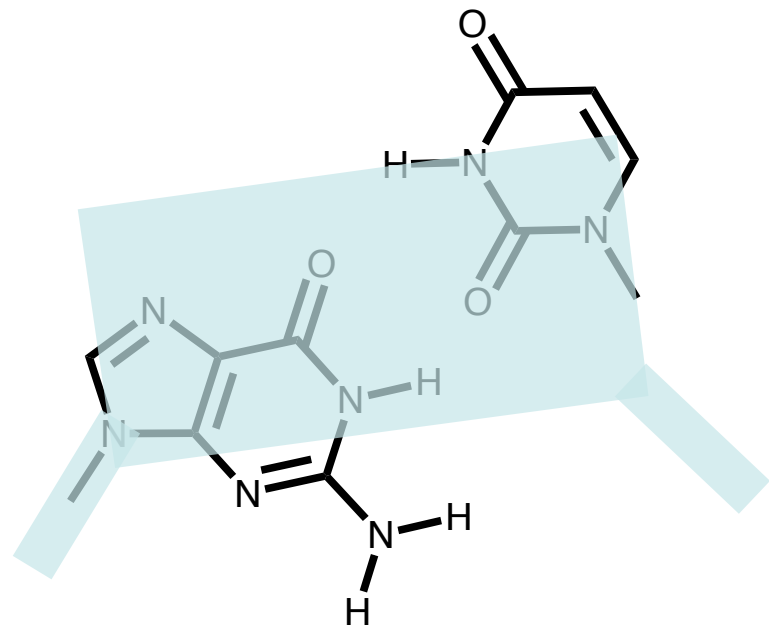
Base Pairs



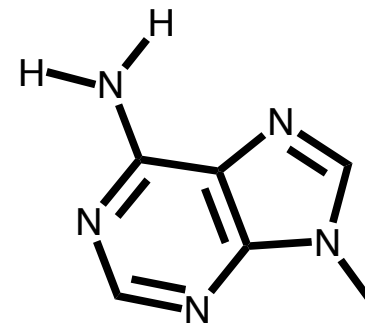
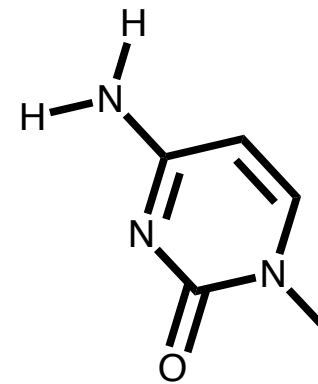
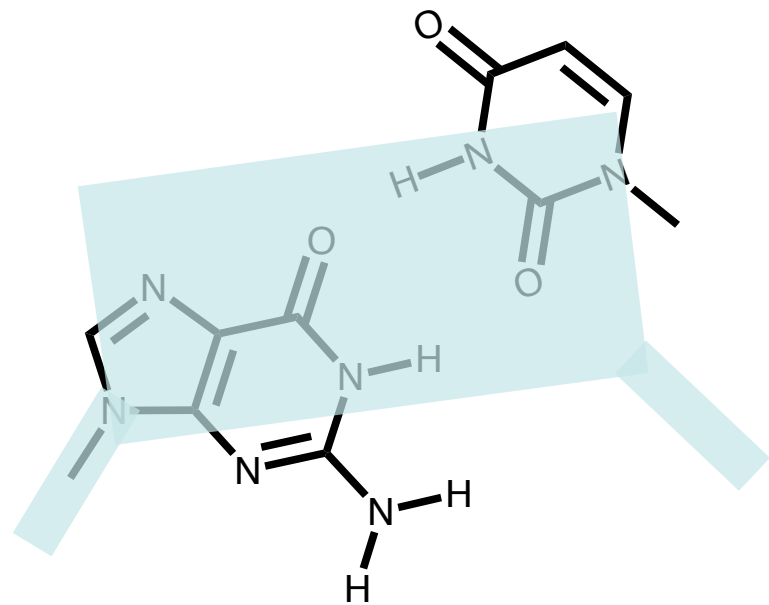
Base Pairs



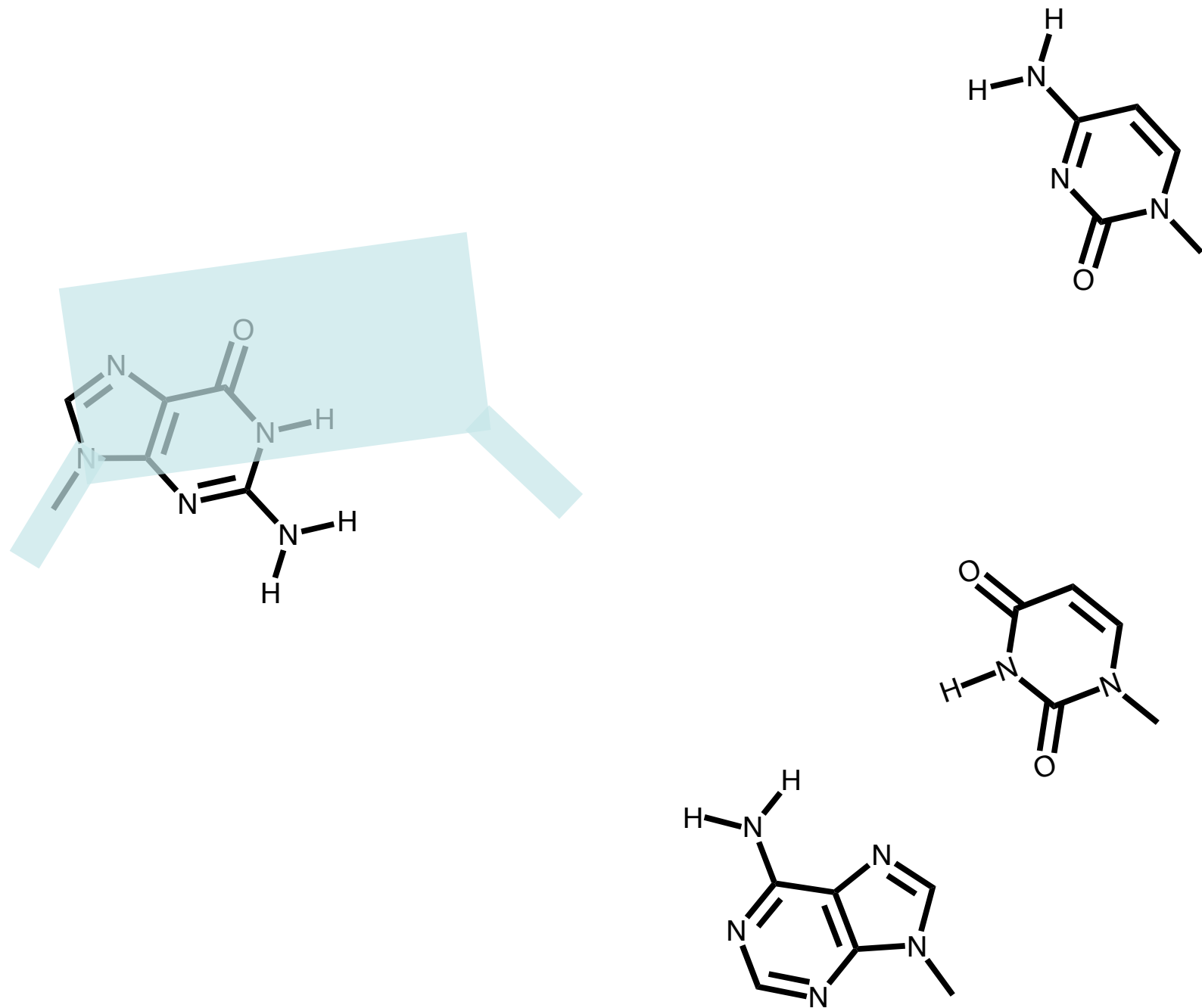
Base Pairs



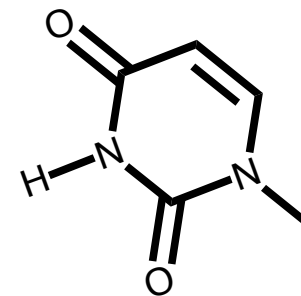
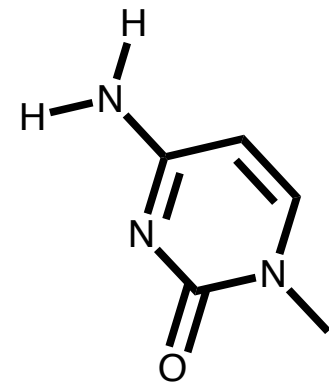
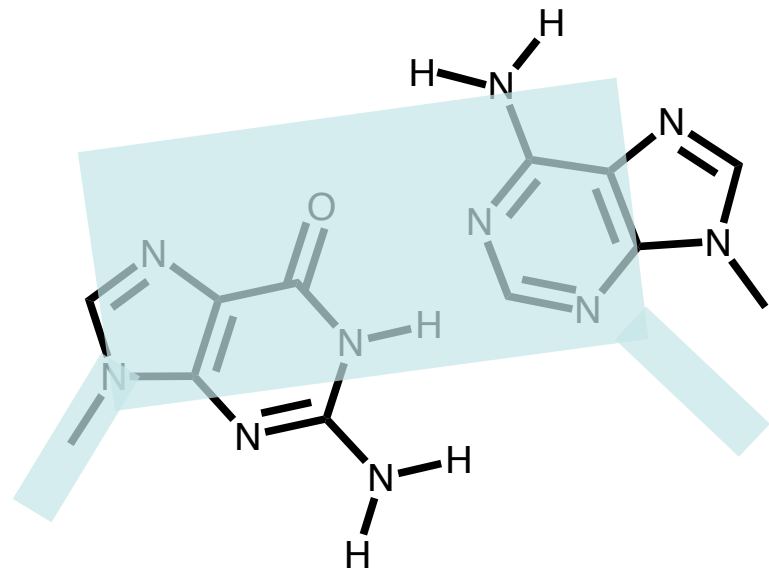
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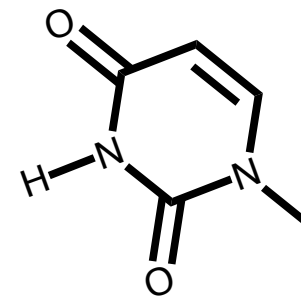
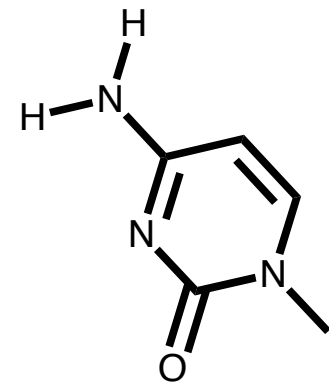
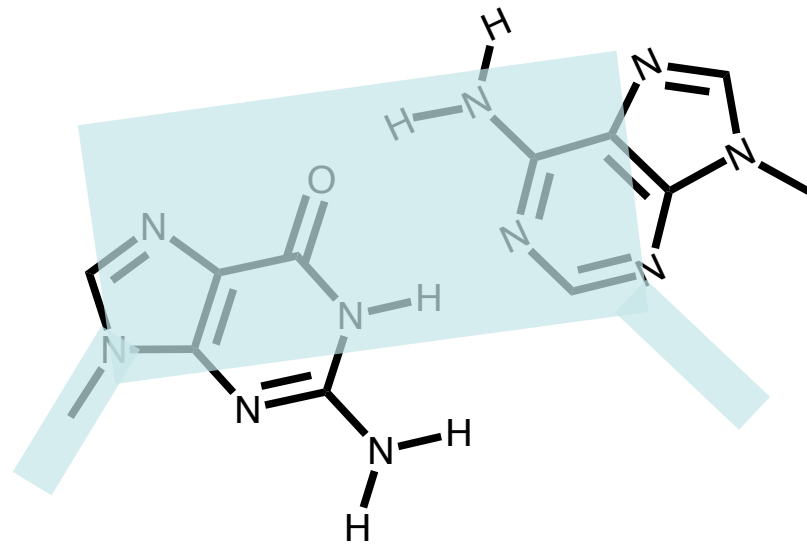
Base Pairs



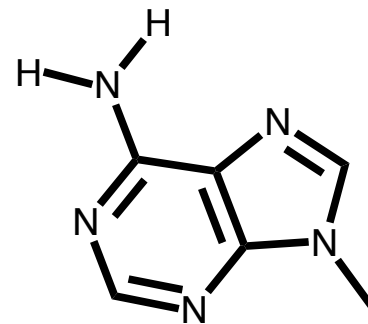
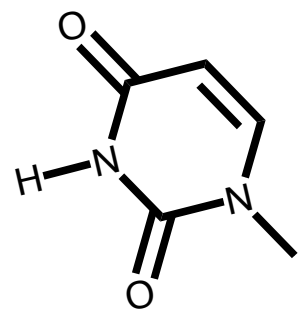
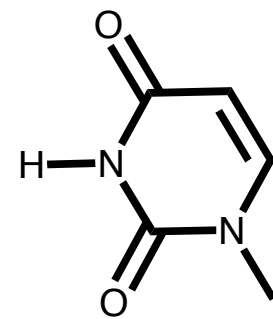
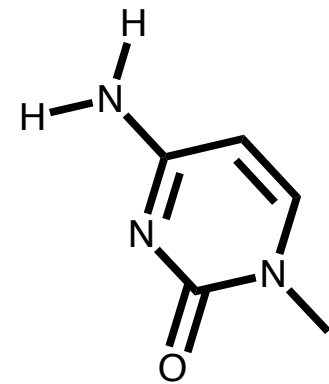
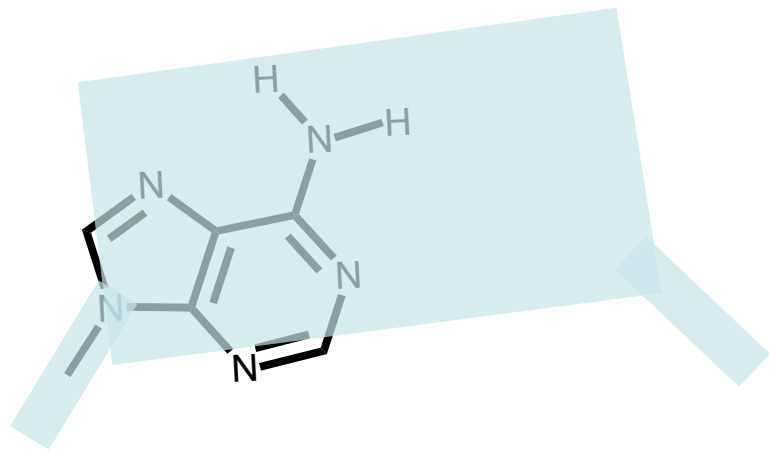
Base Pairs



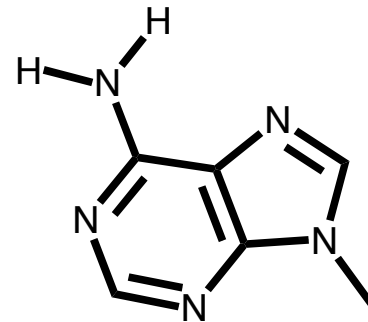
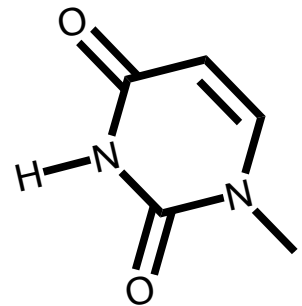
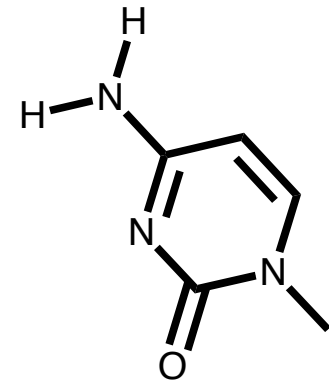
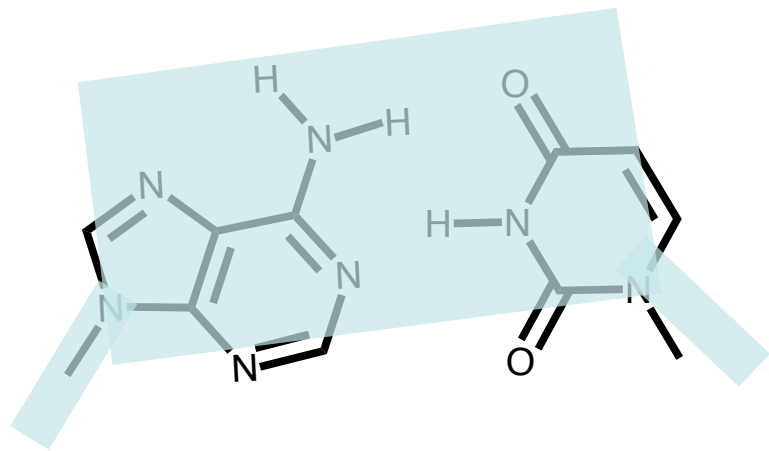
Base Pairs



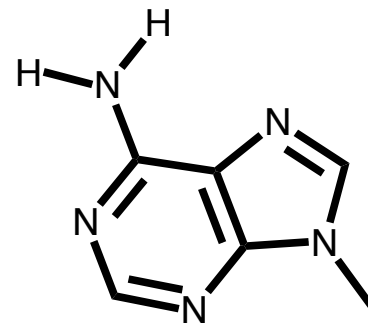
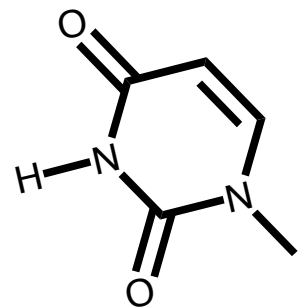
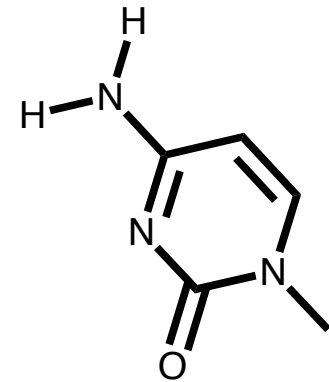
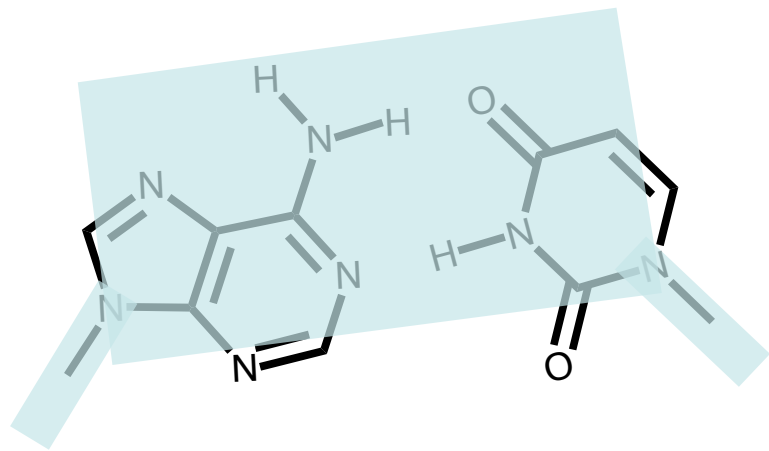
Base Pairs



Base Pairs



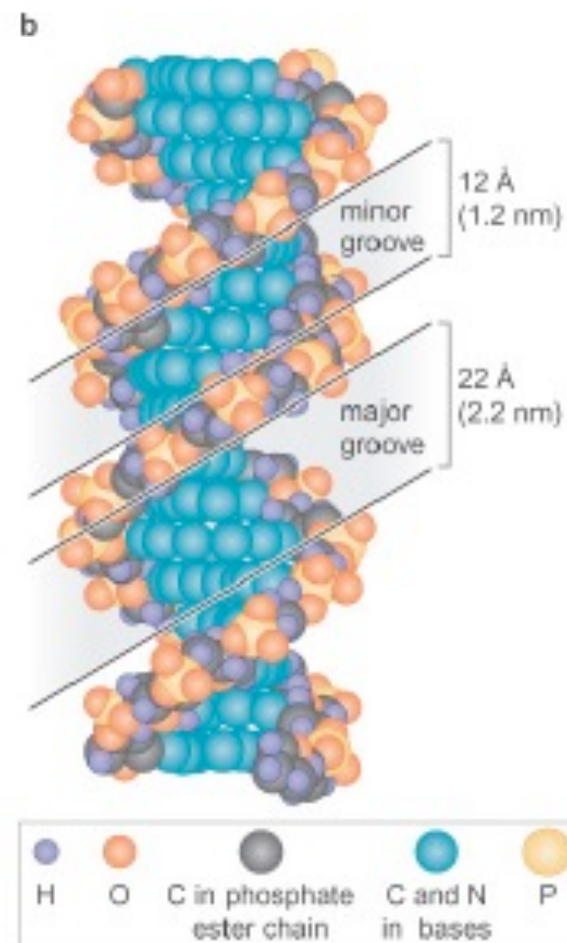
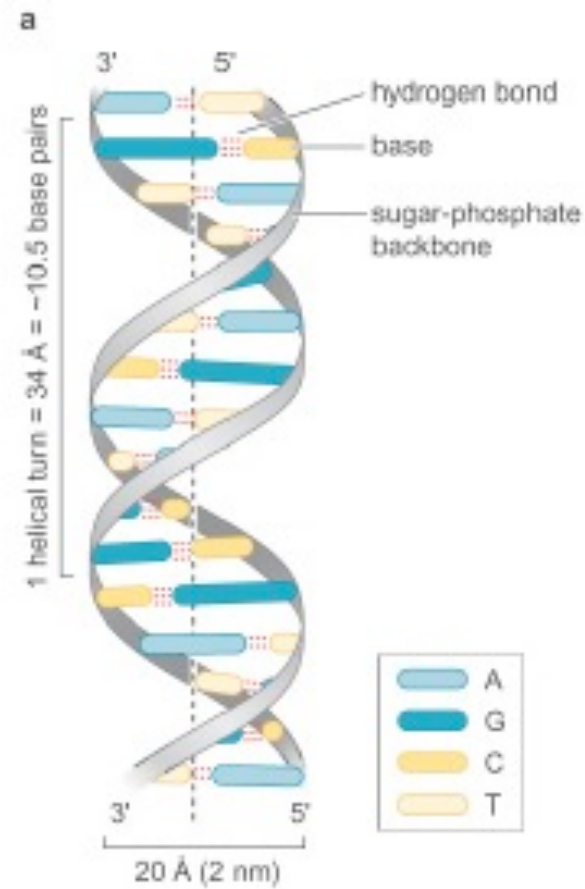
Base Pairs



B-form in 3D

A-form in 3D

More in 3D



DifluoroToluene

The difluorotoluene debate—a decade later

Eric T. Kool* and Herman O. Sintim

Chem. Commun., 2006, 3665–3675

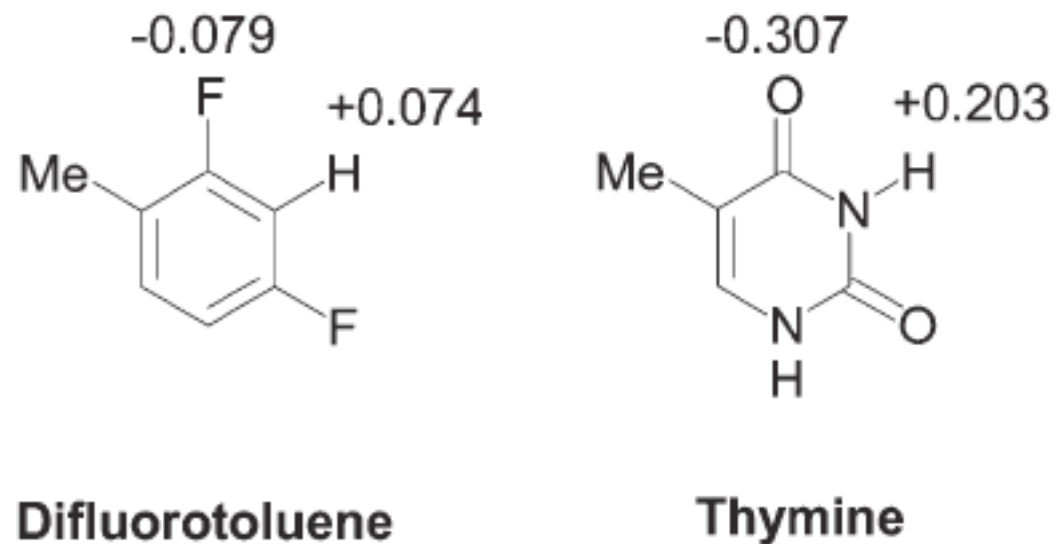


Table 2 Pairing stabilities of 12 bp DNA duplexes containing difluorotoluene or thymine paired opposite natural bases

5'-C T T T T C X T T C T T

3'-G A A A A G Y A A G A A

X.Y	$T_m/^\circ\text{C}$	$\Delta G^\circ_{37}/\text{kcal mol}^{-1}$	X.Y	$T_m/^\circ\text{C}$	$\Delta G^\circ_{37}/\text{kcal mol}^{-1}$
T.A	42.4	-9.7	A.T	42.4	-9.6
T.G	33.3	-7.5	G.T	34.7	-7.9
T.C	29.5	-6.6	C.T	25.7	-6.0
T.T	29.1	-6.8	T.T	29.1	-6.8
F.A	26.2	-6.2	A.F	27.5	-6.6
F.G	23.6	-6.0	G.F	25.1	-6.4
F.C	23.7	-5.8	C.F	25.1	-6.4
F.T	24.0	-5.9	T.F	24.8	-6.4

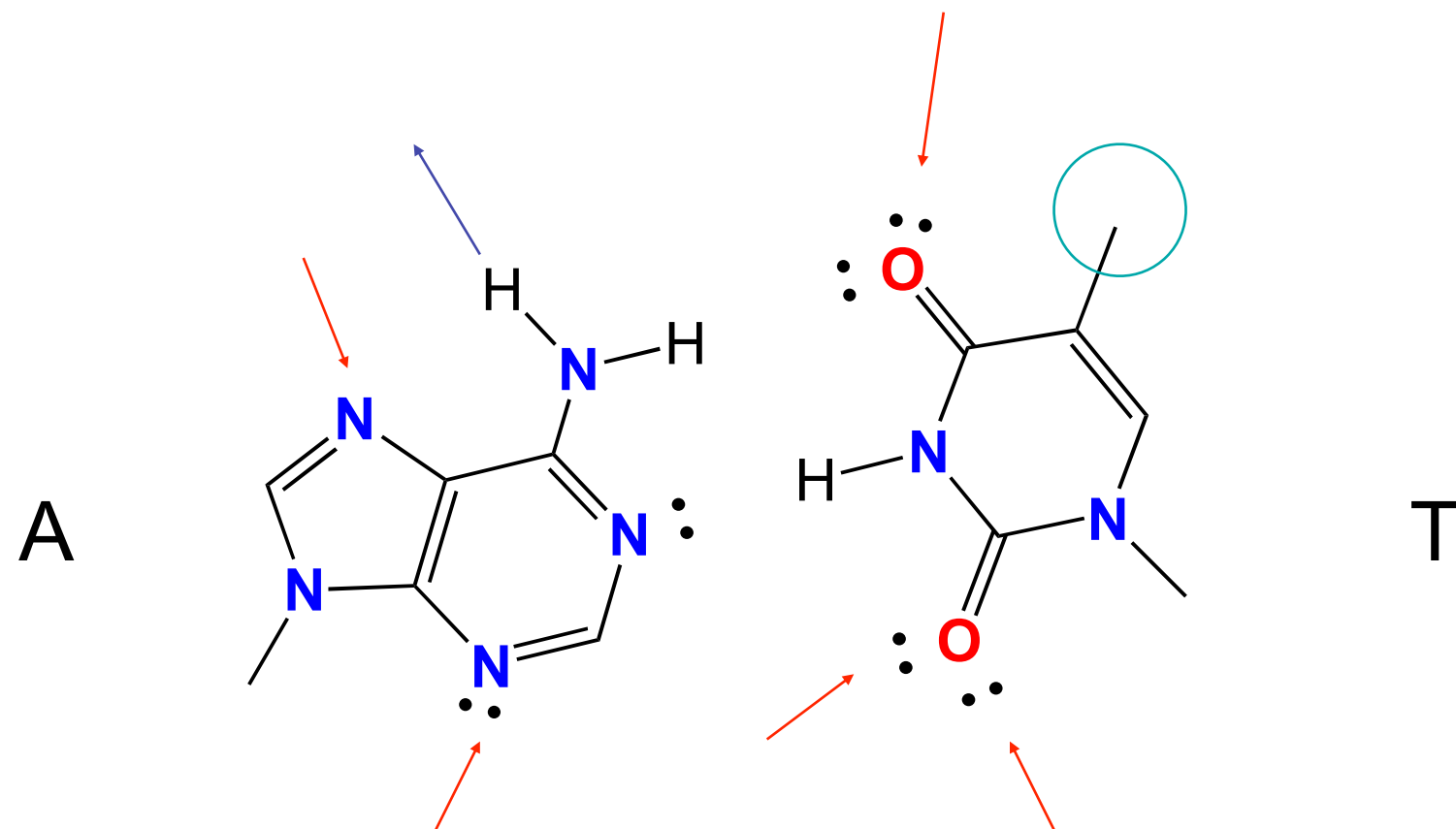
^a Conditions: 1 M NaCl, 10 mM phosphate buffer (pH 7.0) with 0.1 mM EDTA, 2.5 μM each strand, monitored at 260 nm. Error in T_m is estimated at $\pm 0.5^\circ\text{C}$, and in free energy, $\pm 10\%$.

- Difluorotoluene pairs poorly with Adenine, in contrast to the normal partner Thymine
- DNA polymerase incorporates T opposite A
- DNA polymerase does not incorporate G, C, or A opposite A
- DNA polymerase incorporates Difluorotoluene opposite A

Wednesday

Why is the major groove so good?

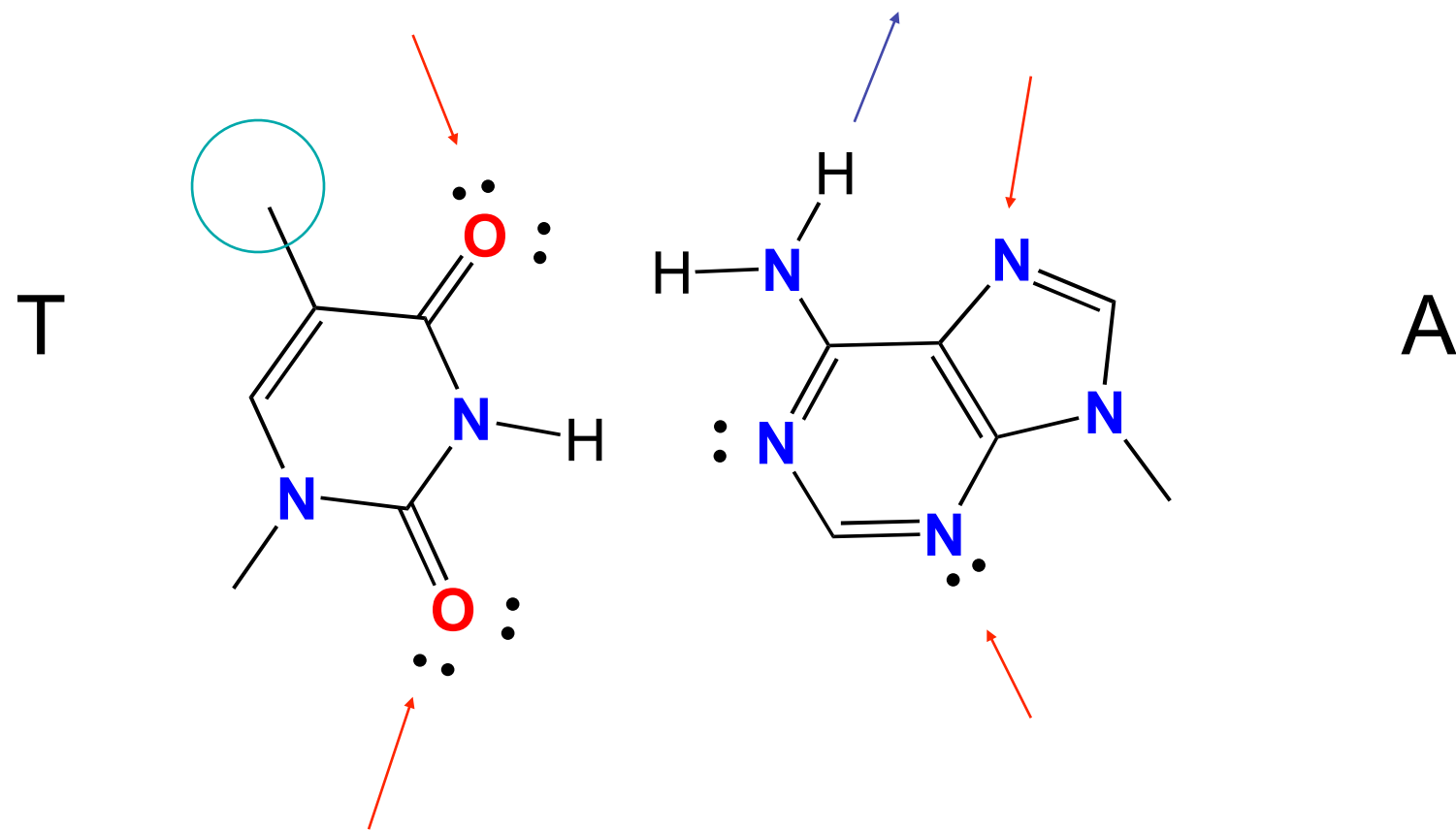
Major Groove



Minor Groove

Why is the major groove so good?

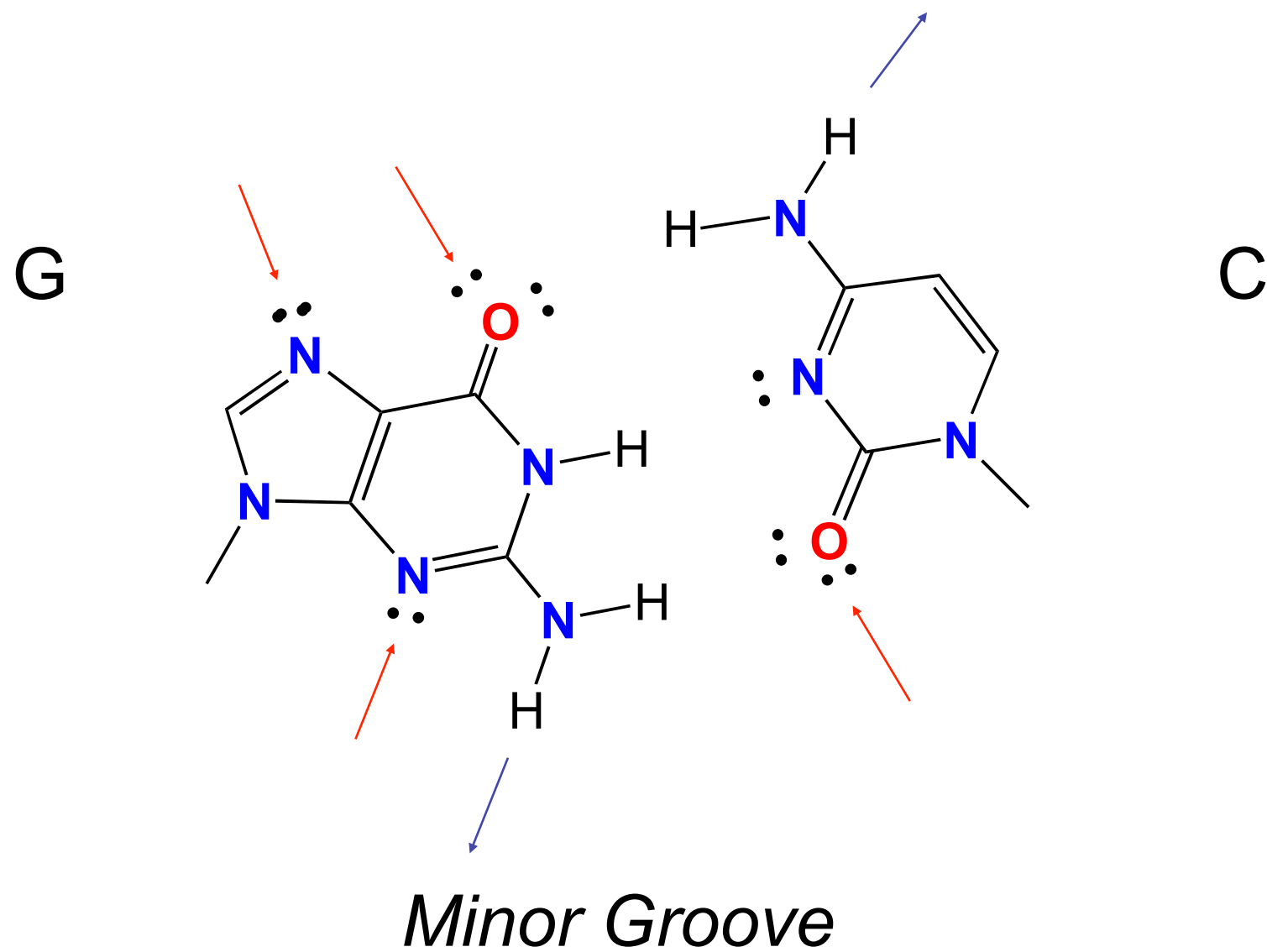
Major Groove



Minor Groove

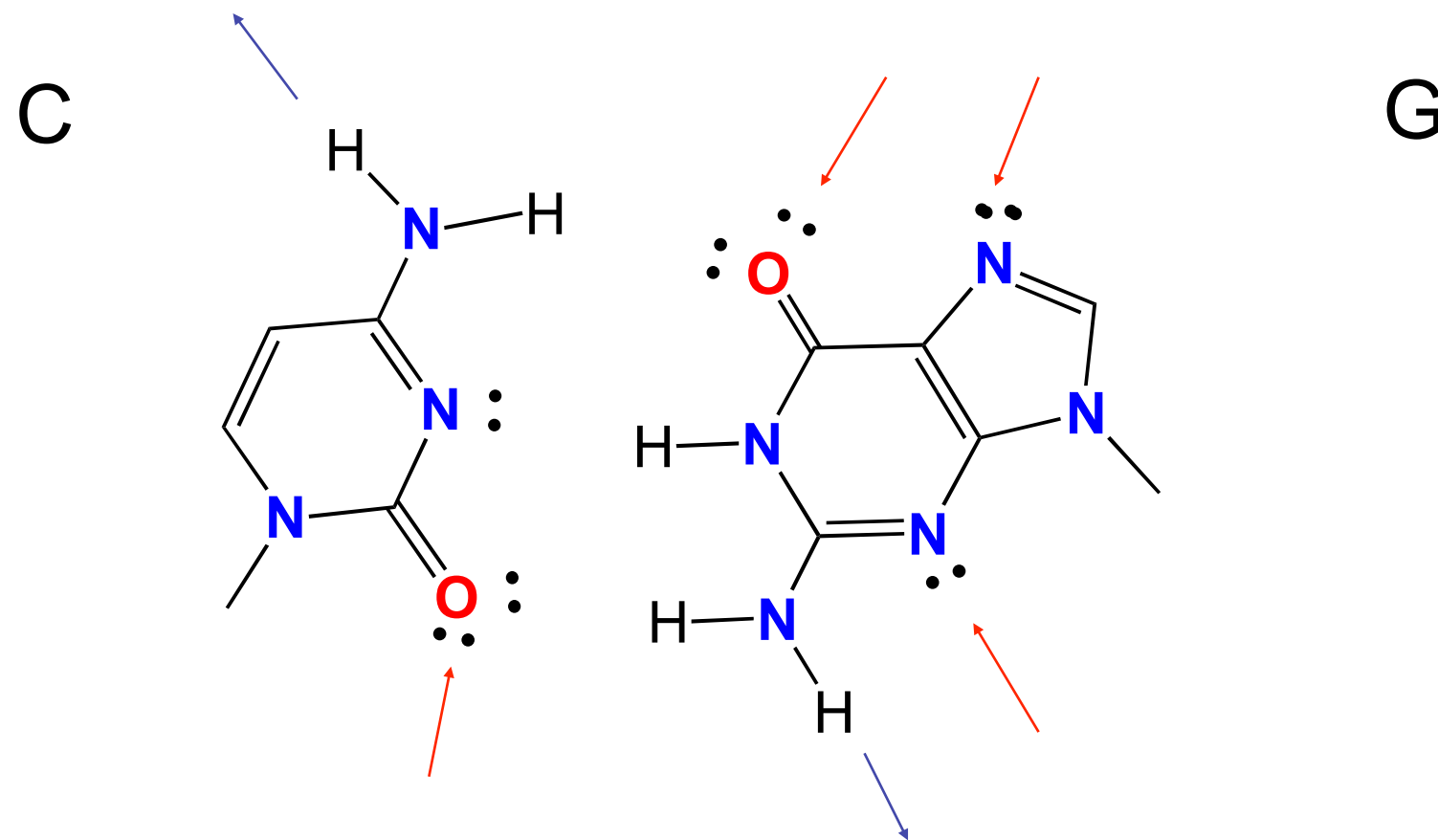
Why is the major groove so good?

Major Groove



Why is the major groove so good?

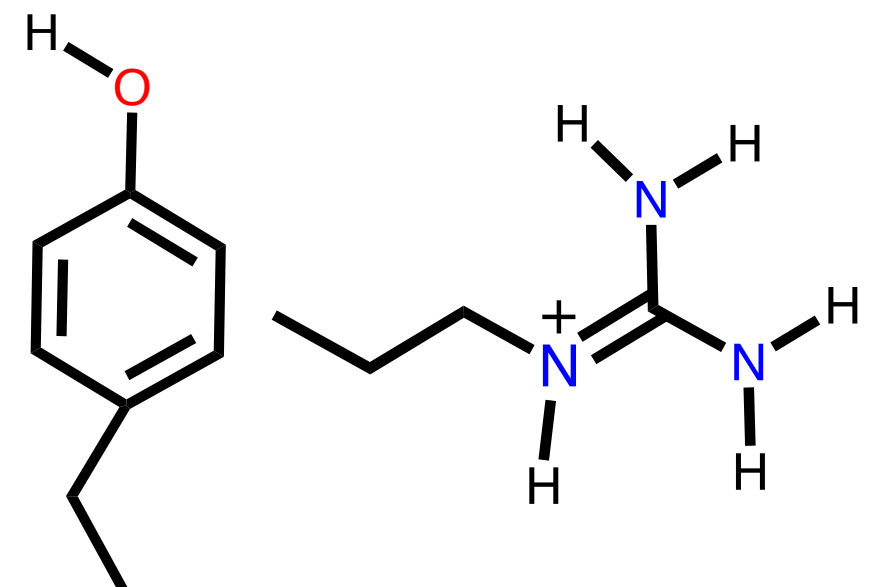
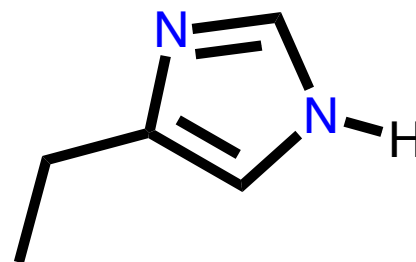
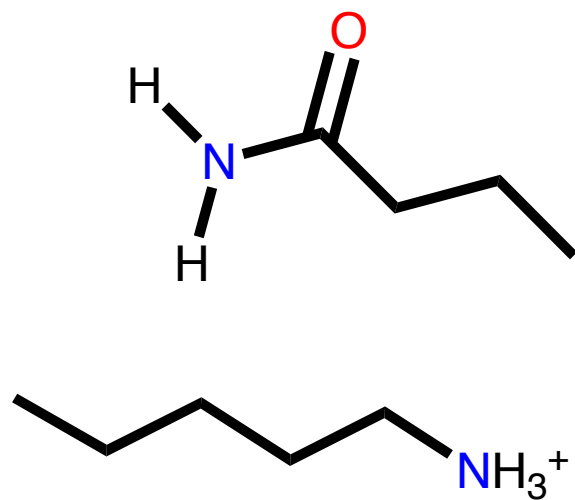
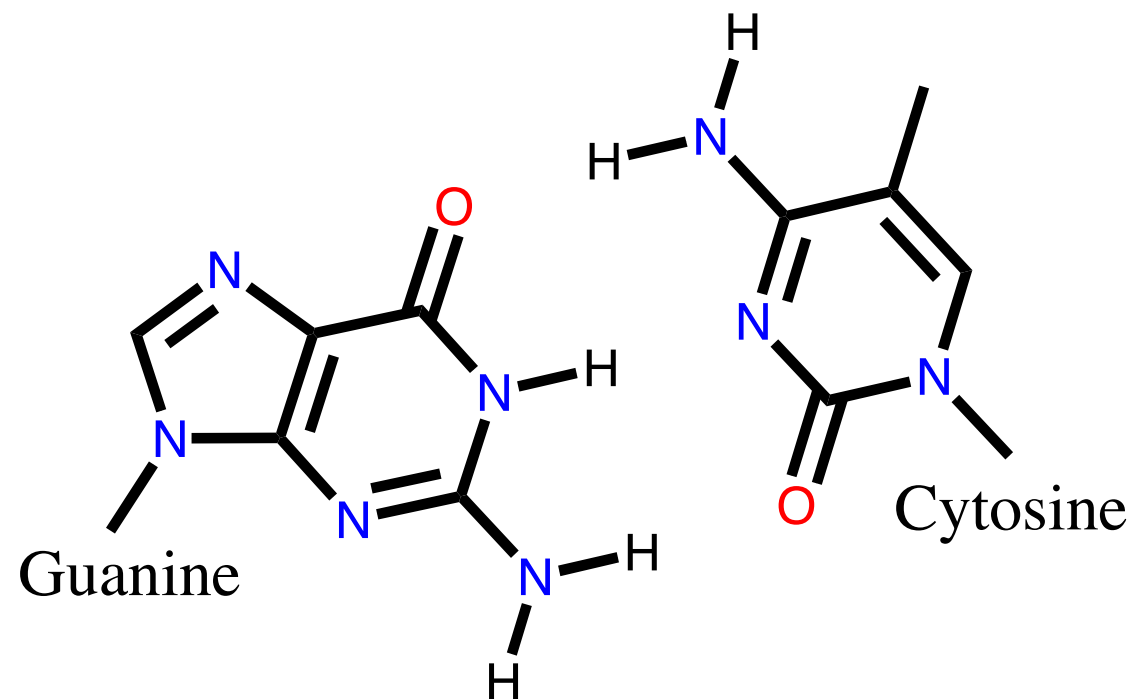
Major Groove



Minor Groove

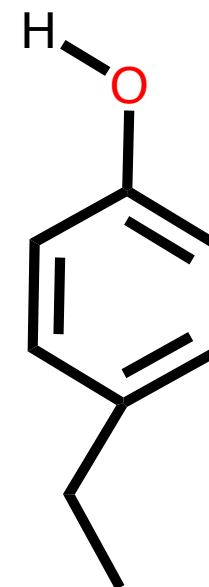
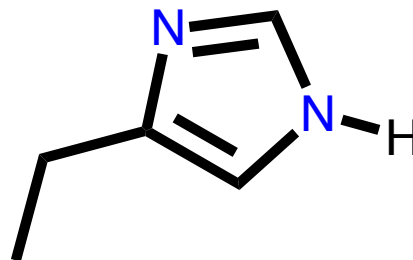
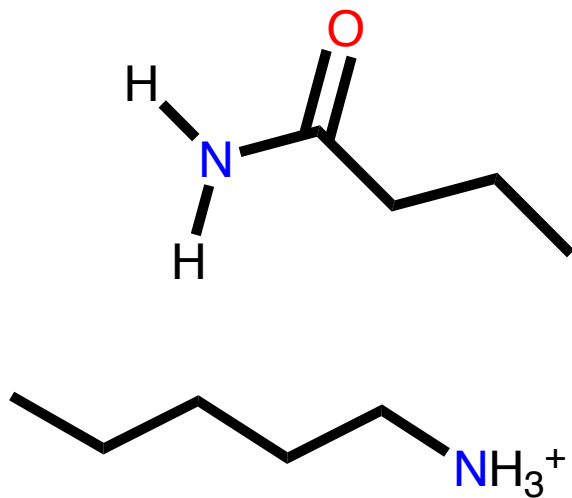
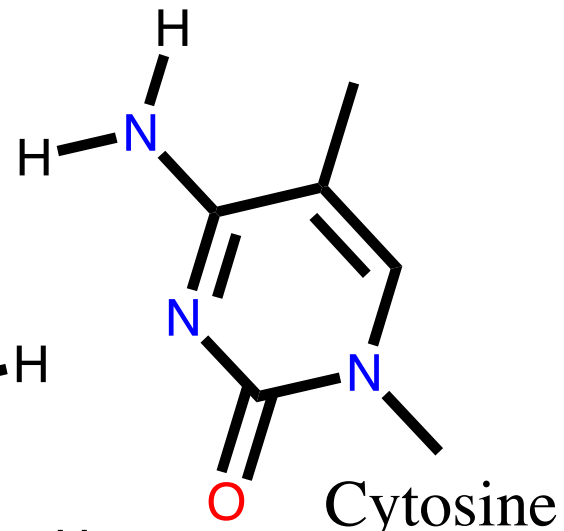
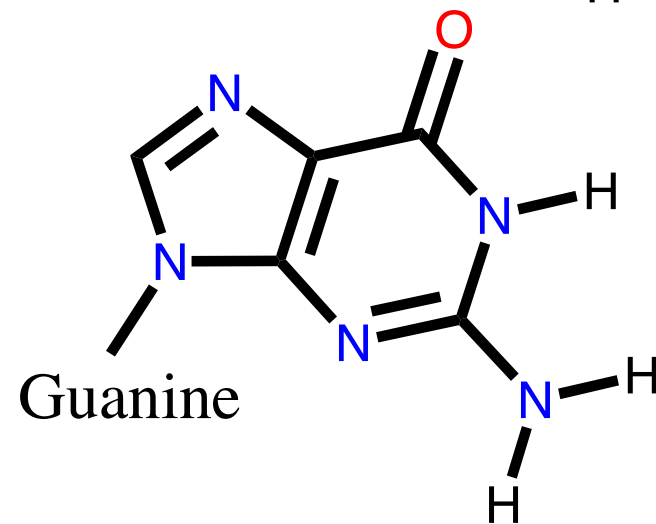
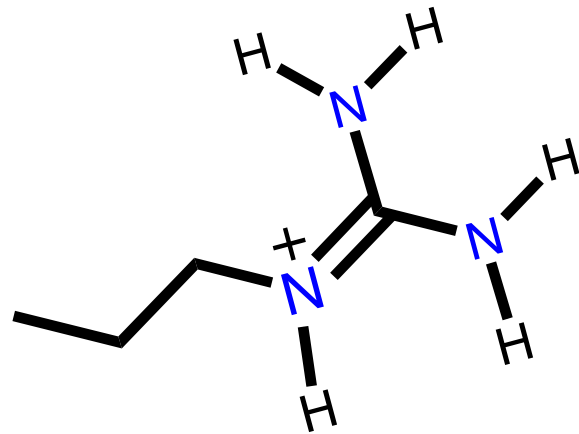
Sequence-specific binding by proteins

Find “two-fer” interactions



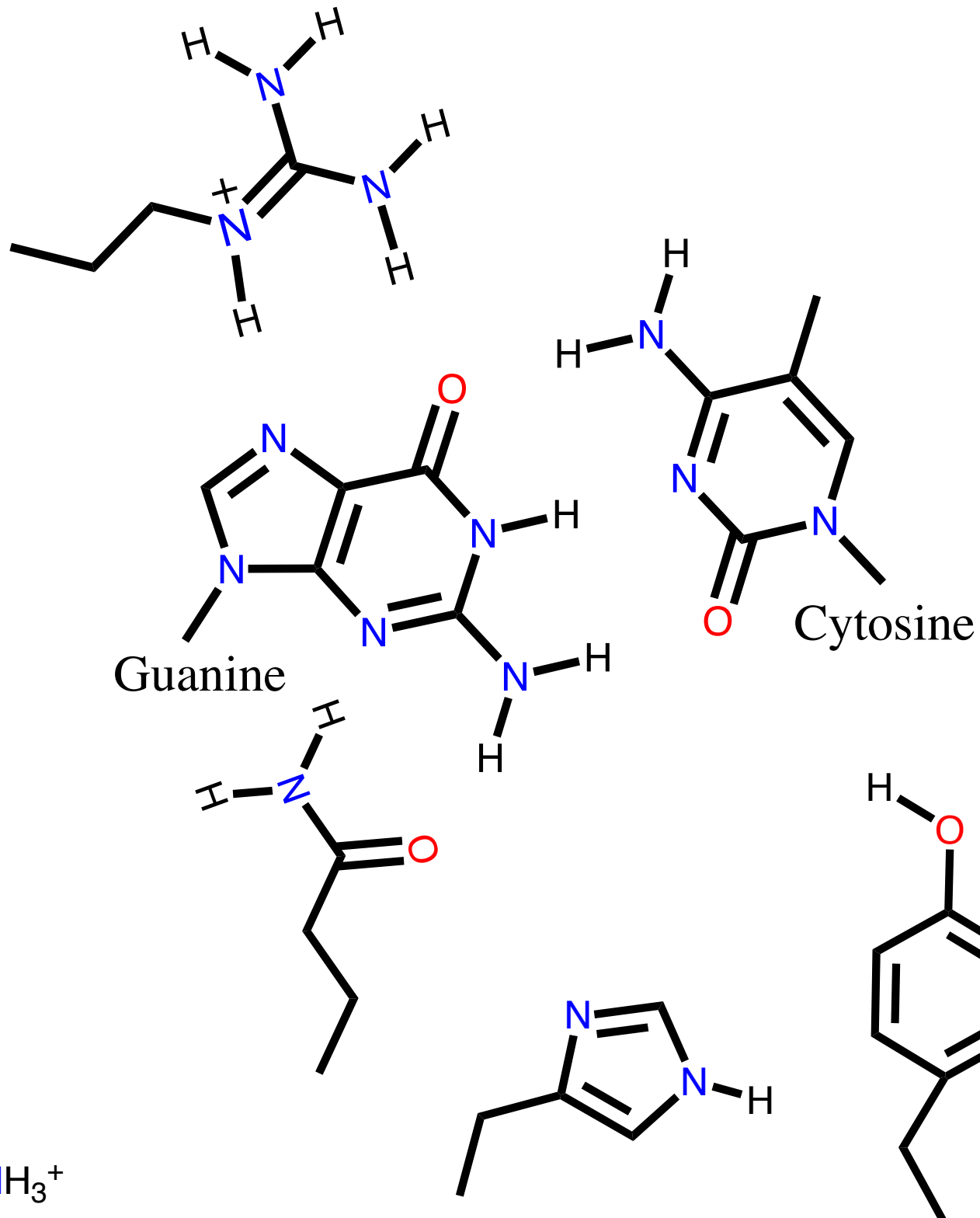
Sequence-specific binding by proteins

Find “two-fer” interactions



Sequence-specific binding by proteins

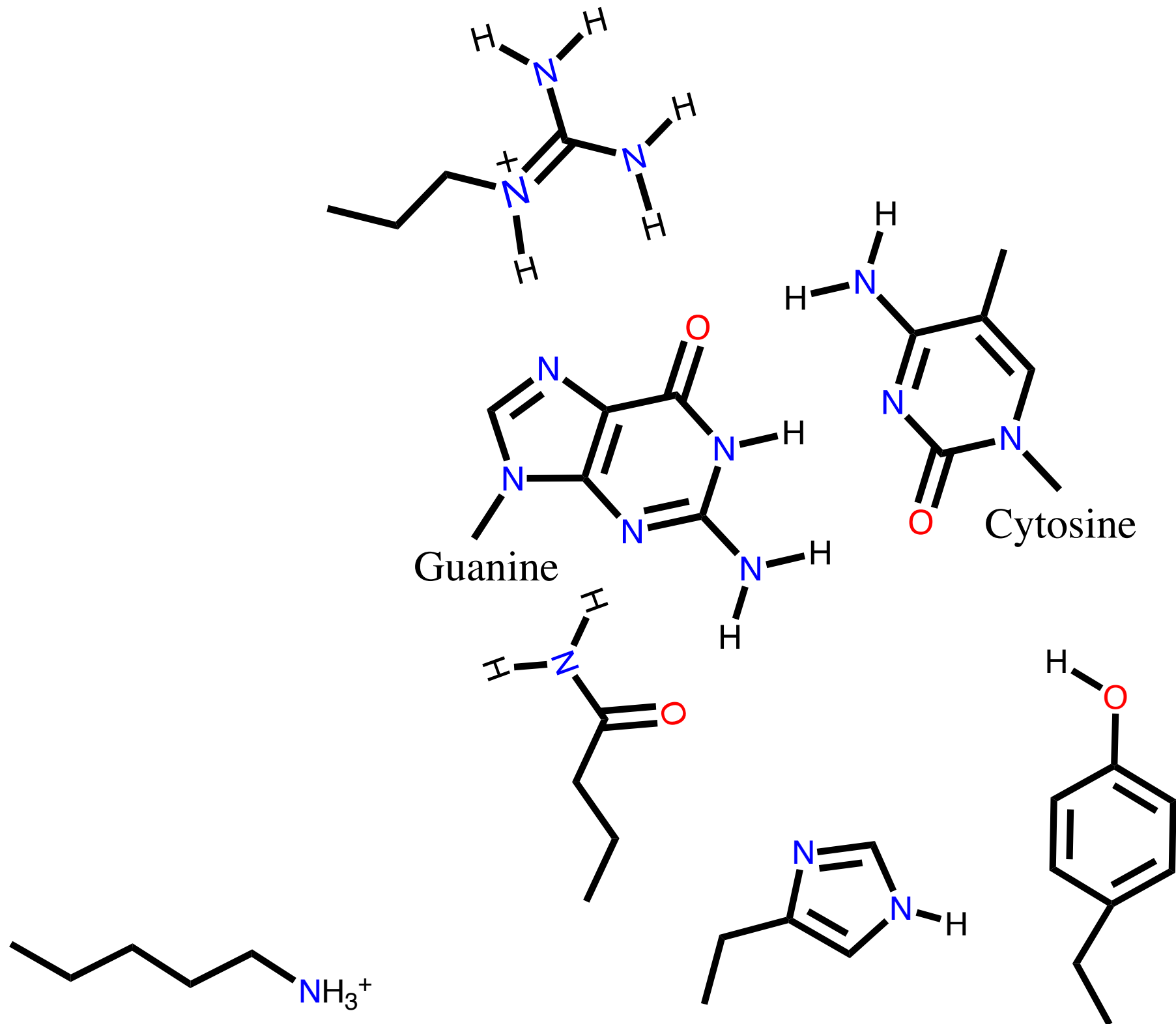
Find “two-fer” interactions



Sequence-specific binding by proteins

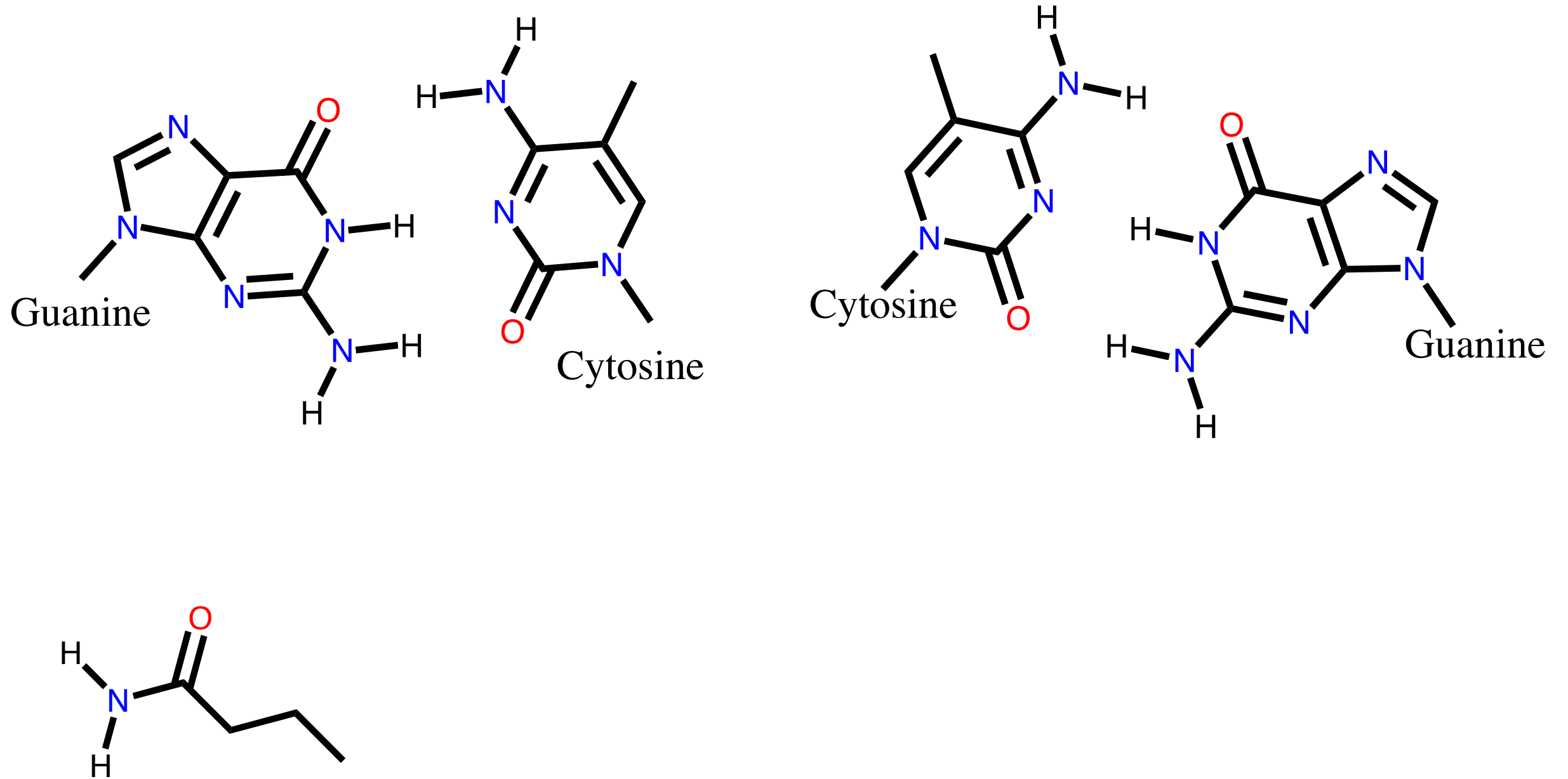
Find “two-fer” interactions

[See an example](#)



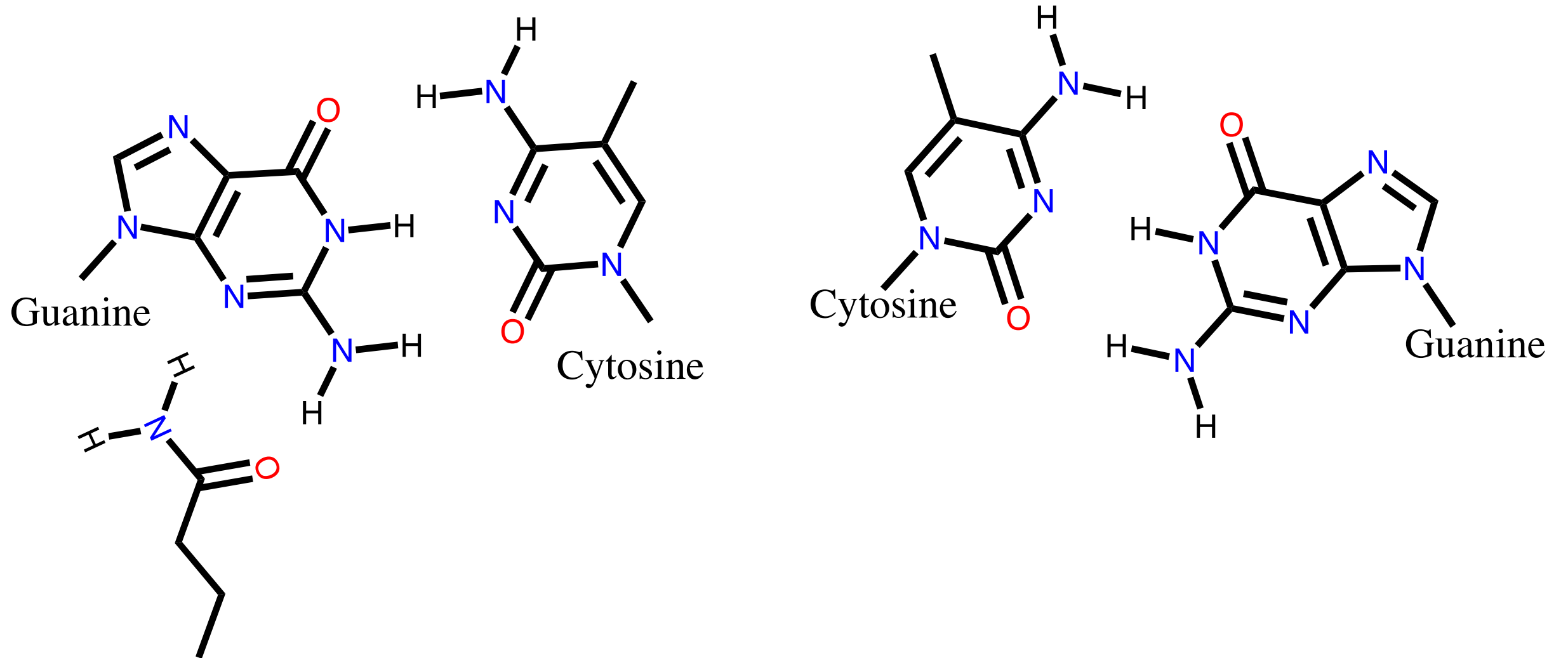
Sequence-specific binding by proteins

Major groove is better



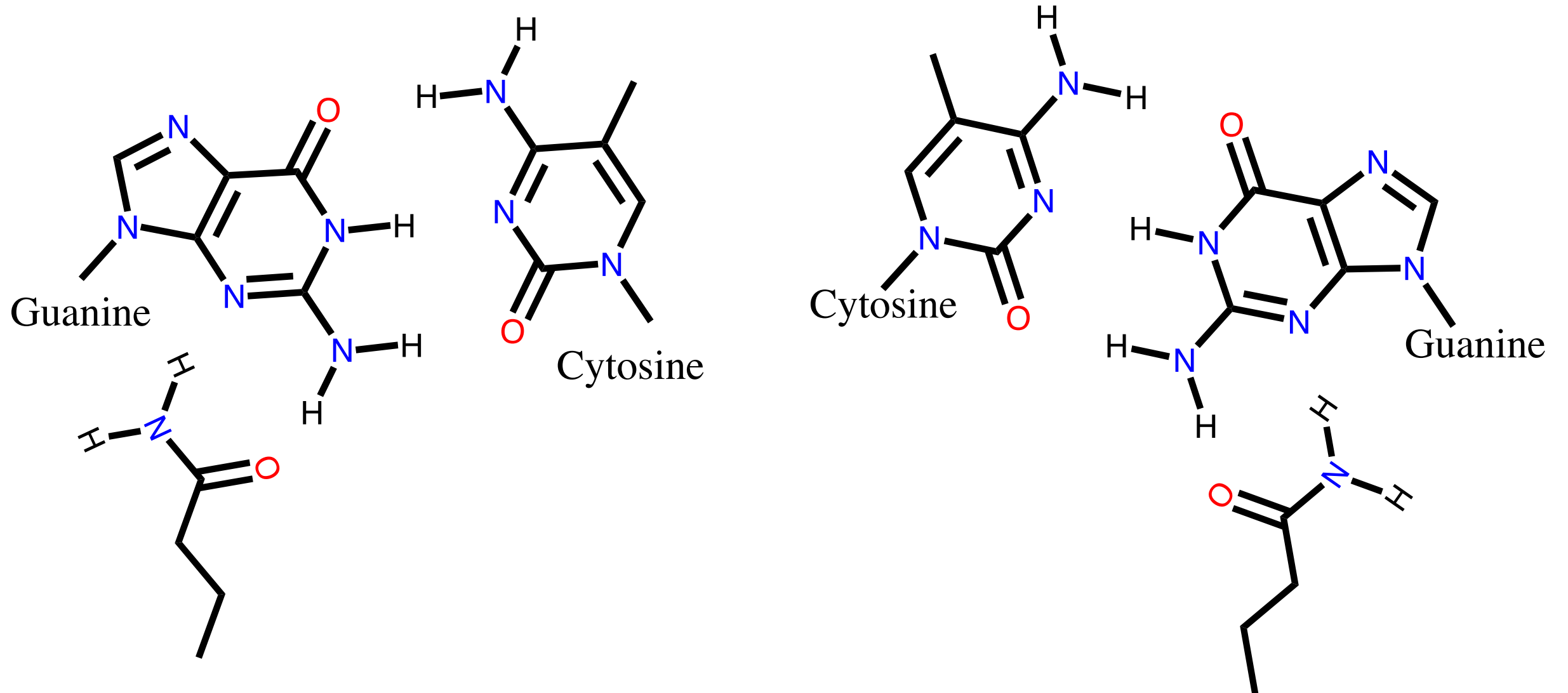
Sequence-specific binding by proteins

Major groove is better



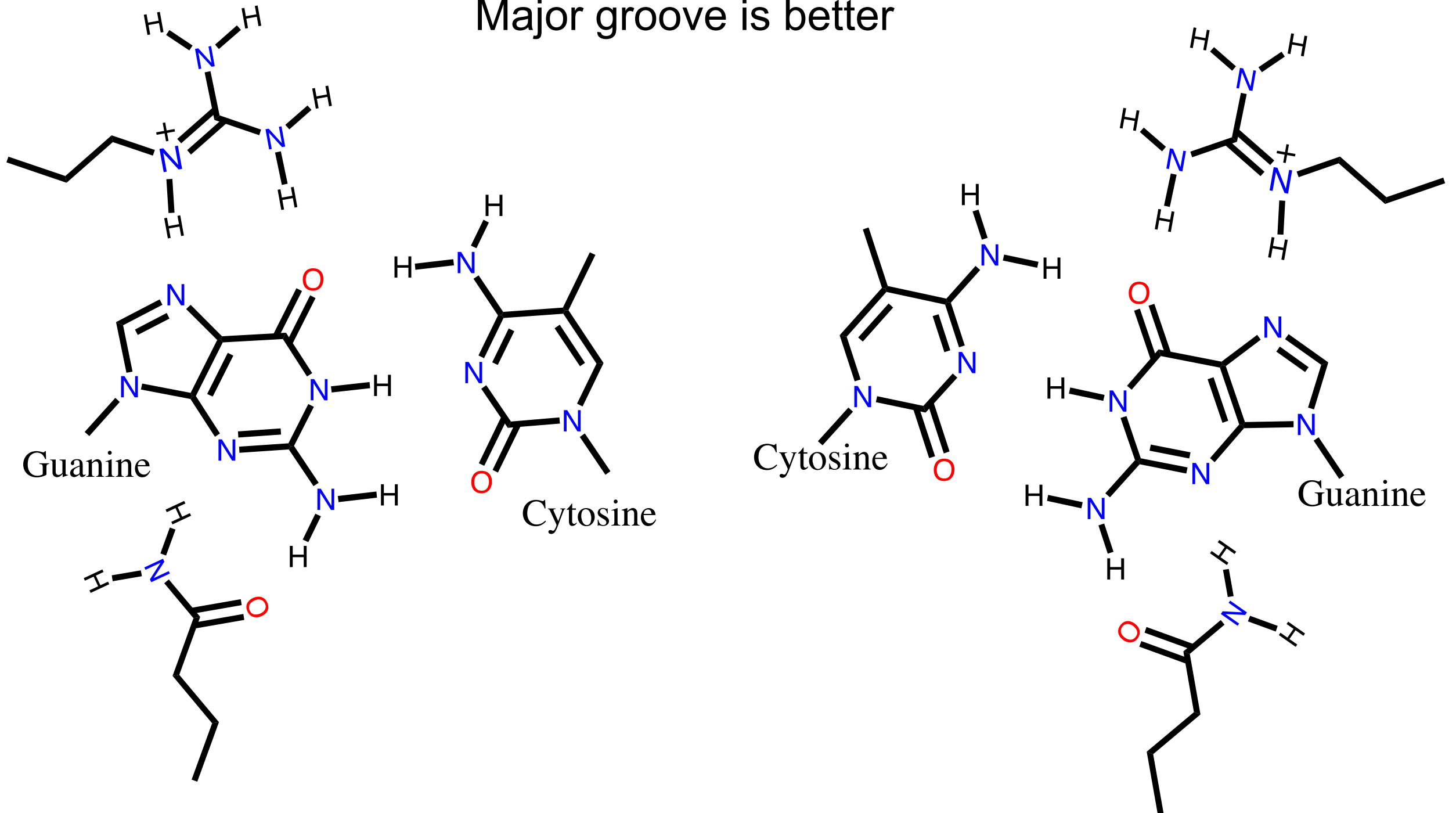
Sequence-specific binding by proteins

Major groove is better



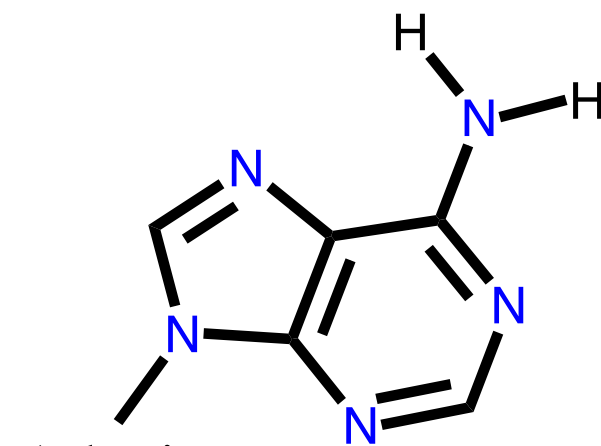
Sequence-specific binding by proteins

Major groove is better

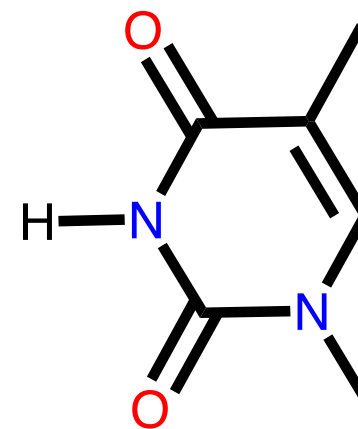


Sequence-specific binding by proteins

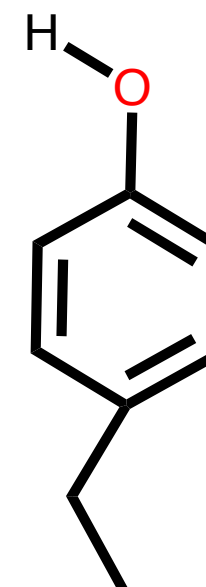
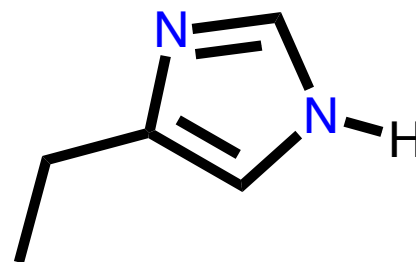
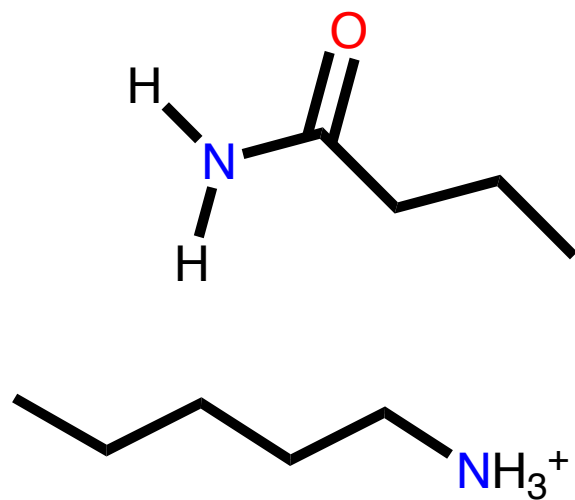
Find “two-fer” interactions



Adenine

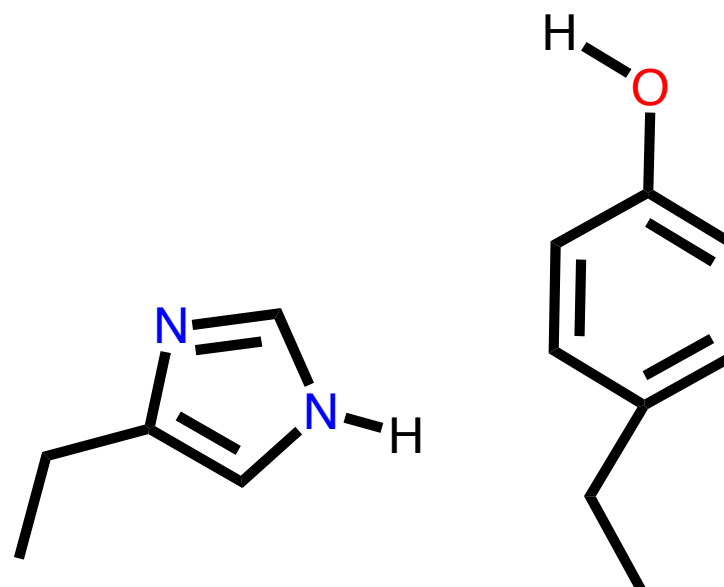
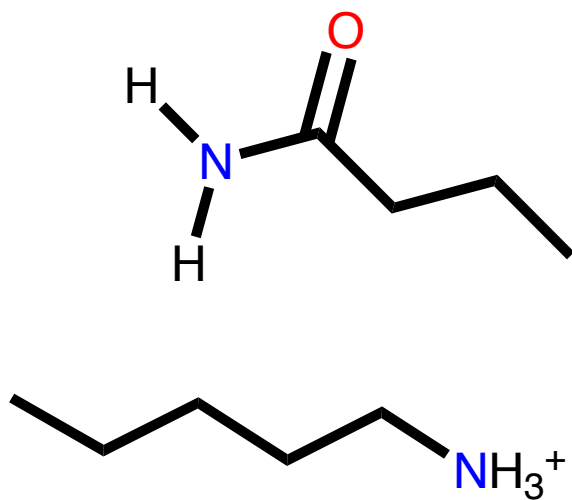
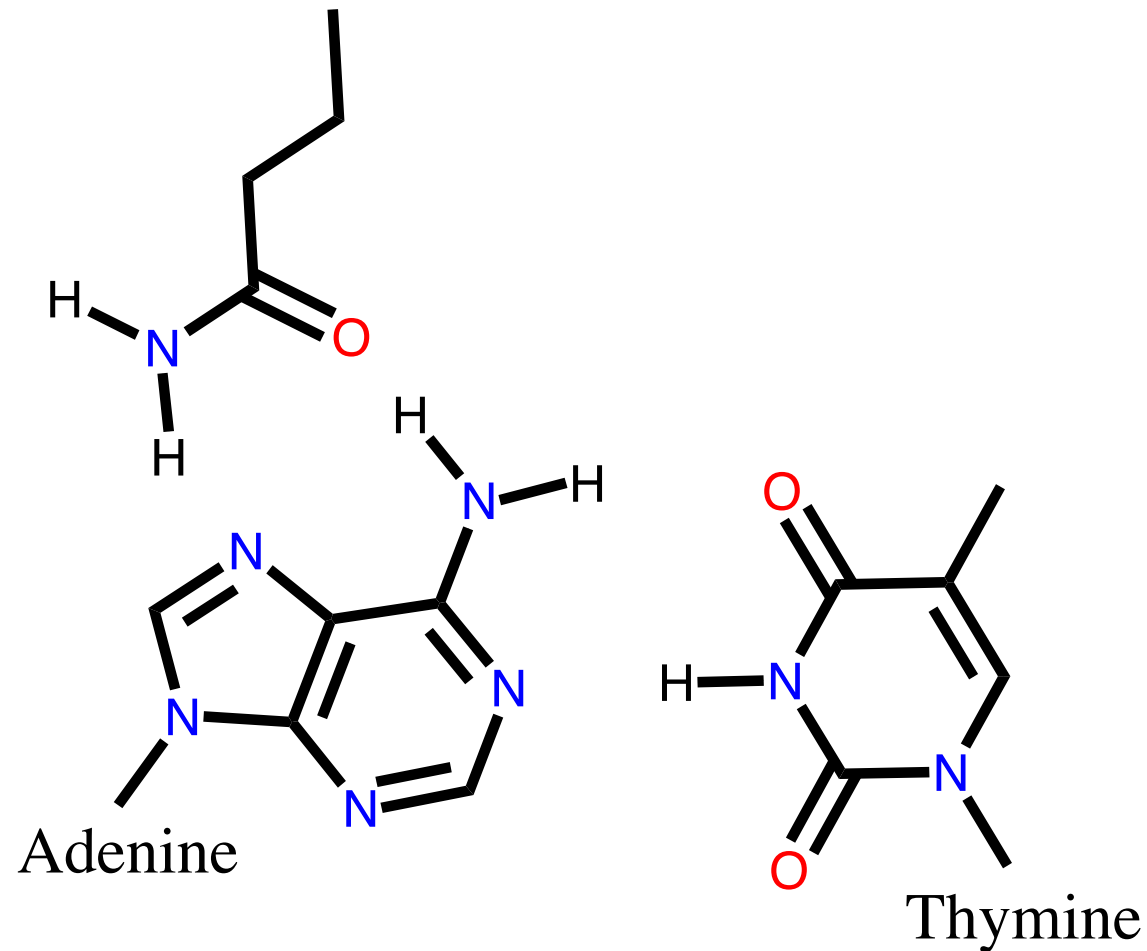


Thymine

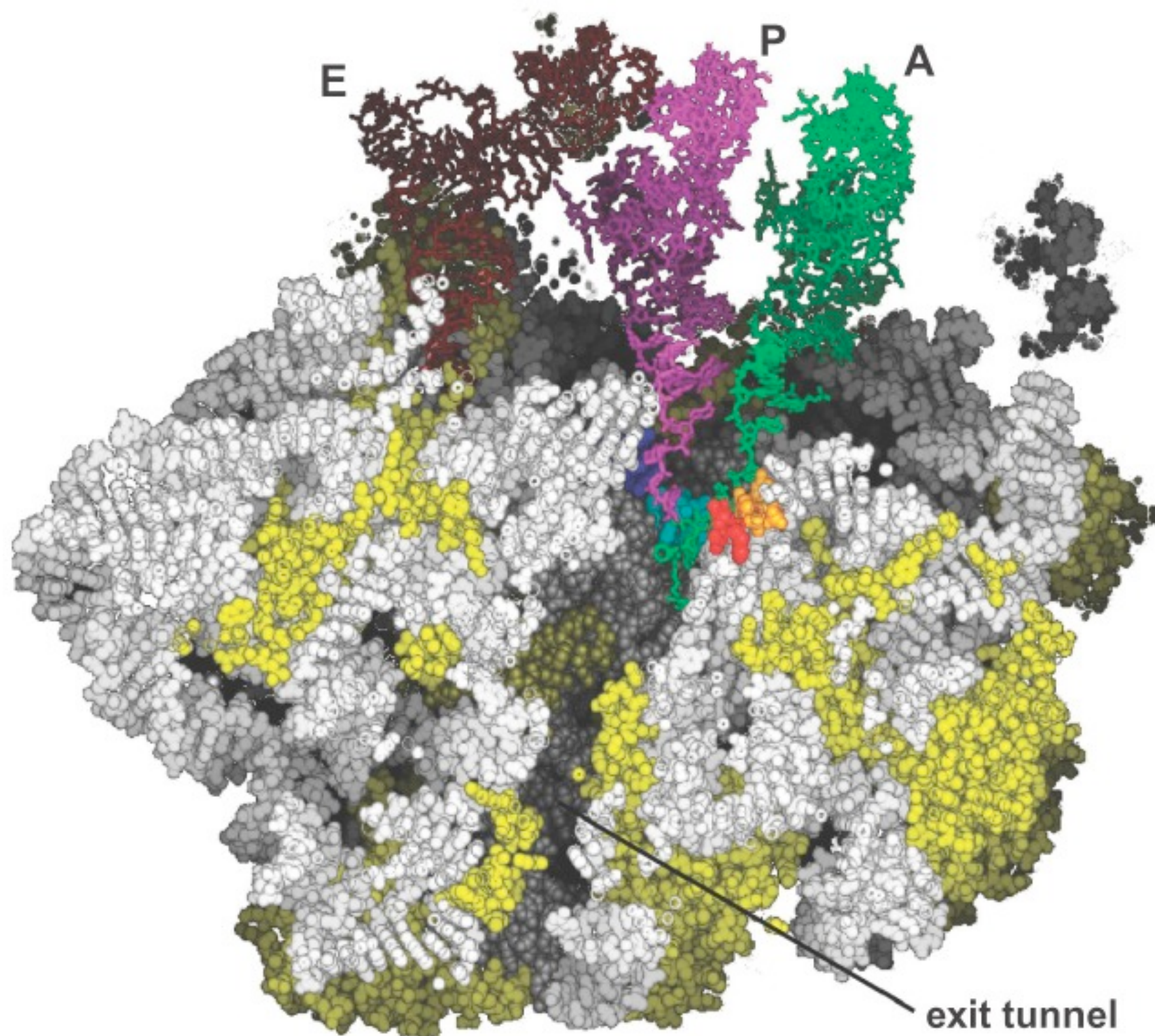


Sequence-specific binding by proteins

Find “two-fer” interactions



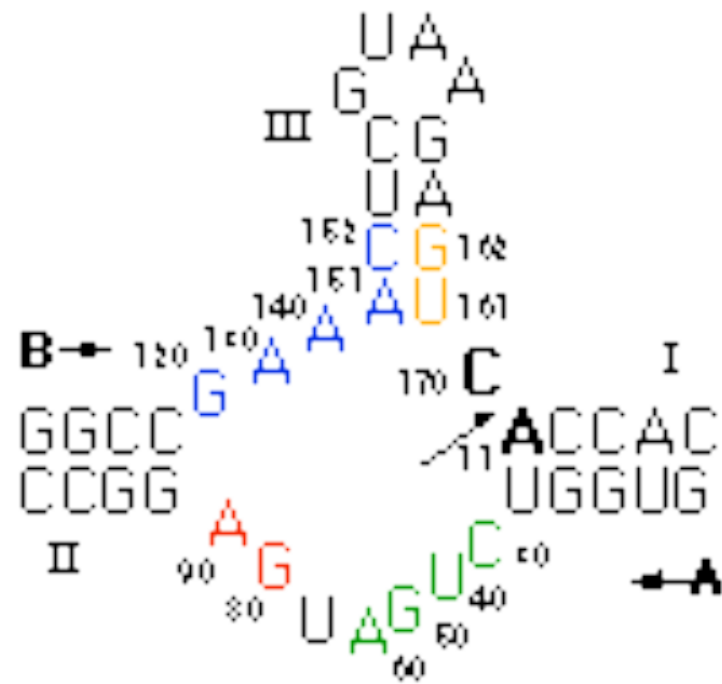
Ribosome



RNA structures are diverse!!

More in 3D

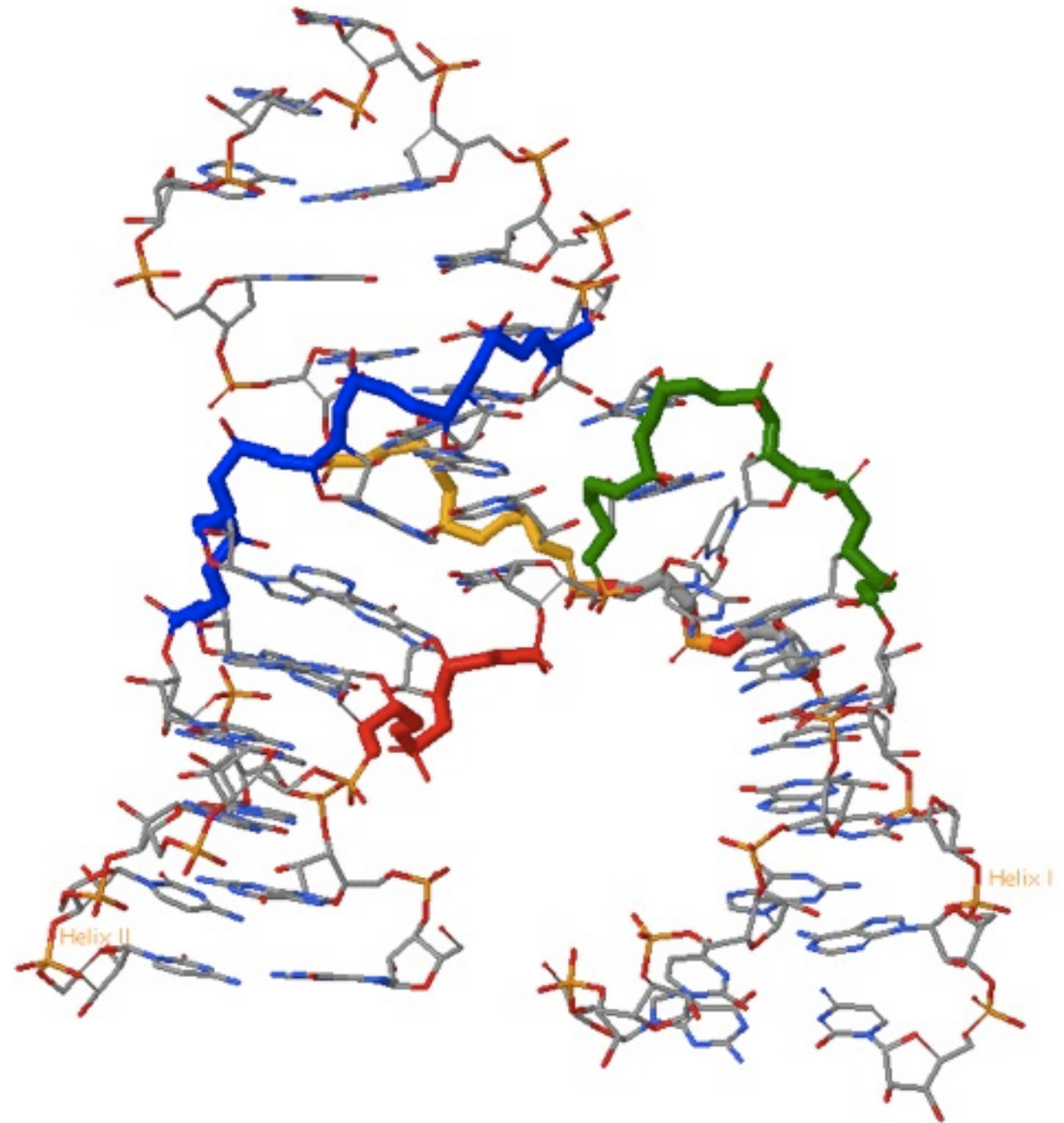
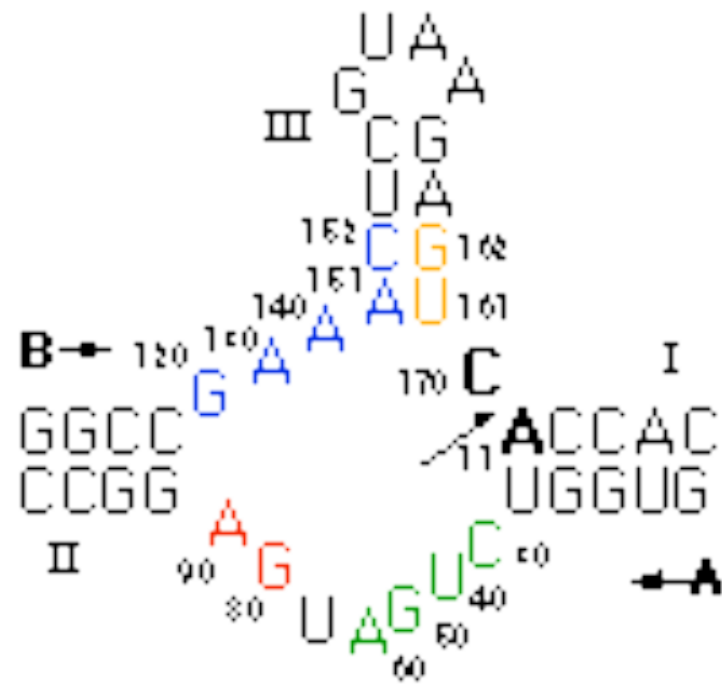
- “Hammerhead” ribozyme



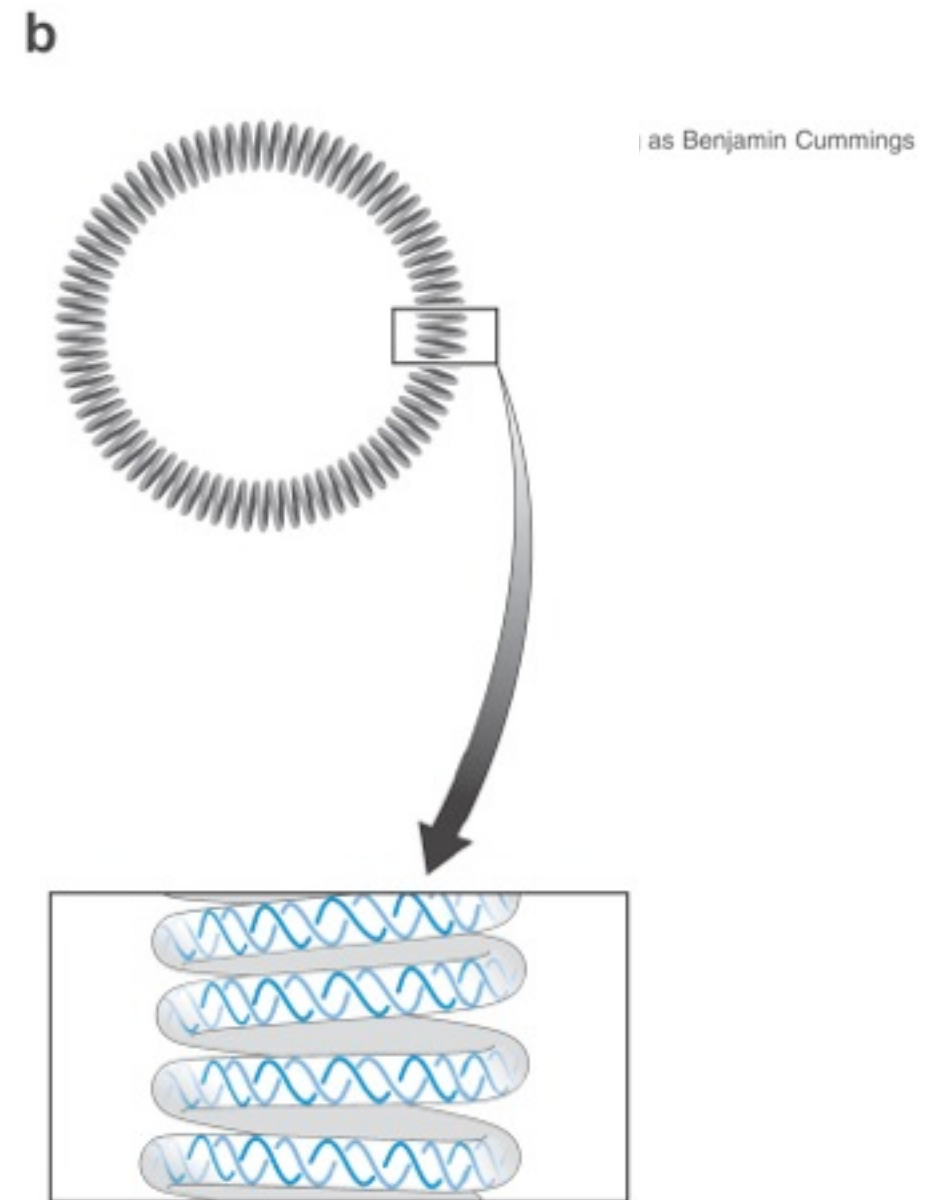
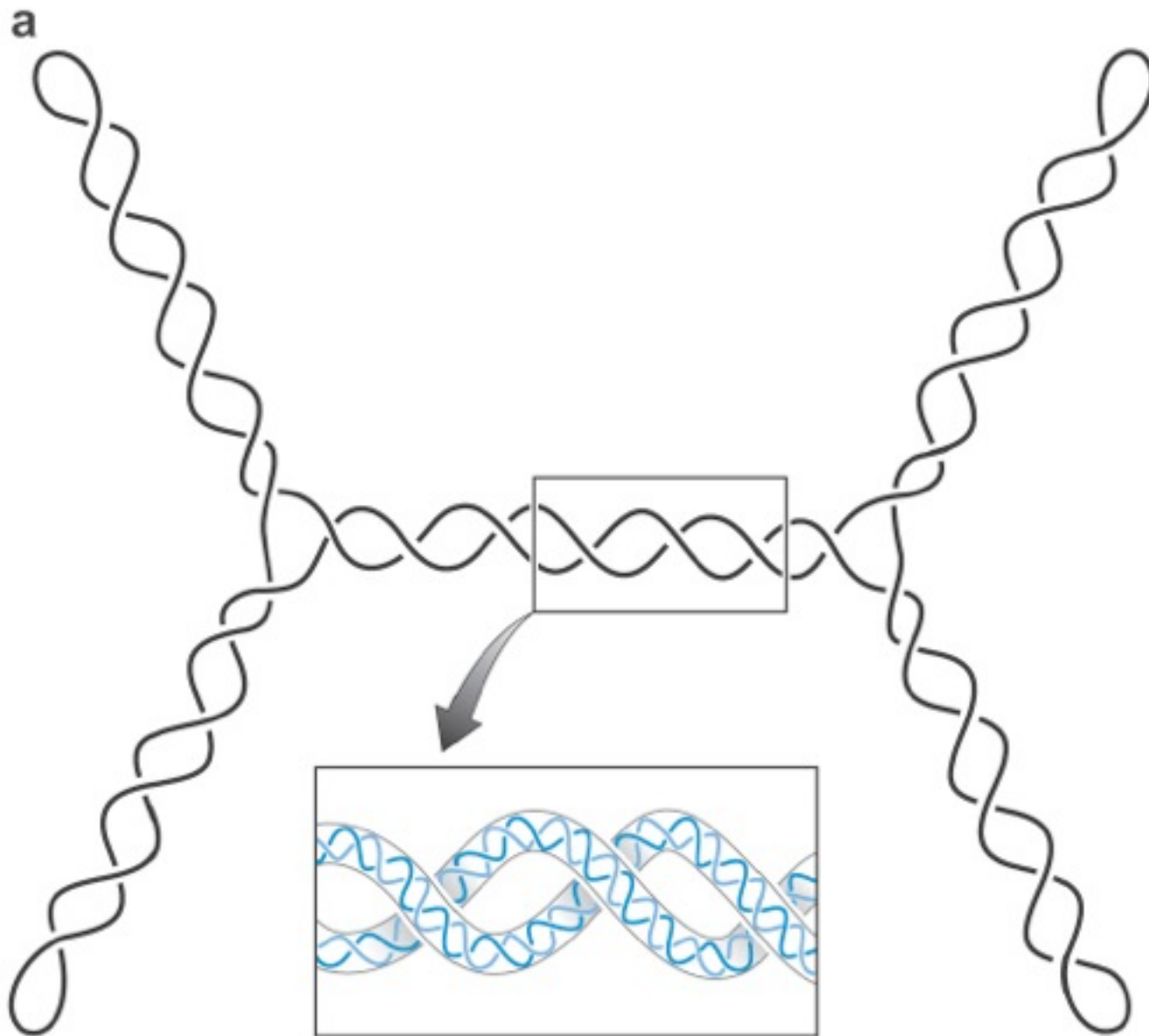
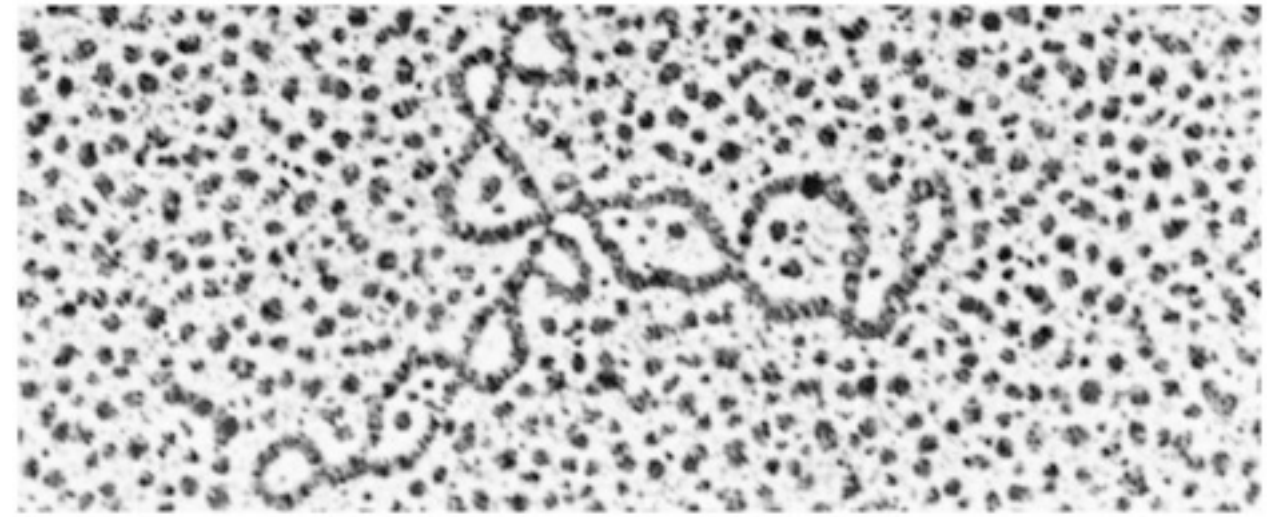
RNA structures are diverse!!

More in 3D

- “Hammerhead” ribozyme



Supercoiling



Supercoiling

